

In Vitro Antidiabetic, Free Radical Quenching Effect and Phytochemical Profiling of Shankpushpi With Special Reference To Chitrakoot Region

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ABSTRACT

Traditionally medicinal plants and herbs have always been a potential source to cure various diseases as well as in the fields of agriculture, veterinary etc. Several phyto-chemical and pharmacological investigation have clearly shown multidisciplinary usages of these plant derived medicine. Development of organic phyto-pharmaceutical agents is very interesting creation in field of medicinal phyto-chemistry and these phyto- agents are being used to treat various chronic diseases with the help of essential bioactive compounds. In the present research tenure represents the *in vitro* quenching of radicals (DPPH/ABTS) & enzyme (α glucosidase) by 80% hydro-methanolic extract of Shankpushpi(*Convolvulus pluricaulis*), also have been analyzed for their total polyphenolic and flavanoidic content. These study points out that Shankpushpi have potent antiradical and antidiabetic agents.

KEY WORDS :Shankpushpi, medicinal plant, phytochemical, antidiabetic activity.

INTRODUCTION

The rapidly increasing incidence of diabetes mellitus is becoming a serious threat to mankind health in all parts of the world (Malviya et al.,2010). Hyperglycemia is state of higher blood glucose level in body. It is cause by insulin not to activity of key anabolic hormone known as insulin, which is secreted in response to increased blood glucose level for control post prandial hyperglycemia (Bhowmik et al., 2009). The acquiescence of disturbed glucose and insulin metabolism respond in development of type 2 diabetes mellitus and cardiovascular disease, has given rise to the concept of the metabolic syndrome(Pandeya et al.,2013). Pancreatic/intestinal enzymes including α -amylase/ α -glucosidase play a key action in glucose metabolism catalyze hydrolyzing dietary starch and other polysaccharides which then breakdown to monosachharides like glucose and fructose prior to absorption in whole blood stream.

Glucose homeostasis is the solution for the treatment of Diabetes(Kaskoos RA., 2013). An effective antidiabetic compound should posses both antihyperglycemic and free radicals scavenging properties, with minimal or no side effects(Griffiths D W & Moseley G 1980).Acarbose-like drugs, competitive inhibitors of α -glucosidase present in the epithelium of small intestine, have been established to reduces postprandial digestion, and improve glycemic index without promoting insulin secretion in NIDDM patients. But this type of medication has serious side/adverse effects(Suvarchala Reddy., 2010 and Verma et al., 2012).

In diabetic patients oxidative stress caused by uncontrolled glucose metabolism. The potentially reactive derivatives of oxygen and nitrogen radicals continuously produced in cells, are detoxified by free radical scavengers and antioxidants(Tripathi et al., 2014). Hence mitigation of oxidative stress or α -glucosidase activity play a vital role to prevent diabetes mellitus and its complications. A compound with free radical quenching effect combined with inhibitory activities against α -glucosidase may be a more effective antidiabetic agents, with minimal or no side effect with minimal or no side effects (Griffiths D W & Moseley G., 1980).

Convolvulus Pluricaulis Choisy. also known as Shankpushpi belongs to family convolvulaceae, is a perennial herb that seems like morning glory. All parts of the herb are known to possess therapeutic benefits. The plant is used locally in Indian and Chinese medicine to cure various diseases. It is an important herb in Ayurveda for the therapy of chronic cough, sleeplessness, epilepsy, hallucinations, anxiety, liver disease, microbial disease, cytotoxic & viral diseases, central nervous system disease and its crude extract or isolated bioactive compound exhibited *in vitro* & *in vivo* pharmacological effect like CNS depression tranquilizing, antidepressant, antistress, neurodegenerative, anti-amnesic, antioxidant, anticatatonic activity etc (Agarwa et al., 2014).

Current research was designed to investigate phytochemical profile of Shankpushpi (leaves, stem, flower & root) and this study also undertaken to evaluate free radical quenching activity and α -glucosidase inhibitory effects of separated parts of shakpushpi.

MATERIAL AND METHOD

Sample collection of Plant Materials

Different parts including Leaves, Stem, Flower & Root of Shankpushpi (*Convolvulus Pluricaulis*)were collected in March 2017 from Chitrakoot region and identified. All plant parts were collected, washed with fresh water and dried under

shade at room temperature separately. Selected parts were powdered and stored separately in sterile and air tight container for further research.

Chemicals

Methanol, water, DMSO, tris HCl, Folin & Ciocalteu's Phenol Reagent (FCP), p-nitrophenyl- α -D-glucopyranoside, Sodium carbonate and ascorbic acid were obtained from SRL, India. While 2,2-Diphenyl-1-picrylhydrazyl (DPPH) were purchased from Alfa acer, Great Britain. Acarbose, α -glucosidase rat intestinal acetone powder procured from Sigma Chemicals, USA. All solvents were HPLC grade while Chemicals were AR grade and used without further purification.

Preparation of Plant extracts

100 mg powdered sample of plant parts were extracted with 10 ml HPLC grade 80% hydro-methanol through open air reflux process at 40°C for 6 hours till dried than make the volume again 10 ml with hydro-methanol and reflux, this process was repeated several times. The extracts were filtered through filter paper (Watman no.1) to remove free unextractable substances. The filtrates of plant extract were evaporated at room temperature at dryness, finally dissolve with 10 ml with DMSO and preserved at 4-5°C for further process.

Phytochemical profiling (Primary screening)

Qualitative estimation of phytochemicals in selected parts of Shankpushpi under study were carried out in different organic solvent extracts (double distilled H₂O, Methanol & Petroleum Ether), which were prepared by using standard method of Tripathi et al., 2017. The extracted samples were screened for carbohydrates, alkaloids, flavonoids, proteins resin, anthocyanin & betacyanin, saponin, steroid & phytosterol, tannin, galactotannin, starch, glycoside, phenol, phlobatannin, terpenoid using our previous reported methods (Tripathi et al., 2017).

Determination of Total Polyphenolic Content:

Total polyphenolic content of extracts of plant leaves was measured using Folin-Ciocalteu reagent method, adopted as it is described by Tripathi et al., 2014. The 25 μ l of plant extract diluted with 125 μ l water followed by addition of 150 μ l of Folin-Ciocalteu reagent (1N) & 25 μ l of Na₂CO₃ (20% w/v) incubated at 45°C for 60 min absorbance was measured spectrophotometrically at 765nm (Synergy H₄ multimode micro plate reader, biotek instrument, inc Winooski, VT, USA). quantification was performed with respect to the standard curve of Catechol. ($y = 0.018X + 0.045$, $R^2 = 0.998$) results were expressed as milligram of catechol equivalent per ml of extract (Tripathi et al., 2014).

Determination of Total Flavonoid contents:

Basic principle of aluminum chloride method: The principle of aluminum chloride colorimetric method is that aluminum chloride forms acid stable complexes with the C4 keto group and both C3 & C5 hydroxyl group of flavones and flavonols. In addition it also forms acid liable with ortho-dihydroxyl groups in the A- or B- ring of flavonoids. Standard quercetin solutions of various concentrations were used to build up the calibration curve. Total flavonoid in the plant extracts, in brief, 50 μ l of sample, followed by 50 μ l of AlCl₃.6H₂O in ethanol and 25 μ l Sodium acetate, solution added. The absorbance at 430nm was taken (BioTek^{synergyH4} multi-mode microplate reader, BioTek Instruments, Inc Winooski, VT, USA), after 2.5 h of incubation at 20°C. Total flavonoid contents were calculated with respect to the standard curve of the flavonoid quercetin dehydrate ($Y = 0.017x + 0.025$, $R^2 = 0.941$). Results were expressed as micrograms of quercetin dehydrate equivalents (QE) per ml of the extract (Shoib A. B. & Shahid A. M., 2015).

Determination of Total Flavonol contents:

Flavonol content in the sample (100 times diluted with methanol) was measured by mixing equal volume of plant extract with 2% AlCl₃.6H₂O in a 96 well plate. Absorbance was recorded at 420 nm spectrophotometrically (BioTek^{synergyH4} multi-mode microplate reader, BioTek Instruments, Inc Winooski, VT, USA). Flavonol contents in the extracts were determined with respect to the standard curve of the flavonoid quercetin ($Y = 0.017x + 0.025$, $R^2 = 0.941$). Results were expressed as micrograms of quercetin equivalents (QE) per ml of the extract (Miliauskas G. et al, 2004).

DPPH (2, 2-diphenyl-1-picrylhydrazyl) free radical scavenging assay

The assay for free radical DPPH was done by using 2, 2-diphenyl-1-picrylhydrazyl (DPPH) method adopted as it is described by Tripathi et al., 2013. In brief, a 96-well microplate, 25 μ l of various dilutions of methanolic extract 125 μ l of tris-HCl buffer (0.1M, pH 7.4) and 125 μ l of DPPH solution (0.004 in methanols) were added. The reaction mixture was shaken well. The DPPH decolorization was recorded at 517 nm on a Biotek Synergy H₄ hybrid multimode micro plate reader after 30 min incubation in dark. The percentage of DPPH scavenging by plant extracts obtained in terms of ascorbic acid equivalent concentration. Quantification was performed with respect to the standard curve of ascorbic acid ($Y = 0.840x + 10.97$, $R^2 = 0.968$). Results were expressed as milligrams of ascorbic acid equivalent per ml of the extract (Tripathi et al., 2013).

ABTS free radical scavenging assay

Method of ABTS was adopted from Pandeya et al., 2016. This assay is based on the ability of different substances to scavenge 2,2'-azinobis (ethylbenzthiazoline-6- sulfonic acid) or ABTS radical cation. In its radical form, ABTS has a characteristic absorbance at 734nm which disappears after its reduction by an antiradical compound. Reduction of blue-green

ABTS radical coloured solution by hydrogen donating antioxidant is measured by the suppression of its characteristic long wave (734nm) absorption spectrum. The ability to test samples to scavenge ABTS radical cation was compared to ascorbic acid standard. The ABTS radical cation was regenerated by mixing 7Mm stock solution with 2.45 Mm potassium persulphate and incubating for 18 hours in dark at room temperature until reaction was complete and absorbance of ABTS cation solution was 0.637 (± 0.02) by diluting water at room temperature then 20 μ l of test sample with different concentrations were mixed with 180 μ l of ABTS solution and absorbance was measured at 734 nm after 5 minutes. Quantification was performed with respect to the standard curve of ascorbic acid ($Y=0.755x+17.50$, $R^2=0.985$). The ABTS scavenging capacity of the test samples was compared with ascorbic acid and percentage inhibition calculated as:

$$\text{ABTS radical scavenging activity (\%)} = \frac{(A_0 - A_1)}{A_0} \times 100$$

Where A_0 = Absorbance of control, A_1 = Absorbance of sample. All the determinations were performed in triplicates ($n=3$). (Pandeya et al., 2016)

Determination of α -Glucosidase Inhibition activity

Method for determination of α -Glucosidase were adopted from Tripathi. et al., 2013. Rat intestinal acetone powder (Sigma chemicals, USA) was sonicated properly in normal saline (100:1 w/v) and after centrifugation at 3000 rpm \times 30 mins the supernatant was treated as crude intestinal α -glucosidase. 50 μ l microliters of test samples dissolved in DMSO (5mg/ml solution) were mixed and incubated with 50 μ l of enzyme in a 96-well microplate for 5 mins. Reaction mixture was further incubated for another 10 mins with 50 μ l substrate (5 mM, p-nitrophenyl- α -D-glucopyranoside) prepared in 100 mM phosphate buffer (pH~ 6.8) and release of nitrophenol was read at, 405 nm spectrophotometrically (Synergy H4 multimode micro plate reader, biotek instrument, inc. Winooski, VT, USA). All the samples were run in triplicate and acarbose was taken as standard reference compound. Several dilutions of primary solution (5mg/ml DMSO) were made and assayed accordingly to obtain concentration of the test sample required to inhibit 50 % activity (IC_{50}) of the enzyme. Quantification was performed with respect to the standard curve of acarbose ($Y = 0.638x+11.94$, $R^2 = 0.974$) results were expressed as milligram of Acarbose equivalent per ml of extract. Percent α -glucosidase inhibition was calculated as using following equation:

$$\text{Percent } \alpha\text{-Glucosidase inhibition (\%)} = A_0 - A_1 / A_0 \times 100$$

Where A_0 was the absorbance of the control and A_1 was the absorbance in the presence of the sample. (Tripathi. et al., 2013)

Statistical Analysis

Data reproduced during the experimental work, were analyzed using Origin Pro8.5 software. All parameters were triplicately recorded for accurate result. Tables show mean value and standard deviation (\pm) of reproduced data. Graphs were also plotted using OriginPro 8.5 software.

RESULT

Herbal medicines are being used since prehistoric times which are widely accepted in advance therapeutics and their utilization stimulated the trend of using of natural products in treatment of various diseases. For present study different parts of Shankpushpi were collected from the local area of Chitrakoot (University campus, Rajulla farms & Deendayal Research Institute) according to their therapeutic value and seasonal availability. It was make sure that plants were healthy and uninfected.

Chitrakoot is a holy place for Hindu pilgrimage, which is situated in the northern region of Satna district of Madhya Pradesh. It has a very rich wealth of medicinal plants, which has also been described in our epics like Ramayana.

वसूकसारं नलीनीमतीत्यैवोत्तरान् कुरुन् ।

पर्वतश्चित्रकूटोऽसौ बहुमूलफलोदकः ।।26।।

(26) Valmiki Ramayana

निशि भान्त्यचलेन्द्रस्य हुताशनशिखा इव ।

औषध्यः स्वप्रभालक्ष्या भ्राजमानाः सहस्रशः ।।21।।

(21) Valmiki Ramayana

According to Valmiki Ramayana, Chitrakoot is a beautiful and sacred place where different types of herbs, shrubs, trees and climbers bearing variety of fruits, flowers and roots are available. Great saint Valmiki and Goswami Tulsidas illustrated a comprehensive account of biodiversity in their texts Ramayana and Ramcharitmanas respectively (Tripathi et al., 2015).

According to Maharshi Valmiki, Kamadgiri hill is rich in diverse plants. These are Aam (*Mangifera indica*), Jamun (*Syzygium cumini*), Asna (*Lagerstroemia indica*), Lodh (*Symplocos racemosa*), Chironji (*Buchanania lanzan*), Kathl (*Artocarpus heterophyllus*), Dhawa (*Anogeissus lantifolia*), Dhak (*Butea monosperma*), Ankol (*Allangium salvifolium*), Bhavya (*Dillenia indica*), Tinsa (*Ougenia oogeinensis*), Bel (*Aegle marmelos*), Tendu (*Diospyros melanoxylon*), Bans (*Dendrocalamus strictus*), Kasmri (*Gmelina arborea*), Neem (*Azadirachta indica*), Sakhua (*Shorea robusta*), Barun (*Crateva unilocularis*), Mahua (*Madhuka longifolia*), Tilak (*Wendlandia exerta*), Ber (*Zizyphus mauritiana*), Anola (*Phyllanthus emblica*), Kadamb (*Anthocephalus chinensis*), Bent (*Calamus rotung*), Indrajau (*Holarrhena pubescens*), Bijak (*Punika granatum*) and Neebu (*Citrus aurantifolia*) with other flowering fruiting and shade giving trees (Sikarwar, 2011).

In addition, Chitrakoot has a rich wealth of medicinal plants too i.e. *Tinospora cordifolia*, *Gymnema sylvstre*, *Achyranthes aspera*, *Urginea indica*, *Curculigo orchioides*, *Dioscorea bulbifera*, *Desmodium gangeticum*, *Coccinia grandis*, *Cordia macleodii*, *Litsea glutinosa*, *Oroxylum indicum*, *Steriospermum suaveolens*, *Pterocarpus marsupium*, *Terminalia arjuna*, *T. Bellirica*, *T. Chebula*, *Actinopteris radiata*, *Cyperus rotundus*, *Vernonia cinerea*, *Sida cordifolia*, *Ampelocissus latifolia*, *Peristrophe paniculata*, *Cassia tora*, *Diplocyclos palmatus*, *Tridax procumbens*, *Phyllanthus fraternus*, *Elytraria acaulis*,

Solanum nigrum, *Termnus labialis*, *Vitex negundo*, *Abutilon indicum*, *Cocculus hirsutus*, *Hemidesmus indicus*, *Enicostemma hyssopifolium*, *Boerhavia diffusa*, *Solanum virginianum*, *Helicteres isora*, *Aegle marmelos*, *Allangium salvifolium* etc (Sikarwar, 2011).

Qualitative estimation phytochemicals of extracts of leaves, Stem, Flower & Root of *Convolvulus pluricaulis* was reported in Table.1. For phytochemical screening, water, methanol and petroleum ether extracts were used. Preliminary phytochemical tests indicate the presence or absence of various phytoconstituents in a given plant sample. The presence of carbohydrate Protein Phenol alkaloids, tannins, flavonoids, steroids and reducing sugars was observed in all the selected plants parts extracts except phlobatannins, & terpenoids all these bioactive compounds were found to be present in almost all the studied samples.

Table 1: Phytochemical Screening of Shankpushpi

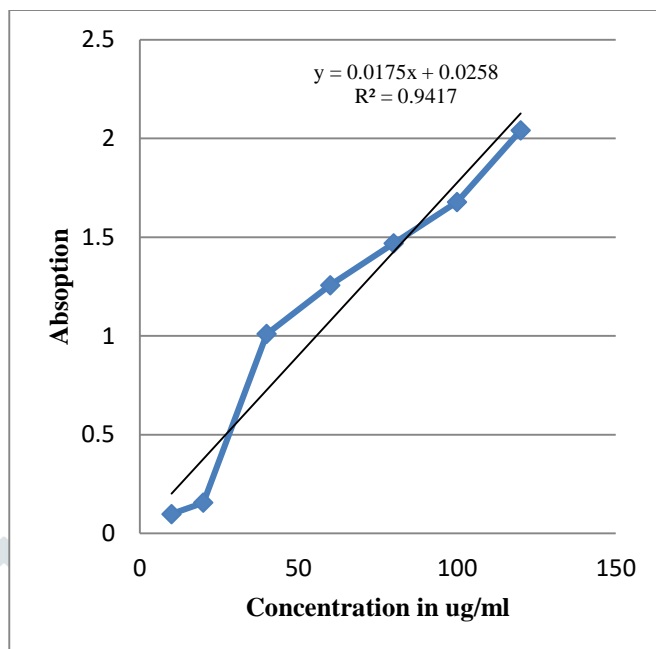
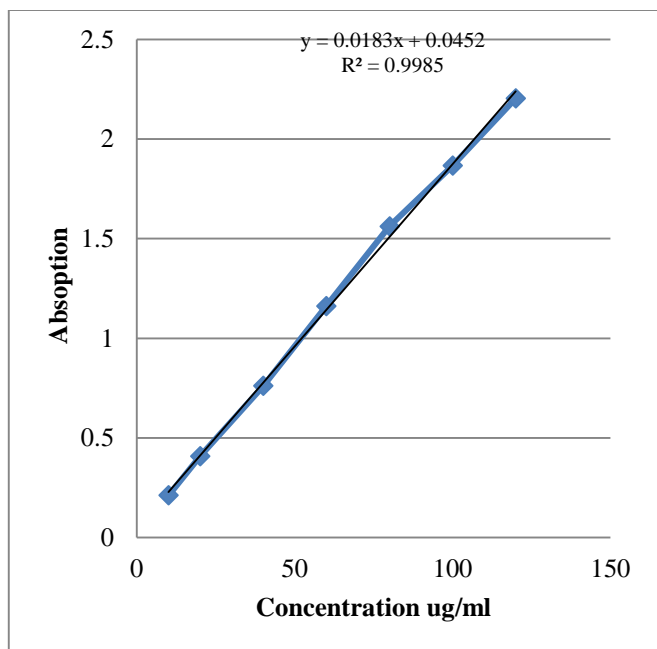
S. No.	Phytochemicals	leaves			Stem			flower			root		
		W	M	PE	W	M	PE	W	M	PE	W	M	PE
1.	Carbohydrate	+	+	-	+	-	-	+	+	-	-	-	-
2.	Alkaloid	+	+	+	+	+	+	+	+	+	+	+	+
3.	Flavonoid	+	+	+	+	+	-	+	-	-	-	-	-
4.	Protein	-	-	+	+	+	-	-	-	-	-	-	-
5.	Resin	+	+	+	+	+	-	+	+	+	+	-	+
6.	Anthocyanin	+	-	-	+	-	-	-	-	-	-	-	-
7.	Saponin	+	-	-	+	-	-	+	-	-	+	-	-
8.	Steroid	-	-	-	-	-	-	+	+	+	-	-	+
9.	Tannin	+	+	+	+	+	+	+	+	+	+	+	+
10.	Starch	+	-	-	-	-	-	-	-	-	-	-	-
11.	Glycoside	-	+	+	-	-	+	+	+	+	-	-	-
12.	Phenol	+	+	+	+	+	+	+	+	+	-	+	+
13.	Phlobatanin	-	-	-	-	-	-	-	-	-	-	-	-
14.	Terpenoid	+	-	-	-	-	-	+	-	-	-	-	-

W = water extract, PE = Petroleum ether extract, M = Methanolic extract

Presence of TPC in plant extract was based on the absorbance of selected plant parts and Folin-Ciocalteu reagent mixture at 725 nm. Since Folin-Ciocalteu assay measures all the phenolics, the choice of catechol as standard is based on the availability of a stable and pure chemical.

Table 2: Represent Total Polyphenolic content (TPC)

S. No.	Name of Plant's Part	TPC equivalent to Catechol $\mu\text{g/ml} \pm \text{SD}$	TFC equivalent to Quercetine $\mu\text{g/ml} \pm \text{SD}$	TFLC equivalent to Quercetine $\mu\text{g/ml} \pm \text{SD}$
1	Leaf	7.28 \pm 0.0002	4.35 \pm 0.001	1.17 \pm 0.002
2	Stem	28.72 \pm 0.001	7.5 \pm 0.004	0.64 \pm 0.001
3	Flower	38.22\pm0.0004	31.470\pm0.0002	6.1\pm0.001
4	Root	5.72 \pm 0.002	3.7 \pm 0.004	0.412 \pm 0.003



Graph 1 : Standard Curve Of Catechol For Estimation Of TPC

Graph 2 : Standard Curve Of Quercetin For Estimation Of TFC & TFLC

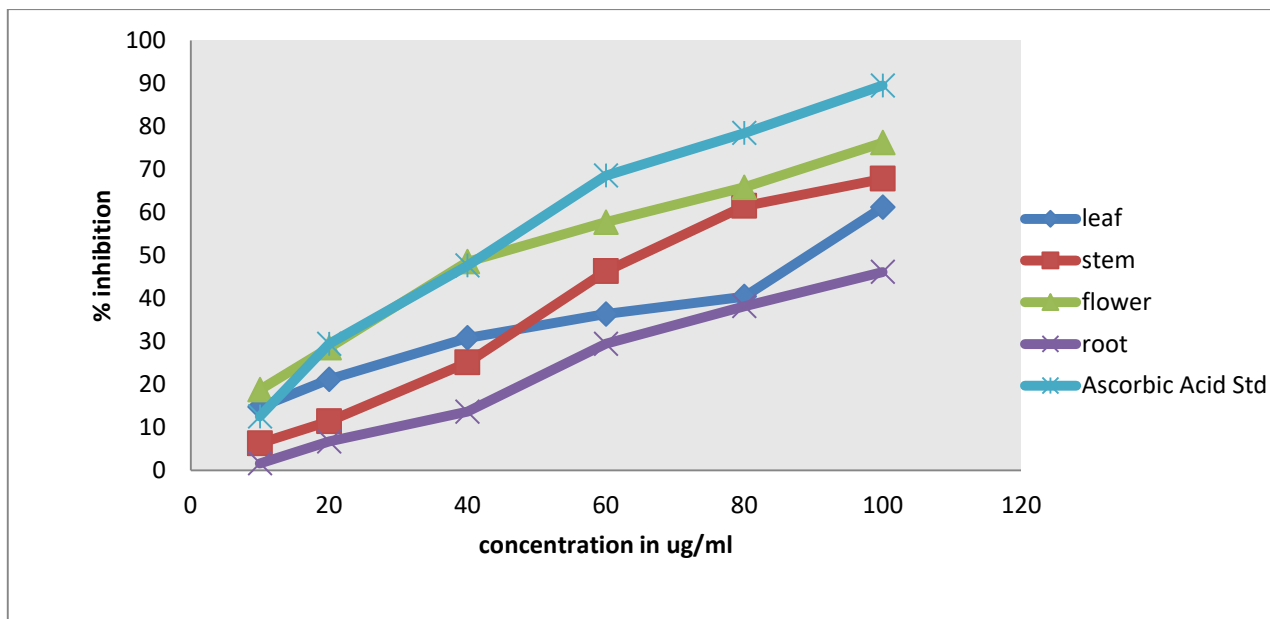
Graph-1 shows the standard curve of catechol, which is used for the quantification of total polyphenolic contents equivalent to catechol $\mu\text{g/ml}$ in methanolic extracts of selected plant samples. Table-1 presents the concentration of total polyphenolic content in catechol equivalent $\mu\text{g/ml}$. In the present investigation concentration of TPC of plants were ranged from $5.72 \pm 0.002 \mu\text{g/ml}$ to $38.22 \pm 0.0004 \mu\text{g/ml}$. Highest concentration of polyphenols was shown by Flower's hydro-methanolic extract ($38.22 \pm 0.0004 \mu\text{g/ml}$), while in stem, it was found to be $28.72 \pm 0.001 \mu\text{g/ml}$. Presence of polyphenolic contents in all the sample was observed in our preliminary phytochemical screening studies. All the selected plant parts showed positive results for the test for phenols. The amount of total flavonoid & total flavonol compounds that are present in same table 1. A standard curve was plotted to quantify Flavonoid/flavonolic compound was reported equivalent to Quercetin $\mu\text{g/ml}$ in methanolic extract of selected plant samples, as the concentration of the extract increases, the mean absorbance value also increases. The amount of TFC & TFLC exhibited by hydro-methanolic flower's extract ($31.470 \pm 0.0002 \mu\text{g/ml}$ & $6.1 \pm 0.001 \mu\text{g/ml}$) was higher compared with other selected plant parts extracts Graph 2 represent the standard curve of quercetin.

DPPH & ABTS scavenging activity (% inhibition) are shown in respectively table 3 & 4 and statistical results of the data are shown in graph 3 & 5. Graph 4 shows the standard curve of ascorbic acid. This test is based on the ability of the stable 1, 1'- diphenyl 2- picryl hydrazyl (DPPH) and ABTS free radical to decolorize in the presence of free radicals. In our investigation DPPH & ABTS radical scavenging assay protocol, ascorbic acid was used as reference standard, 80% methanol was used a blank, while DPPH & ABTS solution without sample served as a control. Percentage inhibition was calculated and a curve between % inhibition and concentrations were plotted. Linear regression analysis was carried out for calculating the inhibition concentration of sample required to scavenge DPPH & ABTS radicals by 50% (IC_{50} value). IC_{50} value of standard ascorbic acid is calculated to be $46.464 \mu\text{g/ml}$. Table -6 shows the IC_{50} values of all the samples. When compared to standard ascorbic acid with selected plant parts exhibited good DPPH & ABTS scavenging activity, in terms of IC_{50} values. Highest DPPH & ABTS radical scavenging activity was shown by flower's hydro-methanolic extracts ($\text{IC}_{50} = 52.850 \mu\text{g/ml}$ & $47.6 \mu\text{g/ml}$). The total phenolic content for 50, 100 and 200 $\mu\text{g/ml}$ extracts was 0.2092, 0.2380 and 0.3608 mg GAE/ gram. Free radicals quenching activity of methanolic extract of *Convolvulus pluricaulis* and the standard was found to be highest at 100 $\mu\text{g/ml}$ which was 52.56% and 93.48% respectively, reported by Balaji et al., 2014.

Table 3: % inhibition of DPPH scavenging activity of 80%hydro- methanolic extracts of Shankpushpi; Leaves, Stem, Flower, Root and ascorbic acid

S. No.	Conc. in	(% of free radical (DPPH) Scavenging)									
		Ascorbic	Error \pm SD	leaf	Error \pm	stem	Error \pm	flower	Error \pm	root	Error \pm SD

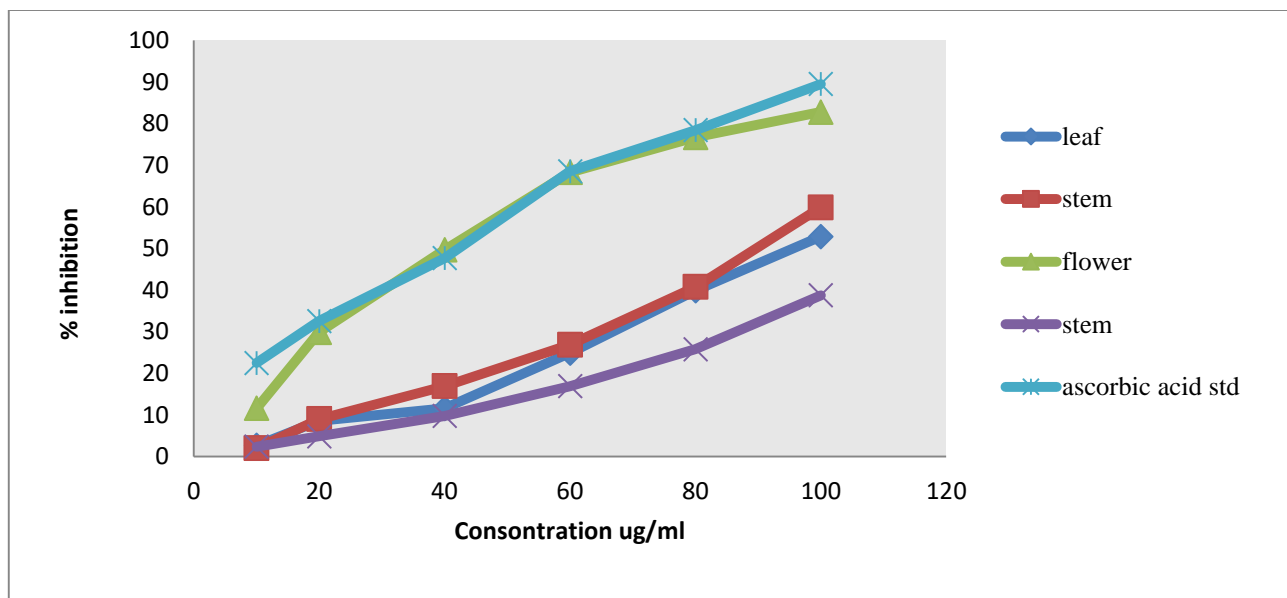
	µg/ml	Acid			SD		SD		SD		
1	10	12.56	0.0004	14.74	0.006	6.32	0.0002	18.76	0.000	1.65	0.0007
2	20	29.45	0.003	21.2	0.000	11.43	0.0004	28.56	0.0002	6.77	0.0003
3	40	47.68	0.0003	30.82	0.0000	25.112	0.005	48.57	0.0001	13.69	0.0002
4	60	68.59	0.002	36.39	0.0000	46.32	0.0006	57.756	0.0006	29.473	0.0002
5	80	78.47	0.002	40.45	0.0005	61.503	0.0026	65.867	0.0004	38.195	0.001
6	100	89.54	0.001	61.2	0.0004	67.82	0.0006	76.25	0.0004	46.165	0.000



Graph 3: Representing graph between % of free rdical (dpph) scavenging and concentration.

Table 4: %inhibition of ABTS scavenging activity of 80% hydro- methanolic extracts of Shankpushpi; Leaves, Stem, Flower, Root and ascorbic acid

S.No.	Conc. in µg/ml	(% of free radical (ABTS) Scavenging)									
		Ascorbic Acid	Error± SD	leaf	Error± SD	stem	Error± SD	flower	Error± SD	root	Error±SD
1	10	12.56	0.0003	2.67	0.001	1.98	0.002	11.58	0.000	2.35	0.0002
2	20	29.45	0.000	8.56	0.0002	8.97	0.0012	29.87	0.0002	4.87	0.0004
3	40	47.68	0.000	11.57	0.0004	16.9	0.0004	49.67	0.0002	9.76	0.0004
4	60	68.59	0.0004	24.87	0.0004	26.87	0.0004	68.29	0.0007	16.9	0.0006
5	80	78.47	0.0002	39.86	0.0005	40.76	0.004	76.73	0.0006	25.79	0.0005
6	100	89.54	0.0004	52.87	0.000	59.87	0.0002	82.8	0.0002	38.76	0.0002



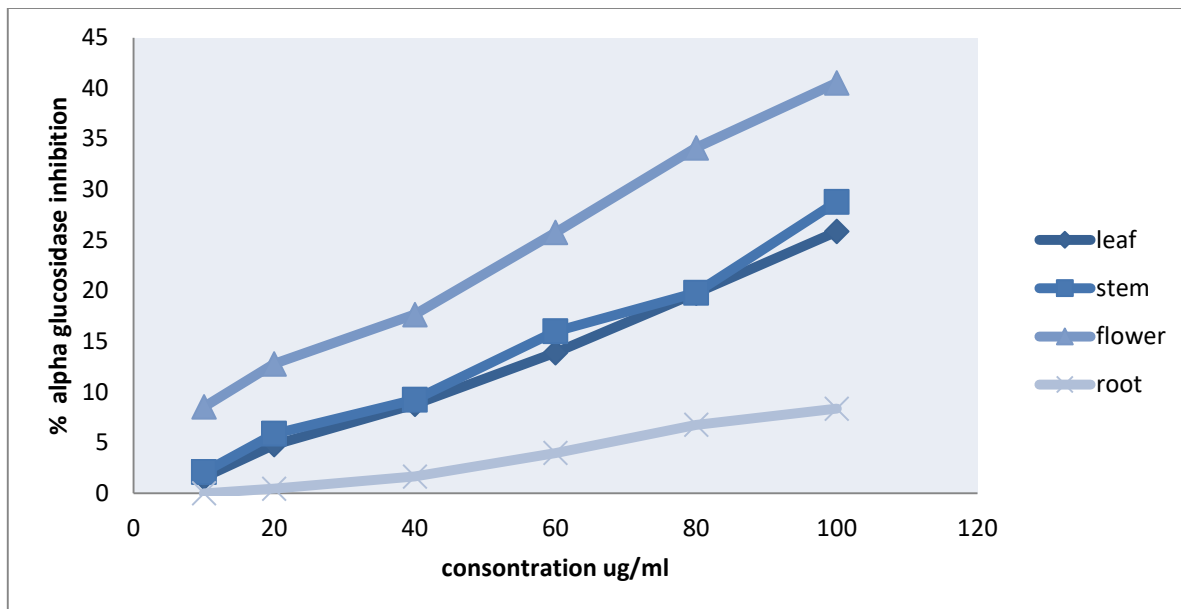
Graph 4: Representing graph between % of ABTS scavenging and concentration.

Alpha glucosidase inhibition activity are shown in table 5. Graph 6 presents the statistical analysis of data. Graph 5 shows the standard curve of acarbose. Glycosidases widely spread in microorganisms, plants and animals, are a very important class of enzymes. The hydrolysis of glucosidic linkage catalyze by these types of enzyme, thereby degrading oligosaccharides and glycoconjugates. Among these glycosidases, alpha glucosidase is able to catalyze the cleavage of glycosidic bonds involving terminal glucose connected at the site of cleavage through a linkage at the anomeric center. Glycosidases are involved in a number of principle biological processes such as digestion, biosynthesis of glycoprotein, lysosomal catabolism of glycoproteins and lysosomal catabolism of glucocenjugates, which are related to many metabolic disorders such as diabetes, obesity, glycosphingolipid lysosomal storage disease, HIV and tumors (Sancheti S et al., 2014). Several alpha glucosidase and alpha amylase inhibitors have been isolated from plants to act as replacement drugs with increased potency and lesser deleterious effects, than existing synthetic drugs (Kazeem et al 2013). Delay of carbohydrate absorption with a plant based alpha glucosidase inhibitor offers a prospective therapeutic approach for the management of type II diabetes mellitus (Reddy N.V.L.S, 2010).

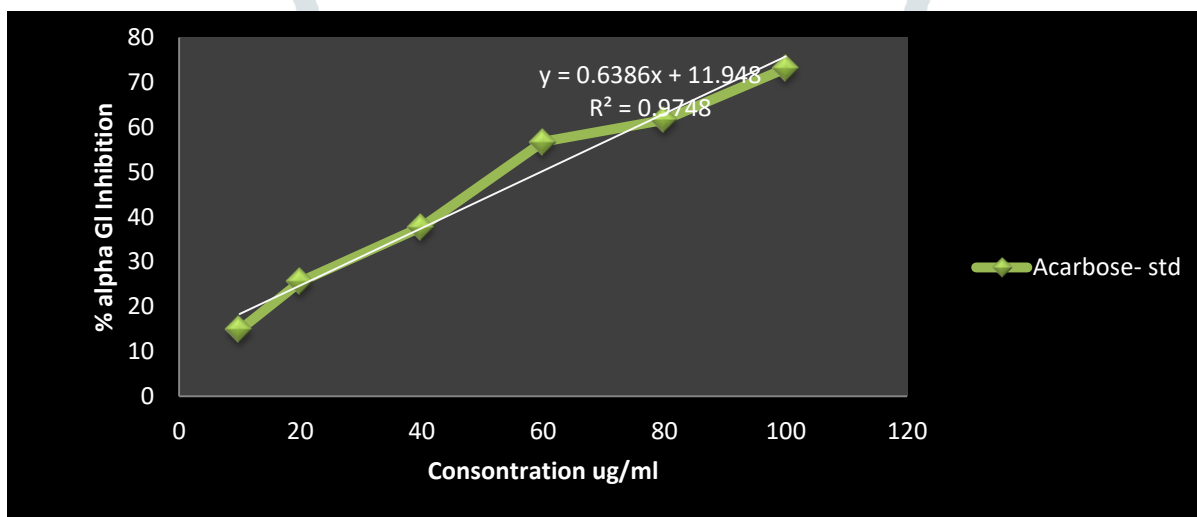
In our experiments, reference standard acarbose showed an IC₅₀ value of 59.65 µg/ml (table 6). Compared to acarbose, some of the extracts exhibited very effective inhibition, and highest was observed in flower's hydromethanolic extract (IC₅₀ value is 127.55 µg/ml).

Table 5: % inhibition of Alpha GI scavenging activity of 80% hydro- methanolic extracts of Shankpushpi; Leaves, Stem, Flower, Root and Acarbose

S. No.	Conc. in µg/ml	(% of alpha Glucosidase enzyme Scavenging)									
		Acarbose	Error±SD	leaf	Error±SD	stem	Error±SD	flower	Error±SD	root	Error±SD
1	10	14.834	0.004	1.57	0.000	2.077	0.000	8.533	0.0004	0	0.0004
2	20	25.56	0.0003	4.8	0.000	5.87	0.001	12.78	0.000	0.45	0.000
3	40	37.764	0.0024	8.78	0.0006	9.245	0.000	17.654	0.000	1.65	0.0006
4	60	56.74	0.0004	13.895	0.0004	15.987	0.000	25.765	0.0007	3.98	0.0004
5	80	61.65	0.000	19.765	0.000	19.8	0.002	34.12	0.000	6.76	0.000
6	100	73.1	0.0000	25.86	0.000	28.78	0.0004	40.54	0.000	8.354	0.0007



Graph 5: Representing graph between % of alpha Glucosidase enzyme inhibition and concentration of selected samples



Graph 6: Representing graph between % of alpha Glucosidase enzyme inhibition and concentration of acarbose as standard.

Table 6: IC₅₀ values of samples in µg/ml equivalent to ascorbic acid in DPPH

S.N.	Selected plant's part	IC ₅₀ Value (µg/ml ± SD)		
		DPPH	ABTS	Alpha GI
1	Leaf	86.62±0.0003	99.75±0.0004	193.91±0.0001
2	Stem	70.151±0.002	91.31±0.002	181.62±0.004
3	Flower	52.850±0.001	46.67±0.001	126.71±0.0024
4	Root	105.095±0.0004	137.6±0.001	531±0.0004

CONCLUSION

It can be concluded that hydro-methanolic solvent is a more efficient solvent in extracting phyto-constituents. TPC, TFC & TFLC of the hydro methanolic extract of *Convolvulus pluricaulis*, point out that it may account for its free radical scavenging (DPPH & ABTS) and also antidiabetic characteristics. Best antidiabetic and free radical scavenger agents present in flower of shankpushpi, because of presence of highest amount of phytochemicals in comparison of other selected parts.

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