

DOMINATOR CHROMATIC NUMBER OF VARIOUS CENTRAL NEIGHBOURLY IRREGULAR CHEMICAL GRAPH among p- BLOCK ELEMENTS

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ABSTRACT

In this paper, we find the dominator chromatic number for central Neighbourly Irregular Chemical cycle graph and central Neighbourly Irregular Chemical path graph. We could derive some Dominator Chromatic of Central Neighbourly Irregular Chemical Graphs with the molecular structure which is derived only among the p-block Elements in the area of Inorganic Chemistry. Also these parameters are compared with the dominator chromatic number of their respective graph families.

KEYWORDS

Regular graph, Irregular graph, NIC Graph, Chromatic number of Central NIC graph, Proper coloring and Dominator coloring.

INTRODUCTION

We concerned with finite, undirected, connected graph G with vertex set $V(G) = \{v_1, v_2, v_3, \dots, v_n\}$ and edge set $E(G) = \{e_1, e_2, e_3, \dots, e_m\}$. If v_i and v_j are vertices of G , then the edge connecting them will be denoted by $v_i v_j$. A graph is said to be regular if all its vertices have same degree. A connected graph G is said to be highly irregular if each neighbor has different degree. The graph G is said to be Neighbourly irregular graph abbreviated as NI graph, if no two adjacent vertices of G have the same degree. This concept was introduced by Gnaana Bhargasaam and Ayyasamy who constructed NI graph.

In this paper, A path graph is a graph whose vertices can be listed in the order $v_1, v_2, v_3, \dots, v_n$ such that the edges are $\{v_i v_j\}$ where $i = 1, 2, 3, \dots, n-1$. A cycle graph C_n is a graph n vertices containing a single cycle through all the vertices

Chemical Term	Mathematical(Graph -theoretical) Term
Atom	Vertex
Molecule	Molecular graph
Covalent bond	Edge
Valency of a atom	Degree of the vertex

Definition 1.1

Let G be a simple and Undirected graph and let its vertex and edge set be denoted by $V(G)$ and $E(G)$.

Definition 1.2

In graph theory, a regular graph is a graph where every vertex has same number of degrees of valancy. A regular directed graph must be satisfy the stronger condition of that the indegree and outdegree of each vertex are equal to each other.

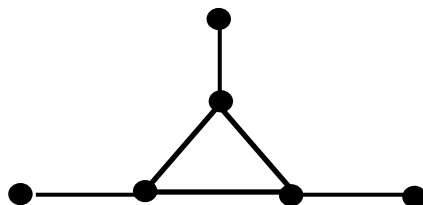
Example:



Definition 1.3

A graph G is said to be Irregular if for each vertex V of G the Neighbours of V having distinct degree.

Example:



Definition 1.4

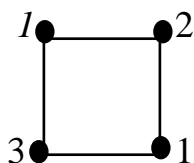
The Central graph of G denoted by $C(G)$ is obtained by subdividing each edge of G exactly once and joining all the non-adjacent vertices of G in $C(G)$.

Definition 1.5

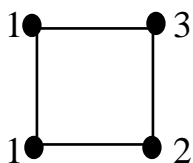
A proper (vertex) coloring of a simple graph $G(v)$ is defined as a vertex coloring from a set of colors such that no two adjacent vertices share a common color.

$$F: V(G) \rightarrow \{1,2,3,\dots,k\}$$

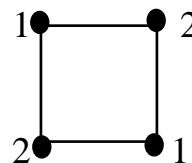
Examples:



a proper coloring
with 3 colors



not a proper coloring



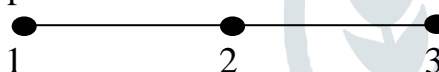
a proper coloring
with 2 colors

Definition 1.6

The chromatic number $\chi(G)$ of a graph is the smallest number of colors needed to color the vertices so that no two adjacent vertices sharing the same color [skiena 1990, P210].i.e) The smallest value of possible to obtain a k-coloring and is often denoted $\chi(G)$.

A subset of vertices assigned to the same color is called a color class, every such class forms an independent set. The color class corresponding to color I is denoted by V_i .

Examples:



P_3 is 3 colourable



P_3 is 2 colourable

Therefore, $\chi(G) = 2$

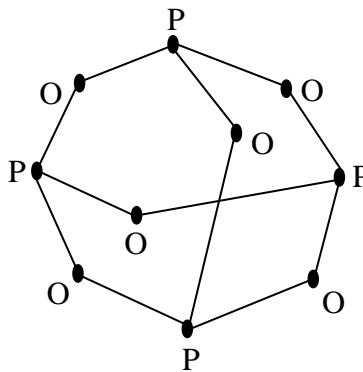
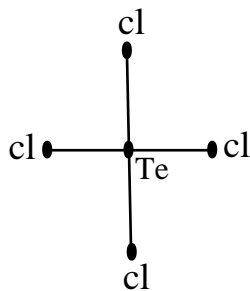
SOME BASIC FACTS:

- * If G has n vertices, then $\chi(G) \leq n$.
- * $\chi(G) = 1 \Leftrightarrow G$ has no edges.
- * $\chi(C_{2n}) = 2$ and $\chi(C_{2n+1}) = 3$.
- * $\chi(K_n) = n$.
- * If H is a subgraph of G then $\chi(G) \geq \chi(H)$.

Definition 1.7

A graph is said to be a Neighbourly Irregular Chemical Graph(NIC Graph) if molecular structure of corresponding element of the atoms has different valency bond in its adjacent atoms.

Examples:



TeCl₄ (Tellurium Chloride)

P₄O₆ (Tetra phosphorus Hexaoxide)

2. DOMINATOR CHROMATIC NUMBER OF CENTRAL NIC GRAPH

Dominator chromatic of central NIC graph of various classes is obtained in this section.

Theorem 2.1

For cycle graph c_n of order $n \geq 3$,

$$\chi_d [C(c_n)] = \begin{cases} [2n/3]+2 & \text{when } n=3 \\ [2n/3]+1 & \text{otherwise.} \end{cases}$$

Proof:

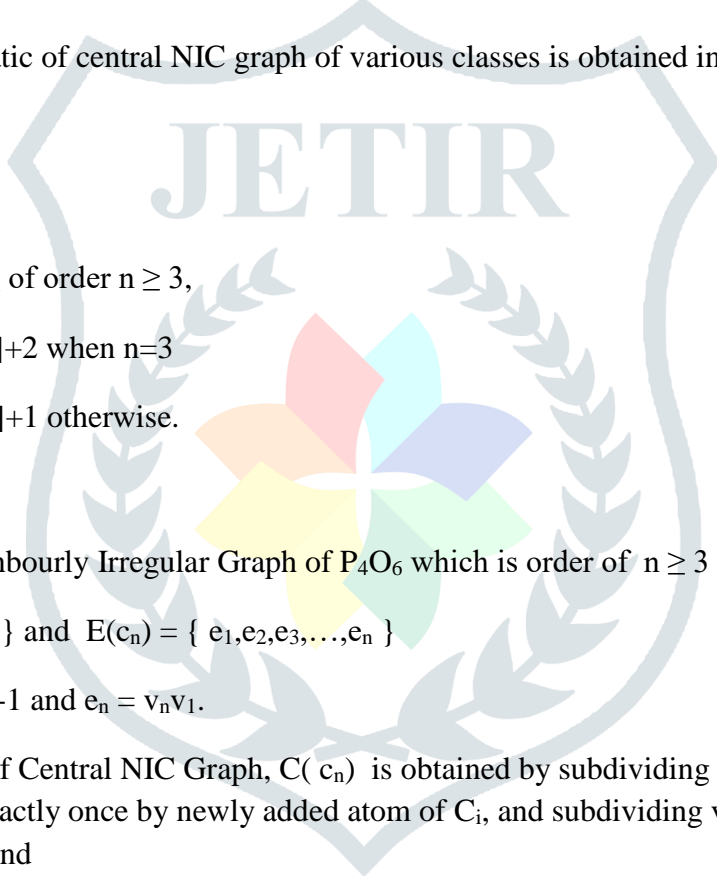
Let c_n be the Neighbourly Irregular Graph of P₄O₆ which is order of $n \geq 3$ and

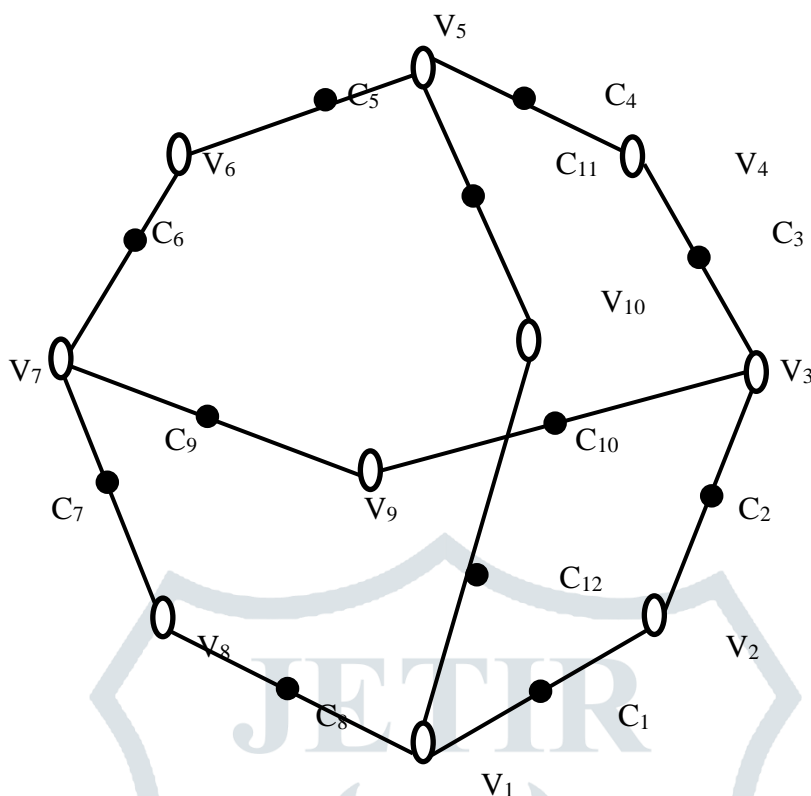
Let $V(c_n) = \{ v_1, v_2, \dots, v_n \}$ and $E(c_n) = \{ e_1, e_2, e_3, \dots, e_n \}$

where $e_i = v_i v_{i+1}$, $1 \leq i \leq n-1$ and $e_n = v_n v_1$.

By the definition of Central NIC Graph, $C(c_n)$ is obtained by subdividing each covalent bond of $v_i v_{i+1}$, $1 \leq i \leq n-1$ of C_n exactly once by newly added atom of C_i , and subdividing $v_n v_1$ by C_n and joining v_i with v_j , $1 \leq i, j \leq n$, $i \neq j$ and

$v_i v_j \neq E(C_n)$.





Central graph of NIC on P₄O₆(Tetra Phosphorus Hexaoxide)

Let $V_1 = \{ v_1, v_2, v_3, \dots, v_n \}$ and $V_2 = \{ c_1, c_2, c_3, \dots, c_n \}$ then $V(C(c_n)) = V_1 \cup V_2$.

This above procedure gives a dominator coloring of $C(c_n)$. Therefore all the newly added atoms $c_i, 1 \leq i \leq n$ in $C(c_n)$ forms an independent set, color these atom of the molecular graph by color 1.

When $n = 3k, k \geq 2$, the atoms v_i and $v_{i+1}, i = 1, 4, 7, \dots, (n-2)$ are colored by color $2\lceil i/3 \rceil$ and the remaining atoms $v_j, j = 3, 6, 9, \dots, n$ are colored by $1 + (2j/3)$ when $n = 3k-1, k \geq 2$, the atoms v_i and $v_{i+1}, i = 1, 4, 7, \dots, (n-4)$ are colored by color $2\lceil i/3 \rceil$ and the remaining atoms $v_j, j = 3, 6, 9, \dots, (n-2)$ are colored by color $i + (2j/3)$ and v_{n-1} and v_n are colored by $\lceil 2n/3 \rceil$ and $\lceil 2n/3 \rceil + 1$.

When $n = 3k+1, k \geq 2$, the atoms of the molecular v_i and $v_{i+1}, i = 1, 4, 7, \dots, (n-3)$ are colored by color $2\lceil i/3 \rceil$ and the remaining atoms $v_j, j = 3, 6, 9, \dots, n-1$ are colored by $1 + (2j/3)$ for and v_n is colored by $\lceil 2n/3 \rceil + 1$. When $n = 3$, the atoms v_i is colored by color $i+1, 1 \leq i \leq 3$. When $n = 4$, the atoms of v_i are colored by the color sequence $(2, 2, 3, 4)$

The vertices of the atom v_i and $v_{i+1}, i = 1, 4, 7, \dots$, dominates one of the color classes of $v_j,$

$j = 3, 6, 9, \dots$ the atoms $v_j, j = 3, 6, 9, \dots$ dominates itself. The centre atoms $c_i, i = 1, 4, 7, \dots$ dominates the color class of v_i . The centre atoms c_i and $c_{i+1}, i = 2, 5, 8, \dots$ dominates the color class of v_{i+1} , as they are adjacent to v_{i+1} .

When $n = 3$, it is easy to see that $\chi_d [C(C_3)] = \lceil 2n/3 \rceil + 2$ and when $n=4$, it is seen that $\chi_d [C(c_4)] = \lceil 2n/3 \rceil + 1$.

Hence,

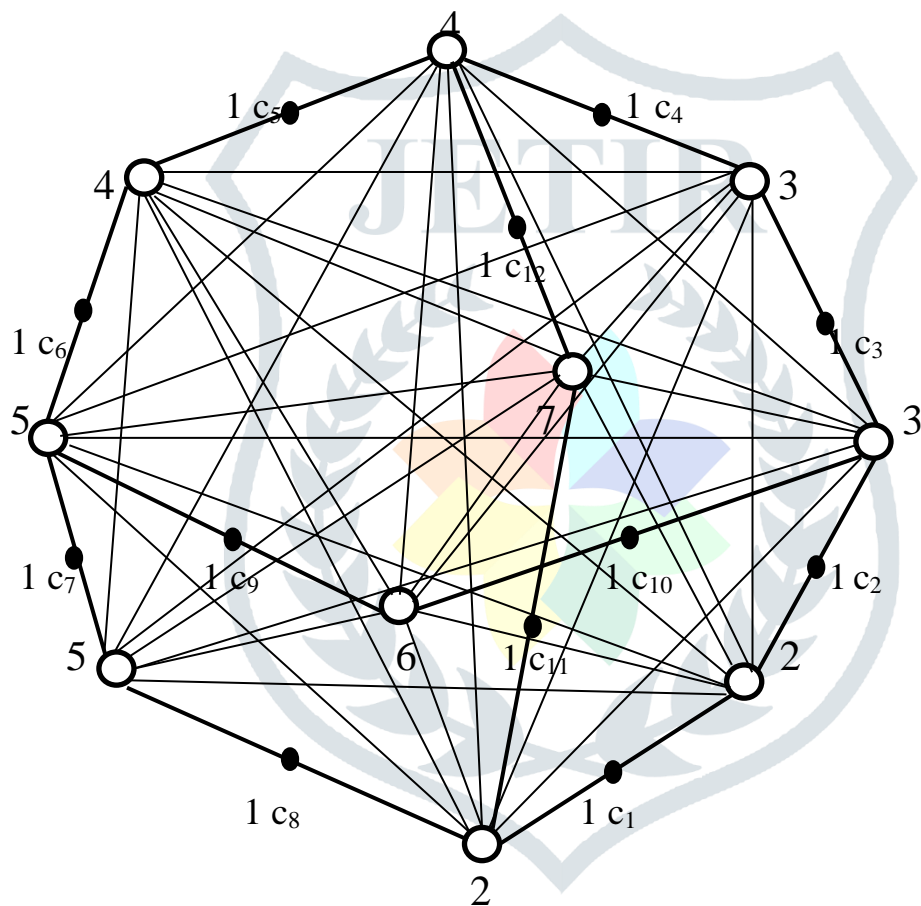


$$\chi_d[C(C_n)] = \begin{cases} [2n/3]+2 & \text{when } n=3 \\ [2n/3]+1 & \text{otherwise} \end{cases}$$

The following illustration of the example of Central Neighbourly Irregular Chemical Graph would be discussed in the above result.

Example: 2.1

In this figure Central of Neighbourly Irregular Chemical Graph of P₄O₆(Tetra Phosphorus Hexaoxide) C₁₀ is depicted with a dominator colouring.



Chromatic Number of NIC on P₄O₆(Tetra Phosphorus Hexaoxide)

The color classes of c₇ are

$$V_1 = \{ c_1, c_2, c_3, c_4, c_5, c_6, c_7, c_8, c_9, c_{10}, c_{11}, c_{12} \},$$

$$V_2 = \{ v_1, v_2 \}, V_3 = \{ v_3, v_4 \}, V_4 = \{ v_5, v_6 \},$$

$$V_5 = \{ v_7, v_8 \}, V_6 = \{ v_9, v_{10} \}, V_7 = \{ v_{10} \}.$$

Therefore, the dominator chromatic number is $\chi_d[C(C_{10})] = 7$.

Theorem: 2.2

For Neighbourly Irregular Chemical Path Graph P_n of order $n \geq 2$,

$$\chi_d[C(P_n)] = \begin{cases} \lfloor n/2 \rfloor + 1 & \text{when } n \text{ is odd.} \\ \lfloor n/2 \rfloor + 2 & \text{when } n \text{ is even.} \end{cases}$$

Proof:

Let P_n be a path of order $n \geq 3$ and 1

Let $V(P_n) = \{v_1, v_2, v_3, \dots, v_n\}$ the central NIC graph $C(P_n)$ is obtained by subdividing each covalent bond of $v_i v_{i+1}$, $1 \leq i \leq n-1$ of P_n exactly one by adding new atom c_i in $C(P_n)$ and joining each atom v_j , $1 \leq i \leq n-2$ with each atom v_k , $j+2 \leq k \leq n$.

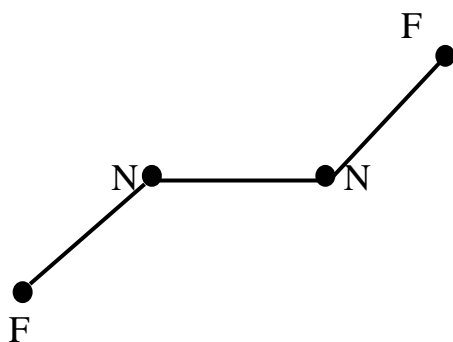
Let $V_1 = \{v_1, v_2, v_3, \dots, v_n\}$ and $V_2 = \{c_1, c_2, c_3, \dots, c_{n-1}\}$. Then $V(C(P_n)) = V_1 \cup V_2$ Relabel the atom of $C(P_n)$ by $u_1 = v_1$, $u_2 = c_1$, $u_3 = v_2, \dots, u_{2n-1}$ alternatively.

For the following procedure gives a Dominator Coloring of $C(P_n)$. When

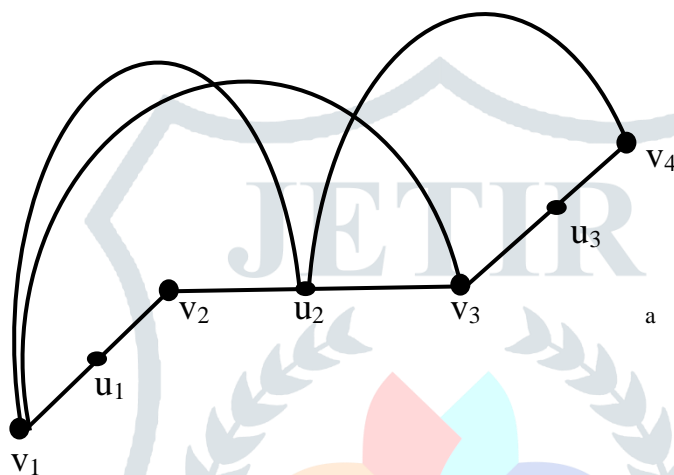
$n \geq 4$, the atom u_1 is colored by color 1, atom u_2 is colored by a color 2 and the vertex u_i $i = 4, 6, 8, \dots, 2n-2$ is colored by color 1.

If n is odd, the atom u_i , $i = 5, 9, 13, \dots, 2n-1$ is colored by color 2 and the vertices u_j , $j = 3, 7, 11, \dots, 2n-3$ are respectively colored by individual color 3, 4, 5, ..., $\lfloor n/2 \rfloor + 1$. If n is even, the atom u_i , $i = 5, 9, 13, \dots, 2n-3$ is colored by color 2 and the atoms u_j , $j = 3, 7, 11, \dots, 2n-1$ are colored respectively by individual colors 3, 4, 5, ..., $\lfloor n/2 \rfloor + 2$. When $n = 3, 4$, or 5, the atoms of $C(P_n)$ are colored by the color sequences (1, 2, 1, 3, 2) (1, 2, 3, 1, 2, 4, 2, 1) in order to get a dominator coloring. When $n \geq 4$, the atom u_1 dominates the color class of u_7 .

If n is odd, atom u_i and u_{i+2} , $i = 2, 6, 10, 14, \dots, 2n-3$ dominate the color class of u_{i+1} . The atom u_j , for $j = 3, 7, 11, \dots, 2n-3$ dominates itself. If n is even, atom u_i and u_{i+2} , $i = 2, 6, 10, 14, \dots, 2n-6$ dominate the color class of u_{i+1} and u_{2n-2} dominates the color class of u_{2n-1} . The atom u_j for $j = 3, 7, 11, \dots, 2n-3$ dominates itself. The atom u_j , for $j = 5, 9, 13, \dots, 2n-1$ dominates either u_{j+6} , u_{j-6} or both.



N₂F₂(Nitrogen Floride)



Molecular of the Central graph of P₄ is depicted with dominator coloring

The color classes of $c(P_4)$ are

$$V_1 = \{ u_1, u_2, u_3 \}, V_2 = \{ v_1, v_2 \}, V_3 = \{ v_3 \}, V_4 = \{ v_4 \}$$

The dominator chromatic number,

$$\chi_d [C(p_n)] = [n/2] + 2 \text{ when } n \text{ is even.}$$

$$= [4/2] + 2$$

$$= 2 + 2$$

$$= 4$$

Hence,

$$\chi_d [C(p_n)] = [n/2]+1 \text{ when } n \text{ is odd}$$

$$[n/2]+2 \text{ when } n \text{ is even.}$$

Some Structural Properties of C(P_n) has

- i) (n+1) vertices of degree n.
- ii) n vertices of degree 2.

iii) The number of Vertices in $P_c(P_n) = (2n+1)$.

iv) The number of edges in $q_c(P_n) = \frac{n^2+3n}{2}$.

CONCLUSION

In this paper, We consider the Neighbourly Irregular Chemical Graph only among the p – block elements and further we constructing Dominator Chromatic Number of above graph.

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