# FORMULATION AND EVALUATION OF MUCCOADHESIVE BUCCAL PATCH CONTAINING DILTIAZEM HYDROCHLORIDE

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**Abstract** : The present study aims to formulate mucoadhesive buccal drug delivery system of Diltiazem Hydrochloride, in view to improve the bioavailability of drug; which undergoes extensive hepatic first-pass metabolism, leading to increased dose and side effects. The concentration of polymers like HPMC-E-5, PVA, and plasticizers like Glycerin, Dibutyl-phthalate, and solvents was optimized through preliminary study. Depending upon the results of Trial Batches, the concentration of polymers and plasticizers was optimized. The results were analyzed. The optimized formulation batch was found to be D4D, containing diltiazem hydrochloride 30mg, PVA 250mg, DBP 5%, and D/W 2ml and Acetone 8ml. The stability study of D4D formulation revealed that formulation was stable.

#### Key Words: Diltiazem hydrochloride, buccal delivery, buccal patches.

#### INTRODUCTION

Diltiazem hydrochloride is Diltiazem is a potent vasodilator increasing blood flow and variably decreasing the heart rate via strong depression of A-V node conduction. Its pharmacological activity is somewhat similar to verapamil. Potent vasodilator of coronary vessels.Vasodilator of peripheral vessels. This reduces peripheral resistance and afterload. Negative inotropic effect. Diltiazem causes a modest decrease in muscle ability and reduces heart muscle chemical element consumption. Negative chronotropic effect. Diltiazem causes a modest lowering of heart rate. This effect is due to the slowing of the SA (sinoatrial) node. It results in reduced myocardium oxygen consumption. Negative dromotropic effect. By deceleration conductivity through the Av (atrioventricular) node, diltiazem increases the time needed for each beat. This ends up in reduced heart muscle chemical element consumption by the body.

Systemic bioavailability of diltiazem hydrochloride following oral administration is about 40%, due to extensive presystemic metabolism in the gut wall and wall 75% - 80%.plasma half-life of diltiazem hydrochloride is 3 - 4 hrs.molecular weight is 414.51, and it does not have an objectionable taste. All these factors and problems associated with oral and parenteral routes make diltiazem hydrochloride an appropriate candidate for Bucco-adhesive formulations.

#### MATERIALS AND METHODS

CHEMICALS: Diltiazem hydrochloride, HPMC – E – 5, Polyvinyl alcohol, Glycerin, Di-butyl phthalate, Acetone.

#### **Instruments/Equipments:**

- I. UV double beam spectrophotometer(Shimadzu Corporation, Japan)
- II. FTIR 200 Spectrometer(Spectrum one, PerkinElmer, USA)
- III. Hot air oven(spectrumPvt. Ltd)
- IV. Franz diffusion cell( Bhanu Scientific Instruments Co. Bangalore)
- V. Electronic Balance (CG 153)(Scaltech)
- VI. Screw gauge(Dolphin)
- VII. Magnetic stirrer(Remi Equipments,Mumbai, India)
- VIII. pH-meter(Elico India Pvt. Ltd)
- IX. Sonicator(Dolphin)
- X. Stability chamber (Thermolab India)

#### METHODOLOGY

#### **IR** spectrum interpretation

The infrared spectrum of the pure Diltiazem Hydrochloride sample was recorded and the spectral analysis was done. The dry sample of the drug was directly placed after mixing and triturating with dry potassium bromide.

# UV spectroscopy:

### I. Determination of $\lambda$ max

A 10mg of Diltiazem Hydrochloride was accurately weighed and was first dissolved in 35ml methanol. The solution was then diluted using phosphate buffer (pH- 7.4) to 100 ml. The UV spectrum was recorded in the wavelength range of 200-400 nm.

## II. Preparation of calibration curve for Diltiazem hydrochloride

A standard curve was prepared by dissolving 10 mg of diltiazem hydrochloride in 50ml of d/w. It was further diluted with d/w to get solutions in the concentration range of 3 to 15  $\mu$ g /ml. The absorbances of these solutions were determined spectrophotometrically at 237 nm.

# FORMULATION OF BUCCAL PATCHES:

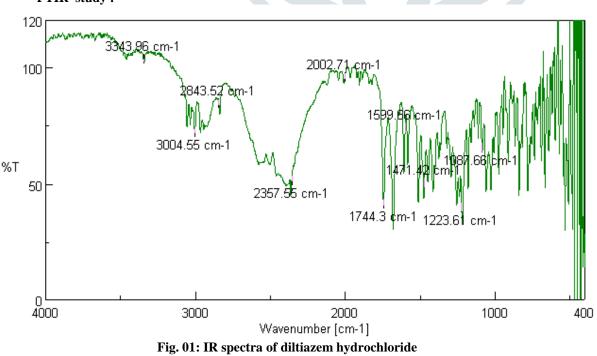
### Formulation of Drug Incorporated Buccal Patches:

The reservoir-type buccal patches containing diltiazem HCl were prepared using different ratios of HPMC -E-5 and polyvinyl alcohol. The polymers in different ratios were dissolved in the respective solvents. Then the drug was added slowly in the polymeric solution and stirred on the magnetic stirrer to obtain a uniform solution. Di-n-butyl phthalate and glycerin were used as plasticizers. Then the solution was poured on the Petri dish having a surface area of 12.99 cm2 and dried at the room temperature. Then the patches were cut into 1cm2 patches.

## Table: 1.Formulation ingredients of DHCL patches

FORMULATION CODE	DRUG (DHCL)	POLYMER	PLASTISCISER	SOLVENT
D1G	30mg	HPMC-E-5	Glycerin10%	D/W : ACETONE
		250gm		2M:8ML
D2G	30mg	HPMC-E-5	DAB 5%	D/W : ACETONE
		250gm		2M:8ML
D3G	30mg	HPMC-E-5	Glycerin10%	D/W : ACETONE
		250gm		2M:8ML
D4G	30mg	HPMC-E-5	DAB 5%	D/W : ACETONE
		250gm		2M:8ML

#### **RESULT& CONCLUSION:** FTIR study :



# UV spectroscopy

#### I. Determination of $\lambda$ max :

The peak showed in the figure is much similar to the reported peak. The spectrum obtained is shown in the figure i.e. 237nm.

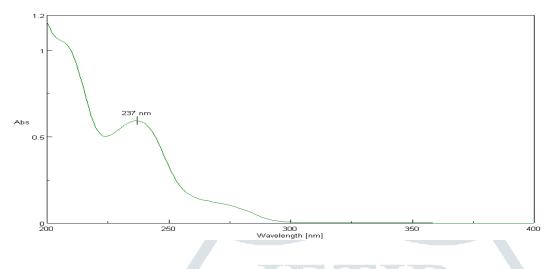


Fig.02: UV spectrum of diltiazem hydrochloride

# EVALUATION OF BUCCAL PATCHES:

# table 02: physical characterization of buccal patch

SK	CHARECERIZATION	OBSERVATION
1	SHAPE	ROUND
2	COLOUR	TRANSPARENT
3	APPEARANCE	SMOOTH



fig.03buccal patch

table :weight variation, thickness and % moisture absorption and water vapor transmission of buccal patch

F O F c	RMULAT od		Waight variation ( m g )	Thickness (mm)	% MOISTURE UPTAKE TATER TAPOUR TRANSMISSION, GM (M-2 A-
D	1	G	3 2 9 . 2 5 ± 7 . 3 6 5	$0.98 \pm 0.01$	4 . 5 % $3.930 \times 10-6 \pm 0.384$
D	2	D	3 3 1 ± 7 . 9 5 8	0.98±0.01	4 . 8 % $2.216 \times 10-6\pm 0.580$
D	3	G	3 3 6 . 5 ± 2 . 3 8 0	$0.98 \pm 0.01$	5 . 0 1 % $1.634 \times 10-6\pm 0.226$
D	4	G	3 5 3 ± 8 . 5 2 4	0.98±0.01	5 . 8 % $1.882 \times 10-6\pm 0.182$

n=3; standard deviations for three determinations

Table03: weight variation, thickness and % moisture absorption and water vapor transmission of buccal patch

Tensile strength and % elongation at break of buccal patch
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FOR C	M U L A T I O D	O N E	T	ΕN	S	I L	E	S T	<b>R</b> ]	ΕN	GΤ	Н	%	ΕL	0 N	G A	ΤI	O N
D	1	G	2	•	8	9	g	m	/	m	m	2	0	8		3	3	%
D	2	D	3		2	0	g	m	/	m	m	2	1	6		8	8	%
D	3	G	3	•	5	7	g	m	/	m	m	2	3	3		3	3	%
D	4	D	3	•	5	6	g	m	/	m	m	2	5	8		8	8	%

n=3; standard deviations for three determinations

Table 04: Tensile strength and % elongation at break of buccal patch

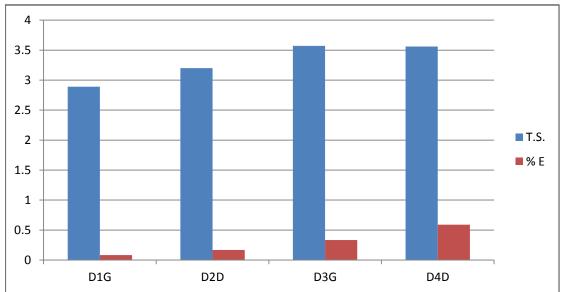


Fig.04: Tensile strength and % elongation at break of buccal patch surface pH and folding endurance of buccal patch

	pir una roranig										
FOF	RMULAT	ION	S	URFA	C E	рН	FO	LDING	ENI	DURANO	СЕ
D	1	G	6		1	6	2	0	8	±	4
D	2	D	5			8	2	1	6	±	2
D	3	G	6			9	2	2	1	±	5
D	4	D	7			2	2	8	4	±	3

n=3; standard deviations for three determinations

Table 05:.surface pH and folding endurance of buccal patch

Drug content	of the buc	cal patch

SR.NO	. FORM	AULATION	CODE	D	R	U	G	С	0	N ′	ΓF	N	Т
1	D	1	G	9	8	•	8	9	±	0	•	6	4
2	D	2	D	9	8	•	6	2	±	0		7	2
3	D	3	G	9	9	•	3	1	±	0		3	7
4	D	4	D	9	9	•	3	4	±	0		9	5

n=3; SD for three determinations

Table 06: drug content of the buccal patch

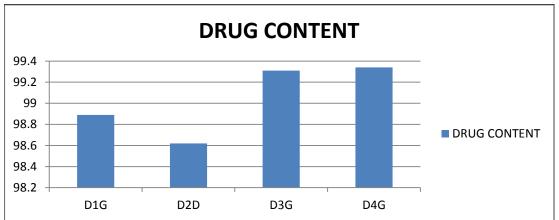


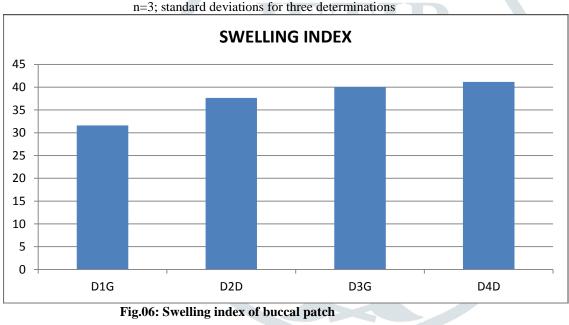
Fig.05: drug content of the buccal patch

# SWELLINGINDEX :

The comparative percentage swelling of various formulation shown in table and fig.also.

Table 07: swelling index of buccal patch

S     R     N     O     .     FORMULATION CODE     S W E L L I N G     I N       1     D     1     G     3     1     .     6     2     ±     0     .       2     D     2     D     3     7     .     6     4     ±     0     .	<b>D E X</b> 8 9
	89
2 D 2 D $3.7.64 \pm 0.9$	0 (
	96
3 D 3 G 4 0 . 0 2 $\pm$ 1 . 4	1 6
4 D 4 D 4 1 . 1 6 ± 1 . 2	4 2



# **DIFFUSION STUDY:**

Among all four formulations shows that formulation D4D has better drug release compared to other formulations.

Table 08:	diffusion st	tudy of D4D																			
ΤΙΜΕ	PURE DRUG	A B S	С	0	N C	•	A	V	G		%	% (	C D	D 4 D	<b>R</b> )	%	L	0	G	C D	R
5	0.0623	0.0124	0		0	3	0		1	0	6	0		1 0	6	0		7	9	5	8
1 0	0.087	0.0488	0		0	3	0		1	0	6	0		2 1	2	0		6	7	3	6
1 5	0.245	0.0882	0		1	1	0		3	5	3	0		5 6	5	0		2	4	7	1
3 0	0.363	0.189	0		•	3	0		9	9	2	1		5 5	7	0		1	9	2	2
4 5	0.454	1.128	0		•	4	1		3	3	5	2		8 9	8	0		4	6	2	0
6 0	6.736	1 . 3 2	2		0	6	6		8	9	2	9		7 4	4	0		9	8	8	7
9 0	14.95	2.122	2		5	6	8		5	4	6	1	8	. 2	9	1		2	6	2	2
1 2 0	24.92	2.462	4		2	1	1	4		0	2	3	2	. 3	1	3		5	0	9	3
1 5 0	36.41	2.652	5		0	6	1	6		8	7	4	9	. 1	8	1		6	9	1	7
1 8 0	49.92	5.344	5		4	8	1	8		2	5	6	7	. 4	3	1		8	2	8	8
2 4 0	66.24	5.963	1	1	. 0	6	3	6		8	7	8	9		3	2		1	4	1	7
3 0 0	84.82	7	1	2	. 9	6	4	3			2	9	8		3	2		1	1	6	8
3 6 0	103.4	7.001	1	5	. 7	1	1	0	3.	0	4	1	2 3	. 8	9	2		1	7	8	6

#### DIFFUSION STUDY OF D4D: Table 08: diffusion study of D4D

D4D

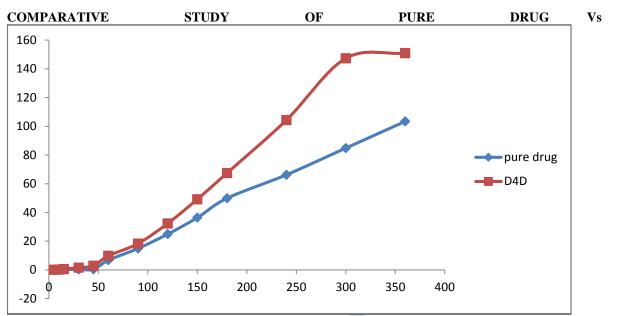


Fig.07: a comparative study of pure drug Vs D4D

#### STABILITY STUDY: Table 00: stability study

1 au	e 09: stability s	tuay				
FO C	RMULATI O D	ON E	ТІМЕ	A P P E A R A N C E	THICKNESS(mm)	DRUG CONTENT
D	1	G	1 m o n t h	Smooth	0 . 9 8	$8 0 . 8 0 \pm 0 . 9 8$
D	2	D	1 m o n t h	Rough	0.98	$8 \ 9 \ . \ 6 \ 2 \ \pm \ 0 \ . \ 6 \ 7$
D	3	G	1 m o n t h	Smooth	0.98	9 0 . 0 0 $\pm$ 0 . 1 2
D	4	D	1 m o n t h	Smooth	0.98	9 9 . 3 4 ± 0 . 1 1

#### REFERENCES

- 1)Gupta, S. Garg, R. K. Khar, Mucoadhesivebuccal drug delivery system-A review, Indian drugs 29 (1992) 586-593.
- 2)D. Harris, J. R. Robinson, Drug delivery via the mucous membrane of the oral cavity, J. Pharm. Sci. 81 (1992) 1-10.
- 3)J. K. Pandit, N. M. Vemuri, S. P. Wahi, R.B. Jayachandra, Mucosal dosage form of ephedrine hydrochloride using Gantrez-AN 139, The Eastern pharmacist (1993) 165-170.
- 4)G. Tocker, A method to study the kinetic of oral mucosal drug absorption for a solution, Chem. Pharm. Bull. 40 (1998) 679-683.
- 5)H.E. Junginger, J.A. Hoogstrate, J.C.Verhoef, Recent advanced in buccal drug delivery and absorption *in-vitro* and *in-vivo* studies, J. Control. Rel. 62 (1999) 149-159.
- 6)H. Shojaei, Buccal mucosa as a route for systemic drug delivery. A review, J. Pharm. Pharmaceut. Sci. 1 (1998) 15-30.
- 7)R. K. Khanna, S. P. Agarwal, A. Ahuja, Mucoadhesivebuccal drug delivery: A potential alternative to conventional therapy, Ind. J. Pharm. Sci. 60 (1998) 1-11.
- 8)H.H. Alur, A.K. Mitra, T.P. Johnston, Modified-release drug delivery technology, Ed: M.T.Rathbone, J. Hadgraft, M.S.Roberts, Marcel Dekker, Inc, New York, 2002, pp. 401-417.
- 9)P. M. Mandaogode, Studies on buccal absorption kinetics of promethazine hydrochlorides, M. Pharm. thesis. U.D.P.S. Nagpur, (1997).
- 10) J. Hao, P. W. S. Heng, Buccal delivery system, Drug Dev. Ind. Pharm. 29 (2003) 821-832.
- 11) M.J. Rathbone, G. Ponchel, F. A. Ghazali, Oral mucosal drug delivery, Volume 74, Marcel Dekker, New York, 1996, pp. 248-261.
- 12) M. N. Chidambaram, A. K. Srivastava, Buccal drug delivery system, Drug Dev. Ind. Pharm. 21 (1995) 1009-1036.
- 13) D. M. Brahmankar, S. B. Jaiswal, Biopharmaceutics and pharmacokinetics A. Treatise, Vallabhprakashan, Dehli, 1999, pp. 71.
- 14) D. M. Brahmankar, S. B. Jaiswal, Biopharmaceutics and pharmacokinetics A. Treatise, Vallabh Prakashan, Delhi, 1999, pp. 64-65.

- 15) R.khanna, S.P.Agrwal, Mucccoadhesivebuccal drug delivery a potential alternative to conventional therapy, I.J.P.S.,1988,60(1),page no. 1-11.
- 16) Sanket D. Gandhi, Priyanka R. Pandya, Mucoadhesive drug delivery systems-an unusual maneuver for site-specific drug delivery system. I.J.P.S., vol 2 july 2011 1-21.
- 17) Anay R. Patel, Dhagash A. Patel and Sharad V. ChaudhryMucoadhesivebuccal drug delivery system.I.J.of Pharmacy and life sciences.June, 2011 1-9
- 18) ShaffiKhurana and N.V. SatheeshMadhavMuccoadhesive Drug Delivery System: Mechanisam and Methods of Evaluation,International Journal of Pharma and Bio-science.Vol. -2,March 2011.1-9.
- 19) Latheeshjlal.L, Sunil Murala, Muccoadhesive Drug Delivery System: An Overview International Journal of PharmTech. Research coden (USA): IJPRIF I: Vol. 3, No.1, pp 42-49, Jan-Mar 2011.
- 20) DebjitBhowmik,Muccoadhesivebuccal drug delivery systems an overview.page no.1-18.
- 21) Amir H. Shojaei, Richard K.Chang, XiaodiGuo, Beth A. Burnside, and Richard A.Couch, Systemic Drug Delivery via the Buccal Mucosal Route, pharmaceutical technology, june 2001, page no.1-12.
- 22) Rajesh Mujoriya, A Review on study of Buccal Drug Delivery System, Innovative Systems Design and Engineering www.iiste.org ISSN 2222-1727 (Paper) ISSN 2222-2871 (Online) Vol 2, No 3.
- 23) Gaurav Kumar Sharma, Pramod Kumar Sharma, MayankBansal, A Review on MuccoadhesiveBuccal patch as a novel drug delivery systems, IJPS, page no.1378-1387.
- 24) Kumar V, Aggarwal G, Zakir F, Choudhary ABuccalbioadhesive drug delivery a novel technique,International Journal of Pharmacy and Biological Sciences(eISSN: 2230-7605) www.ijpbs.com IJPBS ,Volume 1,Issue 3,JULY-SEPT 2011,89-102.
- 25) Khar K, Ahuja A, Javed A: Mucoadhesive Drug Delivery, Controlled and Novel DrugDelivery by Jain NK., First edition, Chapter-16, New Delhi; 1997.
- 26) Vamshi Vishnu, K. Chandrasekhar, G. Ramesh and Y. MadhusudanRao, Development of Mucoadhesive Patches for Buccal Administration of Carvedilol.Current Drug Delivery, 2007, 4, 27-39.
- 27) C.V.S. Subrahmanyam, J. Thimmasetty, D.S. Manish Kumar, Design and Evaluation of FexofenadineHClBuccalMucoadhesive Patches. E-mail :<u>thimmasetty@yahoo.com.page</u> no. 1-18.

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