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Formulation and Evalution of Floating Sustained **Release Matrix Tablet of NSAIDs**

Gaurav Singh Sikarwar, Khushboo Lavania, Vikash Sharma, Vishal Sharma, Gaurang Sharma*

Affiliation- Anand College of Pharmacy, Keetham, Agra (U.P.), Pin-282007

Abstract

To increase lornoxicam's stomach residence time, this study aims to develop floating sustained release matrix tablets. Because of its potent oxicam group of non-steroidal anti-inflammatory drugs' relatively short half-life of 2 to 3 hours and maximal absorption in the proximal gastro intestinal tract region, lornoxicam must be produced as floating sustained release matrix tablets. In the current investigation, gelatin and guar gum were preserved in a variable ratio along with high-grade hydroxyl propyl methyl cellulose K4M, a polymer with an apparent viscosity of 15,000 cps, and calcium carbonate and citric acid were used as a gas generator.

Keywords: Lornoxicam, FTIR, Guar gum, Xanthan gum, NSAIDs

1. Introduction

Controlled release medication administration has just been the norm in contemporary pharmaceutical design, and extensive study has been done to improve the efficacy, dependability, and safety of therapeutic products. The most common drug delivery method will continue to be oral sustained release medicine. In order to circumvent first pass metabolism and boost bioavailability, tablets were created in this study. 1 Given the above, it is clear that many medications are not suitable for conventional sustained release formulations that are retained in the stomach and release the medication in the intestine. It is also clear that such medications would benefit greatly from a sustained release formulation that is retained in the stomach and releases the medication slowly over an extended period of time. ² After oral dosing, bioavailability is 20%. Silent design elements reduce the chance of dose dumping in sustained release pills. less variation between and within subjects. High degree of intestinal dispersion reduces the possibility of high local medication concentrations. Drug bioavailability is reproducible when it successfully reaches the optimal absorption location. Drug transport does not depend on stomach emptying.³

Nsaids for inflammation medicine (NSAID) lornoxicam, which belongs to the oxicams class, has strong analgesic and anti-inflammatory effects. Lornoxicam's painkilling and anti-inflammatory qualities, like those of other NSAIDs, are attributed to its reduction of prostaglandin synthesis through inhibition of cyclo-oxygenase (COX) activity. The ratio of COX-1 to COX-2 inhibition is the same for both isoforms of COX when lornoxicam is used. It quickly reaches the synovial fluid. However, unlike several NSAIDs, Lornoxicam does not reduce 5-lipoxygenase activity and does not shunt arachidonic acid to the 5-lipoxygenase pathway or inhibit leukotriene production. The opioid neuropeptides system's activation may be a factor in how effective Lornoxicam is at relieving pain.⁴

The most popular method of medicine administration is via the oral route. the bulk of oral dosages are in tablet form ⁽¹⁹⁾. They are widely used because they are simple to apply (patients are more likely to comply) and simple to prepare on an industrial scale. The majority of oral tablet formulation represents the so-called immediate release (IR) dosage form. Plasma concentration of drug with an IR dosage form generally raises quickly, peaks, and then declines. If the elimination of drug is fast this result in only a short period during which the plasma concentration of the drug is within the therapeutic window. In figure,

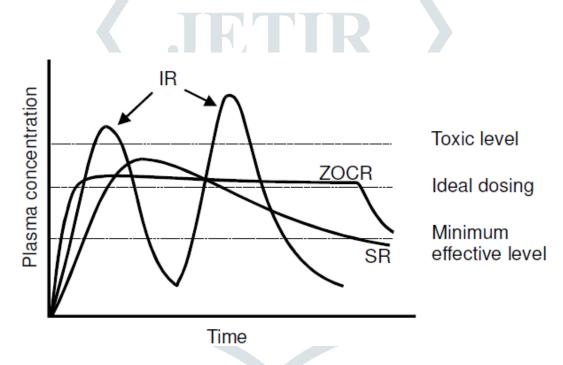


Fig-1.1 Plasma concentrations of an idealised zero order control release (zocr) dose form and a conventional immediate release dosage form are shown in a characteristic way (in combination with a start up dosage)

Material Profile & Method

List of Material & Apparatus which are used in preformulation studies is presented in Tables 5.1 and 5.2.

Table 5.1 List of material used

Sr. No	Material	Manufacturer/Suppliers	
1	Lornoxicam	Morden labs, Indore	
3	Guar Gum	Titan Biotech, Bhivandi, Rajasthan.	
4	Xanthane gum	HiMedia Laboratories Pvt Ltd., Mumbai	
5	Ethanol	Jiangsu Huaxi International Trade Co. Ltd Made in China	
6	Sodium bi carbonate	Thermo Fisher Scietific India Pvt. Ltd, Mumbai	
7	Citric acid	Finar Chemical (India) Pvt Ltd., Ahmedabad.	
8	Calcium carbonate	HiMedia Laboratories Pvt Ltd., Mumbai.	
9	Magnesium stearate	HiMedia Laboratories Pvt Ltd., Mumbai.	
10	Talc	HiMedia Laboratories Pvt Ltd., Mumbai.	

Table 5.2 List of Apparatus

Sr. No	Instrument	Manufactures
1	Weigh Balance	Wensar
2	Melting Point	Khera Industries, Delhi
3	U.V.	Systronic double beam spectrophotometer 2203
4	FTIR	Bruker
5	DSC	

Method

Parameters for Preformulation Study:

- Physical appearance
- Solubility estimation
- Melting point
- Standard absorbance/calibration curve for UV estimate
- Partition coefficient
- FT-IR determination (Compatibility study)
- **1.Outer appearance:** The powdered medication (lornoxicam) was tested for its organo-leptic characteristics, including colour and odour, which were somewhat yellowish.
- **2. Estimated solubility:** The sample's solubility in various solvents was evaluated qualitatively. According to IP, it was determined by putting 10 mg of the drug sample in 10 ml of water, methanol, ethanol, pH buffer

6.8 and pH buffer 7.4 in test tubes and thoroughly blending it by shaking. Table 7.1 lists various solubility terms.

Table 7.1 Various Solubility Terms

Term for describing	parts of the solute needed for
	each part of the solvent
highly soluble	< 1
completely soluble	1 - 10
Soluble	10 - 30
minimally soluble	30 - 100
hardly soluble	100 - 1000
extremely hardly soluble	1000 - 10000
Insoluble or almost insoluble	10,000 more

3. Calculating the melting point:

The melting point was determined using the capillary method utilising a digital melting point apparatus. By gently pressing the open end into a pure drug sample and tapping the capillary's base on a hard surface to force the drug into the tube's bottom, the capillary tube was linked, filled, and packed. The gadget was switched on, the tube with the drug packed inside was put into the slot, and the temperature at which the medication melted was recorded.

4. Development of calibration curves: Acyclovir solution was scanned using a Systronic Double Beam UV Visible Spectrophotometer in the UV range of 200-400 nm (true wavelength about 251 nm to 390 nm).

4.1 Determination of the Maximum Absorbance Wavelength (λ max)

Accurately weighing 10 mg of the medication (aciclovir), it was then put to a 10 ml volumetric flask. After that, 0.1 N HCL was added to completely dissolve the medication. Solvent was used to increase the volume to 10 ml. A 1000 g/ml prepared sample was used. Then, 1 ml of the aforementioned solution was transferred to a second volumetric flask with a capacity of 10 ml and diluted therein. A 100 g/ml prepared sample was used. When 1 more millilitre of this solution was taken and diluted up to 10 millilitres, it produced a solution with a concentration of 10 g/ml. Now, scan the sample in the UV Spectrophotometer between 200 and 400 nm. The same steps were taken to prepare distilled water.

4.2 Making the Lornoxicam Calibration Curve

Between concentration and absorbance, the calibration curve was drawn. In 0.1 N HCL and distilled water, the calibration curve for 2-12 µg/ml was performed.

5. partition coefficient determination

For equilibration, 25 mg of the medication was divided among three separating funnels, which were shaken for two hours in a wrist action shaker. The partition coefficient of the drug in phases was estimated using the following formula after two phases were separated and the amount of the drug in the aqueous phase was spectrophotometrically analysed by using formula:

Partition Coefficient, $K = \frac{Amount of drug in organic layer}{Amount of drug in aqueous layer}$

6. A compatibility analysis of drugs and excipients

Drug – Excipient compatibility study done by FT-IR spectroscopy. The presence of characteristic peaks associated with specific structural characteristics of the drug molecule was noted.

Result and Discusion:

5.1 Solubility: Solubility profile of lornoxicam is shown in table 5.1

 Table 5.1 Solubility Profile of Lornoxicam

Sr. No.	Medium	Solubility Profile	Parts of Solvent
1	Water	Slightly Soluble	100 - 1000
2	Methanol	Vey slightly soluble	10 – 30
3	Ethanol	Very slightly soluble	1000 – 10,000
4	0.1 N hydrochloric Acid	Soluble	10 - 30
5	0.1 N Sodium Hydroxide	Insoluble	>10,000
6	Acetone	Slightly Soluble	100 -1000
7	Simulated Gastric Fluid (pH 1.2)	Slightly Soluble	100 - 1000
9	Chloroform	Insoluble	>10,000

Where * The average of three determinations: easily soluble = 1-10 parts of solvent, soluble = 10-30 parts of solvent, sparingly soluble = 30-10 parts of solvents, little soluble = 100-1000 parts of solvent, and very slightly soluble = 1000-10000 parts of solvent.

5.2 Melting Point: Melting point of lornoxicam is shown in table 5.2

Table 5.2 Melting Point Range of Iornoxicam

Sr. No.	Onset	Complete	Melting Point
1	225° C	230° C	220 ± 2°C
2	220° C	228° C	

5.3 Determination of λ max and Preparation of Calibraton Curve of lornoxicam in 0.1N HCl Solution

Between concentration and absorbance was plotted the calibration curve. It was done using the calibration curve of 2–10 g/ml. The calibration curve's slope and intercept were 0.053 and 0.027, respectively. The

computed value of the correlation coefficient, or "r2," was 0.990. Table 8.3, Figure 8.4, and the Statistical Parameters of Aciclovir demonstrate the calibration curve.

5.4 Determination of λ max and Calibration curve:

Using a Sistronic UV Visible spectrophotometer, the UV range of 200–400 nm was scanned on an acyclovir solution. It was discovered that the spectrophotometric method of examination of lornoxicam at max 370.4 nm was highly sensitive and reproducible. Acyclovir standard curves were created in distilled water and 0.1N HCl solution at a maximum wavelength of 370.4 nm. To generate the straight line, the data were regressed. In every instance, a correlation coefficient larger than 0.99 was found, indicating that the drug complies with Beer-rule Lambert's in the concentration range of 2–10 g/ml. Figures 8.1 and 8.2 display the determination max.

5.4.1. Preparation of Calibraton Curve of lornoxicam in 0.1N HCl Solution

The calibration curve was plotted between the concentration and absorbance. The calibration curve of 2-10µg/ml was carried out. The slope and intercept of the calibration curve were 0.053 and 0.027 respectively. The correlation coefficient 'r²' values were calculated as 0.990. Calibration curve and Statistical parameters of lornoxicam is shown in table 8.3, 8.4 and figure 8.4.

5.4.2. Calibration Curve: Calibration curve and Statistical parameters of lornoxicam is shown in table 5.3 and 5.4

Table 3.5 Candi ation Cut ve of for hoxicam in Different Medium at Amax 57 0.4 min	Table 5.3 Calibration	Curve of lornoxicam in	Different Medium at λ_{max} 370.4 nm
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Sr. No.	Conc (µg/ml)	0.1 N Hydrochloric Acid
1	2	0.016
2	4	0.024
3	6	0.04
4	8	0.05
5	10	0.062

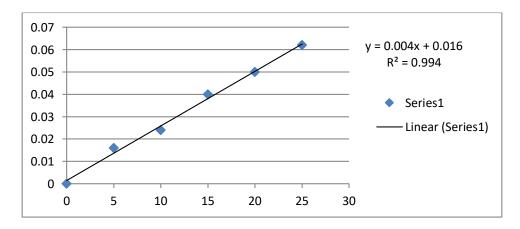


Fig 5.4 Standard curve in 0.1 N HCL of lornoxicam

Table 5.4 Statistical Parameter Related to Standard Curve of lornoxicam at λ_{max} 370.4 nm

		Regressed Line	y = 0.05 x - 0.016
		Equation $(y = mx + c)$	
1	Standard Curve In 0.1 N	Beer's Law Range	1 -10 μg/ ml
	Hydrochloric Acid	Regression	$R^2 = 0.994$
	(pH1.2)	Coefficient	
		Regressed Line	y = 0.004x - 0.016
		Equation $(y = mx + c)$	

5.5 Partition Coefficient: partition coefficient of drug is shown in Table 5.5

Table 5.5 Partition Coefficient Values of Drug

Sr. No.	Medium	Partition Coefficient
	JEII	(Log P)
1	n – octanol : Water	-1.454
2	Cyclohexane : Water	-0.987

5.6 FTIR Study

Fig 5.5 FTIR Spectra of LORNOXICAM

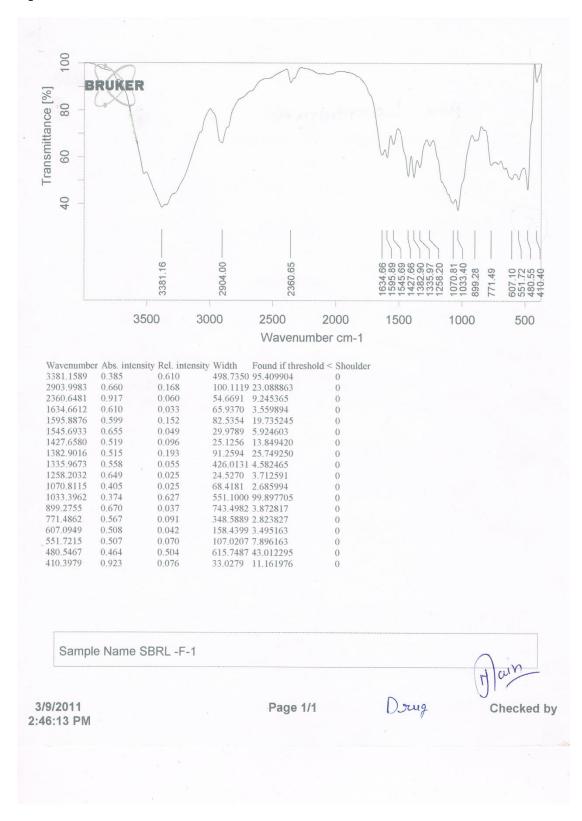


Fig 5.6 FTIR Spectra of Lornoxicam + Gaur gum

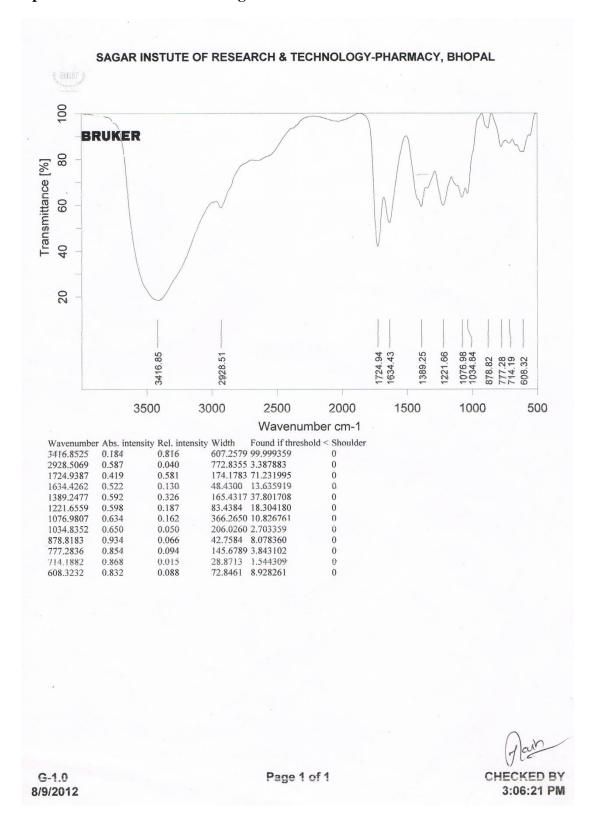
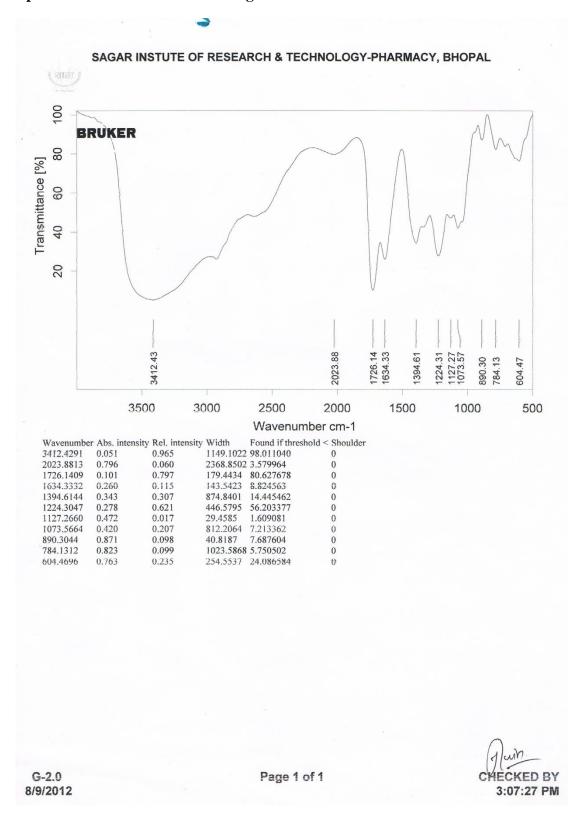


Fig 5.7 FTIR Spectra of Lornoxicam xanthane gum



References

- 1. Jantzen GM, Robinson JR, Sustained and controlled-release drug delivery systems, in Banker GS, Rhodes CT (Eds.) Modern Pharmaceutics, Third Edition, Revised and Expanded, Drugs and the Pharmaceutical Sciences, vol 72, Marcell Dekker, Inc. New York, 1995: 575-609.
- 2. Sheth PR, Novel sustained release tablet formulation, US Patent 4167558, 1979.
- 3. H Bechgaard, G H Nielson. Controlled release multiple units and single unit dosage; Drug Dev. & Ind. Pharm., 1978; 4(1): 53-67
- 4. Jiménez-Martínez I, Quirino-Barreda T, Villafuerte-Robles L. Sustained delivery of Captopril from floating matrix tablets. Int J Pharm 2008;362:37-43.
- 5. Arora G, Malik K, Singh I, Arora S, Rana V. Formulation and evaluation of controlled release matrix mucoadhesive tablets of domperidone using Salvia plebeian gum. J Adv Pharm Tech Res 2011;2:163-9
- 6. Nagaich U, Chaudhary V, Tonpay SD, Karki R. Design and evaluation of ametronidazole central core matrix tablet. J Adv Pharm Tech Res 2010;1(1)-88-96.
- 7. Sathiyaraj S, Devi RD, Hari VB. Lornoxicam gastro retentive floating matrix tablets:Design and in vitro evaluation. J Adv Pharm Tech Res 2011;2:156-62.
- 8. Hamza Yel-S, Aburahma MH. Design and *in vitro* Evaluation of novel sustained-release double-layer tablets of Lornoxicam: Utility of cyclodextrin and xanthan gum combination. AAPS Pharm Sci Tech 2009;10:1357-66.
- 9. Park S, chun M, Choi H, preparation of an extended release matrix tablet using chitosan/ carbopol interpolymer complex, International journal of pharmaceutics, Volume 347, 1 (2), Page no 39-41
- 10. Jamzad S, Fassihi R. Development of a controlled release low dose class II-Glipizide. Int. J. Pharm. 2006; 312: 24-32.
- 11. Nokano M, Ogata A. *In vitro* release characteristics of matrix tablets: Study of Karaya gum and Guar gum as release modulators. Ind. J. Pharm. Sc. 2006; 68(6): 824-826.
- 12. Kang K, Pettitt D. Industrial Gums: Polysaccharides and Their Derivatives, 3rd Ed.; Whistler R, Bemiller J. Eds.; Academic Press, 1993; 341–397.
- 13. Remington: The Science and Practice of Pharmacy, 19th Ed.; Mack Publishing Company: Pennsylvania, 1995.
- 14. Dhopeshwarkar V, Zatz J. Evaluation of xanthan gum in the preparation of sustained release matrix tablets. Drug Dev. Ind. Pharm. 1993, 19, 999-1017.
- 15. 12. Fu L, Woodward L, Borodkin S. Xanthan Gum and Alginate Based Controlled Release Theophylline Formulations. Drug Dev. Ind. Pharm. 1991; 17 (14): 1987-2004.
- 16. Khullar P, Khar RK, Agrawal SP. Evaluation of hydrogel-based controlled-release niacin tablets. Drug Dev. Ind. Pharm. 1998; 24 (5): 479–483.

- 17. Bhalla HL, Shah AA. Controlled release matrices for ketoprofen. Indian Drugs 1991; 28 (9): 420-422.
- 18. Jain NK, Kulkarni K, Talwar N. Controlled-release tablet formulation of isoniazid. Pharmazie 1992; 47 (4): 277–278.
- 19. Oral control release matrix tablets literature review Altaf AS, Friend DR, MASRx and COSRx Sustained-Release Technology in Rathbone MJ, Hadgraft J, Robert MS, Modified Release Drug Delivery Technology, Marcell Dekker Inc., New York, 2003.
- 20. Vidyadhara S, Rao PR., Prasad JA. Indian J.Pharm Sci, 2004; 66: 188-192.
- 21. Reddy KR., Mutalik S, Reddy S. AAPS Pharm. Sci. Tech., 2003; 4: 1-9.
- 22. Mohammed AD., James LF, Michael HR., John EH., RajabiSiahboomi AR. Release of propranolol hydrochloride from matrix tablets containing sodium carboxy methylcellulose and Hydroxypropyl methyl cellulose. Phar. Dev. Tech.,1999; 4: 313-324.
- 23. Lee BJ, Ryu SG, Cui JH, Drug Dev. Ind. Pharm., 1999; 25: 493-501.
- 24. Gwen MJ, Joseph RR, In Banker GS and Rhodes CT, Eds., Modern Pharmaceutics, 3rd Edn, Vol. 72, Marcel Dekker Inc. New York, 1996: 575.
- 25. Salsa T, Veiga F And Pina M.E, Drug Develop. Ind. Pharm., 1997; 23: 931.
- 26. Jantzen GM, Robinson JR, Sustained and controlledrelease drug delivery systems, in Banker GS, Rhodes CT (Eds.) Modern Pharmaceutics, Third Edition, Revised and Expanded, Drugs and the Pharmaceutical Sciences, vol 72, Marcell Dekker, Inc. New York, 1995: 575-609. x
- 27. H Bechgaard, G H Nielson. Controlled release multiple units and single unit dosage; Drug Dev. & Ind. Pharm., 1978; 4(1): 53-67.
- 28. Sayed I. Abdel-Rahman, Gamal MM, El-Badry M, Preparation and comparative evaluation of sustained release metoclopramide hydrochloride matrix tablets, Saudi Pharmaceutical Journal ,2009; 17: 283-288.
- 29. Chandran S, Laila FA and Mantha N, Design and evaluation of Ethyl Cellulose Based Matrix Tablets of Ibuprofen with pH Modulated Release Kinetics, Indian Journal of Pharmaceutical Sciences, September-October 2008.
- 30. Gothi GD, Parinh BN, Patel TD, Prajapati ST, Patel DM, Patel CN, Journal of Global Pharma Technology, 2010; 2(2): 69-74.
- 31. Basak SC, Reddy JBM, and Lucas Mani KP.Indian Journal of Pharmaceutical Sciences, September-October 2006.
- 32. Varshosaz J, Tavakoli N and Kheirolahi. AAPS PharmSciTech, 2006; 7(1).
- 33. Raghvengra Rao NG, Gandhi S, and Patel T. International Journal of Pharmacy and Pharmaceutical Sciences, 2009; 1(1).
- 34. Shivhare UD, Adhao ND, Dr. Bhusari KP, Dr. Mathur VB and Ambulkar UD. International Journal of Pharmacy and Pharmaceutical Sciences, 2009; 1(2).

- 35. Vyas SP, Khar RK. Controlled Drug Delivery: Concepts and Advances. Ist ed. vallabh prakashan, 2002:156-189.
- 36. Brahmankar HA, Jaiswal SB, Biopharmaceutics and Pharmacokinetics A Treatise, Vallabh Prakashan, 2000, 348-357 and 337.
- 37. Wani MS, Controlled Release System- A Review, 2008, 6 (1), www.pharmainfo.net/review
- 38. Shargel L, Yu ABC. Modified release drug products. In: Applied Biopharmaceutics and Pharmacokinetics. 4th ed. McGraw Hill. 1999; 169-171 ICH Guideline on Stability study; 2005
- 39. Park S, chun M, Choi H, preparation of an extended release matrix tablet using chitosan/ carbopol interpolymer complex, International journal of pharmaceutics, Volume 347, 1 (2), Page no 39-44.
- 40. Ford JL, Michael R, Horgan J, formulation of sustained release promethazine hydrochloride tablets using hydroxypropryl methylcellulose matrices, international journal of ppharmaceutics, volume 24, 2 (3), 327-338.
- 41. Makhija S, Vavia P, once daily sustained release tablte of vanlafaxine, a novel antidepressant, European journal of pharamceutics and Biopharmaceutics, volume 54,1 page no 9-15.
- 42. Patil V. Pawar S, formulation and evaluation of floating matrix tablet of locally acting h2-antagonist, international journal of pharmacy and technology, 2010, Vol 2, 3 page no 528-540.
- 43. S Sathiyaraj, Ramya D Devi, Vedha B.N Hari, Lornoxicam gastro retentive floating matrix tablets: Design and *in vitro* evaluation, J adv pharma sci tech, 2011, 2 page no 156-162.
- 44. Reynolds JEF, Eds., In; Martindale; The Extra Pharmacopoeia, 29th Edn., The Royal Pharmaceutical Society of Great Britain, London, 1993, 295.
- 45. Yadav S K, Kavita K, Tamizhamani T; Formulation and evaluation of floating tablets of ranitidine hydrochloride using natural and synthetic polymers. *Int J pharm Tech Res.* 2010; 2(2): 1513-1519.
- 46. Klausner EA, Lavy E, Friedman M, Hoffman A.Expandable gastroretentive dosage form. J. of Contr. Release.2003;90:143-162.
- 47. Jiménez-Martínez I, Quirino-Barreda T, Villafuerte-Robles L. Sustained delivery of Captopril from floating matrix tablets. Int J Pharm 2008;362:37-43.
- 48. Melander, A, Stenberg P, Liedholm H, Schersten B, Wahlin-Boll E. Food induced reduction in bioavailability of atenolol. European Journal of Clinical Pharmacology. 1979; 16: 327–330.
- 49. Patel A Modasiya M Shah D, V Patel, "Development and *in vitro* floating behavior of verapamil HCL intragastric floating tablets". *AAPS PharmSciTech*, Vol 10, 310-315, 2009.
- 50. Ansel HC and Loyyd VA. Pharmaceutical dosage forms and Drug Delivery System. Lippincott's Williams and Wilking, Hong Kong. 1999; 8: 275-280.
- 51. Viriden A, Wittgren B, Larsson A. Investigation of critical polymer properties for polymer release and swelling of HPMC matrix tablets. Eur J Pharma Sci, 2009; 36: 297-309.