

A REVIEW ON EFFECTS OF SOME HEAVY METALS ON PLANTS AND HUMAN HEALTH

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Abstract—Wastes are the major source of soil pollution, originating from mining, chemical, metal processing industries and other allied industries. These wastes include many chemicals like heavy metals, phenolic, organic, non-metals etc. Heavy metals are the intrinsic components of the environment, with essential and non-essential both types. Heavy metal accumulation in soil is major concern in agricultural production due to its adverse effect on food safety, marketability and crop growth and also due to phyto-toxic effects and environment health of soil organism. The toxic effect of heavy metals, even though they do not have any biological role, remain present in some or the other form harmful for human body and its person functioning. Metal toxicity depends upon the absorbed dose, the route of exposure and duration of exposure, acute or chronic. This can lead to various disorder and can also result in excessive damage due to oxidative stress induced by free radical formation. This review paper details about some heavy metals and their toxicity mechanism along with their health effect. Heavy metals exhibit toxic effects towards soil by affecting key microbial process and decrease the number and activity of soil micro-organism. Even low concentration of heavy metal in plants and subsequent accumulation along the food chain is potential threat to human health

Index Terms—Heavy metals, Effects on Human health, Toxicity, Effects on plants

I. INTRODUCTION

Metallic elements are intrinsic components of the environments. Heavy metals or toxic metals when present in excess of required concentration or is unwanted which were found naturally on the earth, and become concentrated as a result of human caused activities, enter in plants, animal and human tissues via inhalation, diet and manual handling can bind to, and interfere with the functioning of vital cellular components. Heavy metals are significant environment pollutants and their toxicity is a problem of increasing significance for ecological, evolutionary, nutritional and environmental reason. They are group of metals and metalloids with atomic density greater than $4g/cm^3$ or those which are 5 times or more, greater than water, including copper, manganese, lead, cadmium, nickel, cobalt, iron, zinc, chromium, silver. Uptake of heavy metals by plants and subsequent accumulation along the food chain is a potential threat to human health. Adverse health effects of heavy metals have been known for a long time. Exposure to heavy metals continues and is even increasing in some areas. For example mercury is still used in gold mining in many part of Latin America. Arsenic is still common in wood preservatives, and tetraethyl lead remain a common additive to petrol, although this use has decreased dramatically in the developed countries. This paper briefly describes the nature and properties of heavy metal and its effect on the plants and human health.

II. SOURCE AND EMISSION

Toxic metals, to a large extent, are dispersed in the environment through industrial effluents, organic wastes, refuse burning and transport and power generation. They can be carried to places many miles away from the sources by wind, depending upon weather. They are found in gaseous form or as particulates.

Table 1 List of Metals

Metals	Manufacturing Industry
Arsenic	Paints and Textile
cadmium-	Electronics and pigments
chromium	Metal plating
copper	Plating
lead	Plating
zinc	Galvanizing, plating iron

III. EFFECTS ON PLANTS

The heavy metals available for plant uptake are those present as soluble components in the soil solution or those soluble by root exudates. Plants require certain heavy metals for their growth and upkeep, but excessive amounts of these metals can become toxic to plants. The ability of plants to accumulate essential metals equally enables them to acquire other non-essential metals. They adversely affect the plant both directly and indirectly. Some of the direct toxic effects caused by high metal concentration include inhibition of cytoplasmic enzymes and damage to cell structure due to oxidative stress. Indirect toxic effect includes replacement of essential nutrients at cation exchange sites of plants. The negative influence of heavy metals on the growth and activities of soil microorganisms also indirectly affect the growth of plants. Reduction in the number of beneficial soil microorganisms due to high metal concentration may lead to decrease in organic matter

decomposition, leading to a less fertile soil. Enzyme activities are very much useful for plant metabolism, hampered due to heavy metal interference with activities of soil microorganisms¹². These toxic effects (both direct and indirect) lead to a decrease in plant growth which finally results in the death of plant¹³. Heavy metal accumulation in plants depends upon plant species and the efficiency of different plants for absorbing metals is evaluated by either plant uptake or soil to plant transfer factors of the metals¹⁴. Elevated lead in soils may decrease soil productivity, and a very low lead concentration may inhibit some vital plant processes, such as photosynthesis, mitosis and water absorption with toxic symptoms of dark green leaves, wilting of older leaves, stunted foliage and brown short.

Effects of copper on plants

Copper is an essential metal for normal plant growth and development, although it is also potentially toxic. In M.P, it was found that copper dust had adverse effect on various photosynthesis pigmentation secretions in many trees species leaves¹⁶. Copper is also an essential component of various proteins like plastocyanin of photosynthetic system and cytochrome oxidase of respiratory electron transport chain¹⁷. Mining activities generate a large amount of waste rocks and tailings, which get deposited at the surface. Excess of Copper in soil plays a cytotoxic role, induces stress and causes injury to plants. This leads to plant growth retardation and leaf chlorosis¹⁸.

Effect of zinc on plants

Zinc (Zn) is an essential micronutrient that affects several metabolic processes of plants¹⁹ and has a long biological half-life. The phytotoxicity of Zn and Cd is indicated by decrease in growth and development, metabolism and an induction of oxidative damage in various plant species such as *Phaseolus vulgaris*²⁰. Excess Zn can also give rise to manganese (Mn) and copper (Cu) deficiencies in plant shoots. Such deficiencies have been ascribed to a hindered transfer of these micronutrients from root to shoot. This hindrance is based on the fact that the Fe and Mn concentrations in plants grown in Zn rich media are greater in the root than in the shoot. Another typical effect of Zn toxicity is the appearance of a purplish red color in leaves, which is ascribed to phosphorus (P) deficiency²¹. Zinc in excess reduces the germination, chlorophyll, carotenoid, sugar, amino acid and growth of cluster beans (*Cyamopsis tetragonoloba*²² whereas, in pea (*Pisum sativum*) reduces chlorophyll, photosynthesis and plant growth²³. In rye grass (*Lolium perenne*) it reduces the growth, nutrient content and photosynthetic energy conversion²⁴.

Effects of cadmium on plants

The permissible limit of cadmium (Cd) in agricultural soil is 100 mg/kg soil²⁵. Plants grown in soil containing high levels of Cd show visible symptoms of injury reflected in terms of chlorosis, growth inhibition, browning of root tips and finally death. Metal toxicity can affect the plasma membrane permeability, causing a reduction in water content; in particular, Cd has been reported to interact with the water balance²⁷. Cadmium treatments have been shown to reduce ATPase activity of the plasma membrane fraction of wheat and sunflower roots²⁸. Cadmium produces alterations in the functionality of membranes by inducing lipid per-oxidation²⁹ and disturbances in chloroplast metabolism by inhibiting chlorophyll biosynthesis and reducing the activity of enzymes involved in CO₂ fixation³⁰. In wheat (*Triticum sp.*) excessive of cadmium reduces the seed germination; decrease in plant nutrient content; reduced shoot and root length³¹. Whereas in garlic (*Allium sativum*) Cd accumulation reduced shoot growth³² Lastlyn in Maize (*Zea mays*) it reduces shoot growth and inhibition of root growth³⁴

Effects of nickel on plants

Nickel (Ni) is an essential nutrient for plants. However, the amount of Ni required for normal growth of plants is very low. Hence, with the level of Ni pollution in the environment increasing, it is essential to understand the functional roles and toxic effects of Ni in plants. Ni⁺² concentration in polluted soil may range from 20 to 30 fold (200–26,000 mg/kg) higher than the overall range (10–1,000 mg/kg) found in natural soil³⁴. However, Ni⁺² concentration is increasing in certain areas by human activities such as mining works, emission of smelters, burning of coal and oil, sewage, phosphate fertilizers and pesticides³⁵ Plants grown in high Ni⁺² containing soil showed impairment of nutrient balance and resulted in disorder of cell membrane functions. Finally, in Rice (*Oryza sativa*) Inhibition of root growth³⁶.

Effects of iron on plant

Iron is mainly involved in the process of plant photosynthesis. The micronutrient's availability to plant roots depends on the pH level of the soil with iron more readily available in soil with a low pH. Iron is a major constituent of the cell redox systems such as heme proteins including cytochromes, catalase, peroxidase and leghemoglobin and iron sulfur proteins including ferredoxin, aconitase and superoxide dismutase (SOD) [38]. Iron as an essential element for all plants and has many important biological roles in the processes as diverse as photosynthesis, chloroplast development and chlorophyll biosynthesis. Iron toxicity in tobacco, canola, soybean and *Hydrilla verticillata* are accompanied with reduction of plant photosynthesis and yield and the increase in oxidative stress and ascorbate peroxidase activity³⁹

Effects of manganese on plants

Manganese (Mn) is an essential plant mineral nutrient, playing a key role in several physiological processes, particularly photosynthesis. Manganese deficiency is a widespread problem, most often occurring in sandy soils, organic soils with a pH above 6 and heavily weathered, tropical soils⁴⁰. Necrotic brown spotting on leaves, petioles and stems is a common symptom of Mn toxicity⁴¹. This spotting starts on the lower leaves and progresses with time toward the upper leaves. Mn in pea (*Pisumsativum*) reduces chlorophylls a and b content; reduction in relative growth rate; reduced photosynthetic O₂ evolution activity and photo system II activity. However, in tomato (*Lycopersicon esculentum*) Mn slower plant growth; decrease in chlorophyll concentration⁴².

Effects of arsenic on plants

In tomato (*Lycopersicon esculentum*) arsenic reduces fruit yield, decreases the leaf fresh weight [50]. Whereas, in canola (*Brassica napus*) arsenic causes stunted growth, chlorosis and wilting⁴³. Further, arsenic in rice (*Oryzasativa*) reduces seed germination, decrease in seedling height, and reduces leaf area and dry matter production⁴⁴.

IV. EFFECTS ON HUMAN HEALTH

Utilization of food crops contaminated with heavy metals is a major food chain route for human exposure. Heavy metals become toxic when they are not metabolized by the body and accumulate in the soft tissues⁴⁴. Chronic level ingestion of toxic metals has undesirable impacts on humans and the associated harmful impacts become perceptible only after several years of exposure⁴⁵.

Cadmium (Cd)

Cadmium is a heavy metal toxicant with a specific gravity 8.65 times greater than water. The target organs for Cd toxicity have been identified as liver, placenta, kidneys, lungs, brain and bones. Depending on the severity of exposure, the symptoms of effects include nausea, vomiting, abdominal cramps, dyspnea and muscular weakness. Severe exposure may result in pulmonary edema and death. Pulmonary effects (emphysema, bronchiolitis and alveolitis) and renal effects may occur following subchronic inhalation exposure to cadmium and its compounds. The Itai-itai disease in Japan brought the dangers of environmental Cd to world attention.⁴⁷

Zinc

The clinical signs of zinc toxicosis have been reported as vomiting, diarrhea, bloody urine, icterus (yellow mucus membrane), liver failure, kidney failure and anemia.

Copper (Cu)

Copper can result in a number of adverse health effects. Excessive human intake of Cu may lead to severe mucosal irritation and corrosion, widespread capillary damage, hepatic and renal damage and central nervous system irritation followed by depression. Severe gastrointestinal irritation and possible necrotic changes in the liver and kidney can also occur⁴⁸.

Nickel

The effects of Ni exposure vary from skin irritation to damage to the lungs, nervous system, and mucous membranes⁴⁹.

Lead (Pb)

Lead poisoning also causes inhibition of the synthesis of haemoglobin; cardiovascular system and acute and chronic damage to the central nervous system (CNS) and peripheral nervous system (PNS). Other chronic effects include anemia, fatigue, gastrointestinal problems and anoxia. Lead can cause difficulties in pregnancy, high blood pressure, muscle and joint pain⁵⁰. Other effects include damage to the gastrointestinal tract (GIT) and urinary tract resulting in bloody urine, neurological disorder and can cause severe and permanent brain damage. Lead affects children; particularly in the 2-3 years old range by leading to the poor development of the grey matter of the brain, thereby resulting in poor intelligence quotient (IQ).

Chromium

Chromium (Cr)-Cr (VI) is toxic to plants and animals, being a strong oxidizing agent, corrosive, soluble in alkaline and mildly acidic water, toxic and potential carcinogen. The toxicity of Cr (VI) derives from its ability to diffuse through cell membranes and oxidize biological molecules⁵⁶.

Mercury

Mercury causes spontaneous abortion, congenital malformation and gastrointestinal disorders (like corrosive esophagitis and hematochezia). Poisoning by its organic forms, which include monomethyl and dimethylmercury presents with erethism (an abnormal irritation or sensitivity of an organ or body part to stimulation), acrodynia (Pink disease, which is characterized by rash and desquamation of the hands and feet), gingivitis, stomatitis, neurological disorders, total damage to the brain⁵⁷.

Arsenic

It is possibly carcinogenic in compounds of all its oxidation states and high-level exposure can cause death. Arsenic toxicity also presents a disorder, which is similar to, and often confused with Guillain-Barre syndrome, an anti-immune disorder that occurs when the body's immune system mistakenly attacks part of the PNS, resulting in nerve inflammation that causes muscle weakness⁵⁸.

V. CONCLUSION

In this review we reviewed the effects of some heavy metals, i.e. arsenic, lead, mercury, cadmium, on the environment and living organisms, mainly human beings. Effective legislation, guidelines and detection of the areas where there are higher levels of heavy metals are necessary. Failure to control the exposure will result in severe complications in the future because of the adverse effects imposed by heavy metals. Occupational exposure to heavy metals can be decreased by engineering solutions. Monitoring the exposure and probable intervention for reducing additional exposure to heavy metals in the environment and in humans can become a momentous step towards prevention. National as well as international co-operation is vital for framing appropriate tactics to prevent heavy metal toxicity. Plants grow on heavy metal polluted soils resultant in reduction in growth due to changes in their physiological and biochemical activities especially true when the heavy metal involved does not play any beneficial role towards the growth and development of plants. Thus, it is evident from the several research findings that judicious use and presence of heavy metals having toxic effects on plants, animals and many living organisms after certain limits. There are two aspects on the interaction of plants and heavy metals, one hand, heavy metals show negative effects on plants and other hand, plants have their own resistance mechanisms against toxic effects and for detoxifying heavy metal pollution. Our review showed that both growth and photosynthetic pigments are affected by the presence of heavy metals. Plants employ different mechanisms in their mediation of heavy metal polluted soils and phyto extraction is the most common method of phyto remediation used for treatment of heavy metal polluted soils which ensures the complete removal of the pollutant.

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