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# A SHORT REVIEW ON FABRICATION OF ZnO BASED AMMONIA GAS SENSOR

## <sup>1</sup>Kirti B S, <sup>2</sup>Neha T, <sup>3</sup>Nithyashree S, <sup>4</sup>Pavitra C

<sup>1234</sup>Student of Electronics and Communication Department at Alva's Institute of Engineering and Technology, Mijar
<sup>1</sup>Department of Electronics and Communication,
<sup>1</sup>Alva's Institute of Engineering and Technology, Mangalore, India

Abstract : The gas sensor which is constructed using metal oxide semiconductors-based have been widely investigated because of its high detecting capacity, cost-effectively, long haul dependability, and basic creation. In any case, their usage at low working temperature is as yet testing. Subsequently, decrease in power utilization is profoundly fundamental for long haul used in gas detectors. The highly qualified competitors, ZnO nanostructures-based gas sensors require continuous identification of unstable and poisonous gas. So, various endeavors have been made to further develop the detecting reaction at decreased working temperature with the help of different techniques. A few procedures connected with the amalgamation of the nanostructures of ZnO and their effective presentation in detecting are surveyed in this report. The review fundamentally centers around various method for further developing the detecting properties, for example, metal nanoparticles fictionalization, metal doping, consideration of carbonaceous nanomaterials, utilizing nanocomposites of various metal oxides, ultra violet enactment, and post-therapy technique for maximum energy illumination on ZnO nanostructures, with their conceivable detecting instruments. Hence reveal insight into future recommendations of zinc oxide-based gas sensors showing maximum responsiveness even at low working temperature.

### Keywords: Zincoxide, gas sensor, thin-film, nanoparticles.

### I. INTRODUCTION

In today's world, the advancement of industrialization and urbanization has brought about a gigantic expansion in air contamination. More specifically ammonia is a bleached, profoundly aggravating gas with a pungent, choking out scent. This ammonia gas causes several effects on environment and. So, it is fundamental to foster the gas sensor for continuous identification and ammonia outflow control. For a really long time, metal oxide semiconductors (MOSs), including ZnO, CuO, SnO2, TiO2, Fe2O3, many more have ruled the field of poisonous gas discovery nanomaterials, due to their minimal expense, non-harmfulness and accessibility. Among these materials, ZnO have been seen a massive investigation because of its larger band gap, enormous exaction restricting energy, versatility of conduction electrons is high, change in physical and substance strength. As of late, ZnO-based gas sensors with various nanostructures have been generally announced although further developed arrangement techniques and adjusted morphologies have essentially upgraded gas detecting execution of ZnO-based sensors, the most detailed sensors actually worked at high temperatures, which restricted their useful appropriateness. Inferable from ZnO-based gas sensors by and large should be warmed the detecting layer to a higher temperature to accomplish the detecting execution, prompting a powerful utilization and changes in the nanomaterial microstructure. The impediment of hightemperature detecting is likewise reflected in the recognition of combustible and touchy gases, with a danger of blast. Furthermore, decline of oxygen particles adsorbed on the detecting materials outer layer that would bring about the lessening of the reaction worth of sensors when the working temperature is excessively high. In light of the wellbeing and dependability, sensors that can work at lower temperatures are turning out to be increasingly alluring. Accomplishing this prerequisite can lessen energy utilization and cost, yet in addition work on the exactness of the recognition results. In this way, an orderly comprehension of procedures of planning less temperature ZnO-based ammonia sensors and relating detecting components is of directing importance for the plan of novel low-temperature ZnO-based ammonia sensors.

#### **II. REVIEW**

Jing Xu et al. [1] focused on manufacturing the sophisticated nanoarchitectures, particularly in three dimensional (3D) hierarchical systems formed from 1D and 2D nanoscale building components. Various kinds of methods have been used for deposition of ZnO thin films such as metal organic chemical vapour deposition, spin coating, thermal oxidation, and magnetron sputtering. Among the all, solution-phase chemical technique is considered as one of the primary method as it's cheaper

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compared to others, environmental friendliness, mild environment, also the ability for huge-scale manufacturing. To deposite films which aligned Chemical bath deposition (CBD) method are used, these films are aligned on the ZnO nanoclusters on a quartz substrate. In addition to this, UV-activated gas sensor prototype is established also tested. The gas detector which is critical for viable application can operate at 23 degree Celsius because of the photo-enhanced gas sensing properties.

Chi-Jung Chang et al. [2] focused on improving high performance of gas sensing property to have low power consumption sensor. This can be achieved by operating the sensors at low temperature. Ce doped ZnO nanorods array sheets were used for the construction of  $NO_2$  sensors with low operating temperatures. Several properties like electrode structure, surface roughness, and amount of Ce on optical and gas sensing capabilities are tested. The journal gives a brief knowledge of how we can enhance sensor response towards  $NO_2$  gas.

Bai Shouli et al. [3] focused on usage of ZnO nanorods in NO<sub>2</sub> sensors. The study says that at 90°C, 1D ZnO nanorods which are having pencil-like shape and a high aspect ratio can be produced at 90°C. X-Ray power Diffraction (XRD), Scanning Electron Microscope (SEM), Transmission Electron Microscope (TEM), and Field Emission Scanning Electron Microscope (FESEM) are the methods used for testing the development of ZnO nanoparticles. This journal says, at the same gas concentration, calculated by the sensor which is of pencil-like ZnO calculated at 400 degree Celsius for one hour had a heavy response from 20 up to 40 ppm NO<sub>2</sub> and the highest selectivity of 10 to CO and 30 to CH<sub>4</sub>. As a result, 1D ZnO nanorods are the excellent semiconductor sensing material.

Hyun-Uk Lee et al. [4] shows how the Nanobarbed ZnO fibres (NBFs) are created by electrospinning and chemical bath deposition methods of 1–D ZnO nanorods (NRs) on ZnO nanofibers (NFs) to form epitaxial growth. The two models such as surface depletion model and modified grain boundary model, are the junction locations for NFs and NRs, these were used to know the variations in resistance of the NBFs when gas is present or absent. Along with the pictorial representation, The Nanofibers and Nanorods were characterized using XRD, SEM, TEM and FESEM. The ZnO NBFs proved to respond quickly to NO<sub>2</sub> gas.

Liangyuan Chen et al. [5] focused on the procedure used to obtain 1D ZnO nanorods. The 1D ZnO nanostructures have been synthesized using a variety of approaches, including thermal evaporation, chemical vapour deposition, sol–gel, and solvothermal methods. Out of these techniques, the hydrothermal process is widely acknowledged as the most promising method for producing highly crystalline materials with well-defined shape under relatively mild circumstances. The process is projected to become more cost-effective when the hydrothermal temperature is reduced and product quality is stabilized.

Miaolinget al [6] proposed work on making distinctive nanosensors. The surface morphology and critical produced methodologies for ZnO nanomaterials, with advantages and disadvantages of each procedure is presented. Then based on the new progression in functional ZnO nanosensors have given the investigation information and necessities improvement of ZnO nanosensor in every characterization. Concluding, the hardships and future scope of ZnO nanosensors.

P Rai et al. [7] focused on the dry gas sensor uses which are flower like ZnO polymeric. The hydrothermal approach is used for synthesis off lower-like ZnO micro structural, Gas detectors is put together by sturdy film automation and checked firstly for nitrogen oxide gas at various testing temperatures in addition with concentrations. Response elevated with increasing of gas concentration also temperature. The result obtained for response of 10 ppm of nitrogen oxide gas at 23<sup>o</sup>C. Response for nitrogen gas was greater compared to carbon monoxide, ethanol and acetaldehyde at low operating temperatures.

Kanget al. [8] reported the various gas detecting tools of ZnO based nanomaterials. The connection between the four distinct dimensional structures of ZnO and its gas sensitive performance is described in this paper. The interaction of ZnO with other materials is thoroughly described. The analysis of the effects on visible light and ultraviolet light on the gas sensitive properties of materials was discussed.

Mani et al [9] reported the growth mechanism of Ni doped nanostructure zinc oxide thin films deposited on glass using the spray pyrolysis method. The variation in the dimensions of crystallite and morphology is observed, which showed the effect on Ni doping. Thin films are very choosytowardsNH<sub>3</sub>when compared to that of  $C_2H_5OH$  and  $C_3H_6O$ . The upper detection range was widened to 1000 ppm by Ni doping.

Kwak et al [10] proposed work on the different sensing techniques used to detect the ammonia gas. The basic workingprinciples of NH<sub>3</sub> detection methods are electronics, electrochemistry, tunable diode laser spectroscopy, surface acoustic wave, and also field effect transistors are discussed. Furthermore, the creation of advanced NH<sub>3</sub> gas sensors was completely observed and summarized. The development of ammonia gas sensors whose performance is high is discussed.

Ozturk et al. [11] focused on creating Zinc oxide (ZnO) nanorods by utilizing aqueous technique and resistive sort; NO2 detecting effects of the nanorods were examined based on temperature, NO2 concentration, cathode position and the component of the nanorods. To deposited ZnO thin layer, zinc acetic acid derivation arrangement was deposited on a glass substrate by utilizing spin coater.

Liu et al. [12] proposed work on adjustment of ZnO nanorod clusters in vertical direction and are prepared by an aqueous technique with zinc acetic acid derivation and hexamethylenetetramine, and utilized for nitrogen dioxide gas detecting. The nitrogen dioxide sensor in view of nanorod exhibits very high sensitivity and low working temperature. Improved sensitivity is put down to maximum perspective proportion of the nanorod structure where the ZnO film was prepared by ultrasonic splash pyrolysis . The reaction of the ZnO nanorod sensor is straightly corresponding to the nitrogen dioxide concentration, and the sensitivity of the sensor increments with the length of the ZnO nanorod.

Jun Zhang et al [13] focused on the most significant aspects of fundamental research, sensing procedures, and application of nano structured materials for room temperature conductometric sensor systems. In order to get structure property correlations, special attention is given to the relationship between nanostructure and sensor qualities.

Hackenberg's et al. [14] proposed work confirmed that the semiconductor gasoline sensor's sensing properties of the usage of  $SnO_2$  can be stepped forward through controlling essential factors which have an effect on the functions of its receptor and transducer. The transducer function is associated with the microstructure of the factors, such as  $SnO_2$  grain length (D) and the floor space fee layer intensity (L). The sensitivity is notably advanced whilst D is made comparable to or less than 2L, both by way of manipulate grain length for pure  $SnO_2$  factors or by the Debye length handling for impurity doped elements. On the other side, the function of the receptor is substantially changed through the creation of overseas receptors at the  $SnO_2$  surface. In particularly the Pd and Ag promoters, its oxides formed in air interact with the surface of  $SnO_2$  to produce the space charge layer with electron deficiency, and this puts up on tons to selling the gas sensitivity.

Chougule's et al. [15] focused on the deposition of thin layers of nanocrystalline zinc oxide (ZnO) on the glass substrate with the aid of a spin coating method. ZnO thin films had been characterized for the structural properties and morphological properties by the method of X-ray diffraction (XRD), the scanning electron microscopy (SEM) and the atomic force microscopy (AFM). The ZnO thin films are oriented alongside 101 planes with the hexagonal crystal shape. The films have been applied on nitrogen dioxide sensors. The dependence of the response of NO<sub>2</sub> on the operating temperature, NO<sub>2</sub> awareness changed into investigated. The ZnO films confirmed selectivity for NO<sub>2</sub> over H<sub>2</sub>S in comparison with NH<sub>3</sub>. The gasoline response of 37.2% was completed with 78 percentage of balance for ZnO layers upon publicity of a 100ppm NO<sub>2</sub> at operating temperature 200 degree Celsius.

Priya et al. [16] focused on the identification of ammonia gas. Metal-oxide has materialized as a reassuring decision in the gas sensor industry. This survey journal centres on the alkali detecting standard of the metal oxides. It additionally incorporates different methodologies embraced for expanding gas awareness in metal-oxide sensors. Expanding a responsiveness of  $NH_3$  incorporates size impacts and doping by metal oxides. Various boundaries which influence the characteristics are examined. Characterization of frequently worked metal oxides with qualities and limits in alkali gas detecting uses were evaluated.

Bharath et al. [17] focused on how Zinc oxide nanoparticles have been a functioning exploration region due to their intriguing physical, synthetic effects and practicality in optoelectronics, substance detecting, bio sensing, and photo catalysis, optical elements. Optical properties can be improved by adding impurities. Here adding impurities to ZnO nanoparticles utilizing particular dopants by several methods and behavioural variation are discussed. It shows that doping adjusts the attributes of ZnO nanoparticles and better its appropriateness for different fields.

Chang et al. [18] focused on an idea about how doping of materials makes a difference in sensing. Briefed how the performance of the gas sensor can be improved so that it can operate in low temperature and consume minimal amount of power. The usage of ZnO nanorods films on Al2O3 substrate with platinum to which the Ce impurities is added. Using Ce as dopant not only response time can be improved also operating temperature required can be improved. An overall picture about the effect of electrode structure, surface texture, and gas sensing performance is examined.

Zhnag et al. [19] worked on the solidity and functionality of nanoscale catalyst where these are two different aspects for improving quality of a metal-oxide based gas sensor. Here MOF is developed to fabricate the Ag nano catalyst. Ag-ZnO hollow NCs enhance the sensing and reduce the operating temperature when compared to that of the pure ZnO nano-particle. The result obtained in this is 3.3 times greater than the pure ZnO at 275°C.

Sergiu et al [20] focused on how NO<sub>2</sub> gas sensor was created by progressive ionic layer adsorption and response (SILAR) procedure and fast photo thermal handling (RPP) of the Sn-doped ZnO film. The outcome is 1.5 ppm NO<sub>2</sub> in air and downshift the working temperature. The impact of variety of Sn fixation in the compound shower and the RPP temperature on NO<sub>2</sub> responsiveness is discussed. Higher responsiveness was gotten at 5-10 in the arrangement of particles and RPP temperature of 550-650 °C. Expanding the Sn focus in doped ZnO tests more than10 %. It looks encouraging to utilize the reasonable tin-doped zinc oxide by SILAR strategy and RPP in brilliant gas detecting techniques.

Mohanta et al [21] focused on utilization of metal natural synthetic fume testimony to concentrate on the primary and optical properties of ZnO nanorods created on c-sapphire substrates. The presence of huge band-edge emanation and higher-request LO phonon dissipating checks the optical nature of ZnO nanorods. The wurtzite character of ZnO is affirmed by the presence of a noticeable E2, high optical phonon mode in apparent Raman spectra at 438 cm. Higher-request LO phonon modes and a recurrence up shift are found in bright thunderous Raman dissipating investigates such nanorods at higher development temperatures.

Dey et al [22] presented work on examination of semiconductor metal oxide (SMO) gas sensors, including a correlation of SMO gas sensors to different gas sensors, especially for NH3 gas detecting. The impacts of a few boundaries on SMO gas sensor execution such as awareness, selectivity, and solidness were investigated. The dopant or contamination actuated changes in SMO materials utilized for gas detecting is additionally discussed. Dopants adjust the microstructure and morphology of SMOs, just as their enactment energy, electrical construction, and band hole, upgrading their capacities for gas detecting Applications.

Bhati et al [23] proposed work shows that RF sputtered Ni-doped ZnO nanostructures can detect extremely low hydrogen gas concentrations of 1 ppm at a moderate working temperature of 75°C. The behaviour of Ni-doped ZnO nanostructures in terms of structural, morphological, electrical, and hydrogen sensing is greatly influenced by doping concentration. The main purpose of

Ni-doping is to increase the number of active sites for chemisorbed oxygen on the sensor's surface and hence improve the sensing response.

Iqbal et al [24] focused on wet chemical approach to produce 1D Cobalt doped ZnO nanostructures. With Cobalt doping, ZnO retains its wurtzite structure, but lattice constants are somewhat reduced, according to X-ray diffraction studies. Co doping changes the dimensions of nano structures that are from nanowires to nanorods, according to SEM data. HRTEM studies indicated that Cobalt dopant cannot affect ZnO preferred growth orientation, despite having a significant impact on grain size and shape. It also discovered that Co doping causes a lot of flaws in the ZnO nanostructure. At room temperature, the doped nanorods reveal ferromagnetic behaviour, which is due to plenty of defects and oxygen vacancies.

Maciel et al [25] focused on structural and morphological studies in pure and Ce-doped tin dioxide nanoparticles with great stability against particle development were conducted in samples acquired using the polymeric precursor approach and prepared at various annealing temperatures. To limit particle size and prevent particle development, a Ce-rich surface layer was utilised. X-ray diffraction, in conjunction with Rietveld refinements, X-ray photoelectron spectroscopy, and transmission electron microscopy, were used to analyze the evolution of crystallite size, micro strain, and shape of nanoparticles as annealing temperatures increased.

Kumar et al [26] presented work on RF superimposed DC faltering being utilized to store as Al doped ZnO (AZO) film. The various RF or RF+DC proportions with a similar power was mainly focused. Grain requirements incorporate optical, underlying nanoscale and electrical potential. The review demonstrates that the films of AZO at 0.75 proportions, use to have a low left over pressure, high electron portability, and high electron versatility. When compared with other AZO, this one has an enormous crystallite size.

Drmosh et al [27] proposed work on H2 recognition under UV light at room temperature, Au nanoparticles were set on a hetero structure nano composite. The unadulterated zinc oxide, rGO or ZnO, and gold/rGO/zinc oxide for H2 detecting with and without UV radiation at different working temperatures were tested. The sensor produced using Au/rGO/ZnO had the most extreme H2 detecting reaction at the surrounding temperature, while different sensors had less reaction at high working temperatures. The elevated surface space of zinc oxide, the creation heterostructures of rGO/ZnO, the contrasting work elements of gold and zinc oxide, and the free electrons by ultraviolet radiation totally added to the further developed H2 detecting reaction.

Katoch et al [28] proposed work on SnO2-ZnO composite nanofibers that were accounted for, with 10 ppm H2 found at 300 degree Celsius. The period of  $SnO_2$ -SnO<sub>2</sub> homo interfaces and the metallization impact at  $SnO_2$ -ZnO hetero-interfaces came about in an improved H2 detecting reaction. A full outline of different works on heterostructures dependent on ZnO nanostructures has been introduced.

Xiao et al [29] proved that one dimensional carbon nanofibers, 1D carbon nanotubes, and 2D grapheme and respective subordinates are largely instances of carbon materials. In view of the absence of practical gatherings and a band hole, unadulterated graphene-based gas sensors are seldom utilized. Subsequently, connecting utilitarian gatherings to the outer layer of graphene is basic for tuning the band hole and providing additional dynamics to target gas particles.

Rajesh Kumar et al [30] focused on overall presentation with a few metal oxides, gas-sensor metals. The fundamental highlights of gas detecting, for example, gas response, response time, recuperation time, procurement cutoff, dependability and reusing, and so forth are additionally discussed. The journal reviews on the usage of nanorod, nanowires, nano-miniature blossoms, quantum specks and nanosheet, and so on which is ZnO nanomaterials. In addition, different factors like NO<sub>2</sub> focus, strengthening temperature, ZnO morphologies and molecule size, relative stickiness, working temperatures influencing NO<sub>2</sub> gas sensor properties are discussed.

## **III. CONCLUSION**

The journal enhances essentials of gas detecting, for example, gas response, response time, recuperation time, determination, obtaining breaking point, dependability, and reusing, among others. Due to the administration of natural NH3 fixations, Znobased ammonia gas sensors could be utilized for ecological checking and control, bringing about more secure and better living conditions.

## REFERENCES

- [1] J. Xu, Y. Yu, X. He, J. Sun, F. Liu, G. Lu, Synthesis of hierarchical ZnO orientation-ordered film by chemical bath deposition and its gas sensing properties. Mater. Lett. 81, 145–147 (2012). doi:10.1016/j.matlet.2012.04.090
- [2] C.-J. Chang, C.-Y. Lin, J.-K. Chen, M.-H. Hsu, Ce-doped ZnO nanorods based low operation temperature NO2 gas sensors. Ceram. Int. 40(7), 10867–10875 (2014)doi:10.1016/j.ceramint. 2014.03.080.
- [3] B. Shouli, L. Xin, L. Dianqing, C. Song, L. Ruixian, C. Aifan, Synthesis of ZnO nanorods and its application in NO2 sensors. Sens. Actuators B: Chem. 153(1), 110–116 (2011). Doi: 10.1016/j.snb.2010.10.010.
- [4] H.-U. Lee, K. Ahn, S.-J. Lee, J.-P. Kim, H.-G. Kim, S.-Y. Jeong, C.-R. Cho, ZnO nanobarbed fibers: fabrication, sensing NO2 gas, and their sensing mechanism. Appl. Phys. Lett. 98(19), 193114 (2011).doi:10.1063/1.3590202.
- [5] S. Bai, L. Chen, S. Chen, R. Luo, D. Li, A. Chen, C.C. Liu, Reverse microemulsion in situ crystallizing growth of ZnO nanorods and application for NO2 sensor. Sens. Actuators B: Chem. 190, 760–767 (2014). doi:10.1016/j.snb.2013.09.032.
- [6] Miaoling Que, Lixiang Chen, Yunfeisun, "Progress in ZnO Nanosensors", College of Electronic and Information Engineering, Suzhou University of Science and Technology, China.2021 Aug 16; 21(16):5502. doi: 10.3390/s21165502.
- [7] P. Rai, S. Raj, K.-J. Ko, K.-K. Park, Y.-T. Yu, Synthesis of flower-like ZnO microstructures for gas sensor applications. Sens. Actuators B: Chem. 178, 107–112 (2013). doi:10.1016/j. snb.2012.12.031.

#### © 2022 JETIR April 2022, Volume 9, Issue 4

[8] Yanli Kang, Feng Yu, Lu Zhang, Wenhao Wang, "Review of ZnO-based nanomaterials in gas sensors", School of Science, Harbin Institute of Engineering and Technology, Shenzhen 518055, PR China.

- [9] Ganesh Kumar Mani, John Bosco Balaguru Rayappan, "Selective detection of ammonia using spray pyrolysis deposited pure and nickel doped ZnO thin films", Applied surface science, India.
- [10] Dongwook Kwak, YuLei, Radenka Maric, "Nanoribbons for Ultrasensitive Ammonia(NH3) Gas Detection", Institute of Material Science, 97 North Eagle ville Road, University of Connecticut, Storrs, CT, 06269, USA.
- [11] S. O " ztu"rk, N. Kılınc, Z.Z. O " ztu"rk, Fabrication of ZnO nanorods for NO2 sensor applications: effect of dimensions and electrode position. J. Alloys Compd. 581, 196–201 (2013). doi:10.1016/j.jallcom.2013.07.063.
- [12] F.-T. Liu, S.-F. Gao, S.-K. Pei, S.-C. Tseng, C.-H.J. Liu, ZnO nanorod gas sensor for NO2 detection. J. Taiwan Inst. Chem. E. 40(5), 528–532 (2009). doi:10.1016/j.jtice.2009.03.008.
- [13] Zhang J, Liu X, Nerie G, Pinna N, 2016. Nanostructured materials for room-temperature gas sensors. Adv. Mater. 28, 795-831.
- [14] N.Yamazoe, New approaches for improving semiconductor gas sensors. Sens. Actuators B 5(1–4), 7–19 (1991). doi:10.1016/0925-4005(91)80213-4.
- [15] M.A. Chougule, S. Sen, V.B. Patil, Fabrication of nanostructured ZnO thin film sensor for NO2 monitoring. Ceram. Int. 38(4), 2685–2692 (2012). doi:10.1016/j.ceramint.2011.11.036.
- [16] Priya gupta, niraj kumar pandey, savita muray, Trends in Chemistry of Mater. doi: 10.1142/9789812833846\_0059.
- [17] C.BharatShubhamS. Mondale H.S, GuptaP.K.Singh, A.K.Dasa Synthesis of Doped Zinc Oxide Nanoparticles. Mechanical Engineering Department, Indian Institute of Technology(ISM), Dhanbad 826004, India.
- [18]Chi-JungChangChang-YiLinJem-KunChenMu-HsiangHsu Ce-doped ZnO nanorods based low operation temperature NO<sub>2</sub> gas sensors, August 2014, <u>Ceramics International</u>:10867-10875doi:<u>10.1016/j.ceramint.2014.03.080</u>
- [19] Jinniu Zhang, HuanLu ,Lizhai Zhang Deying Leng ChunlanWang<sup>e</sup>Metal-organic framework-derived ZnO hollow nanocages functionalized with nanoscale Ag catalysts for enhanced ethanol sensing properties, School of Physics and Information Technology, Shaanxi Normal University, Xi'an 710062, China.
- [20] Sergiu T. Shishiyanu, Teodor S. Shishiyanu, Oleg I. Lupan Sensing characteristics of tin-doped ZnO thin films as NO2 gas sensor.Department of Microelectronics and Semiconductor Devices, Technical University of Moldova, bd. Stefan cel Mare 168, MD-2004 Chisinau, Moldova.
- [21] Mohanta, S.K., Kim, D.C., Cho, H.K., Chua, S.J., Tripathy, S., 2008. Structural and optical properties of ZnO nanorods grown by metal organic chemical vapor deposition. J. Cryst. Growth 310, 3208-3213.
- [22] Dey, A., 2018. Semiconductor metal oxide gas sensors: A review. Mat. Sci. Eng. B 229, 206-217.
- [23] Bhati, V.S., Ranwa, S., Fanetti, M., Valant, M., Kumar, M., 2018a. Efficient hydrogen sensor based on Ni-Doped ZnO nanostructures by RF sputtering. Sensors Actuators B 255, 588-597.
- [24] Iqbal, J., Jan, T., Ronghai, Y., 2013. Effect of Co Doping on morphology, optical and magnetic properties of ZnO 1-D nanostructures. J. Mat. Sci. Mater. El. 24, 4393-4398.
- [25] Maciel, A.P., Lisboa-Filho, P.N., Leite, E.R., Paiva-Santos, C.O., Schrenier. W.H., Maniette, Y., Longo, E., 2003. Microstructural and morphological analysis of pure and Ce-Doped tin dioxide nanoparicles. J. Eur. Ceram. Soc. 23. 707-713.
- [26] Kumar N, Dubey A, Bahrami B, Venkatesan S, Qiao Q, Kumar M, 2018. Origin of high carrier mobility and low residual stress in RF superimposed Dc sputtered Al doped ZnO thin film for next generation flexible devices, Applied Surface Science, Volume 436, p. 477-485.
- [27] Drmosh Q, Hendi A, Hossain M, Yamini Z, Moqbel R, Gondal M, 2019. UV activated gold decorated Zn O heterostructured nanocomposite sensor for efficient room temperature H2 Detection, <u>Sensors and Actuators B: ChemicalVolume 290</u>, 1 July 2019, Pages 666-675.
- [28] Katoch A, Kim JH, Kwon YJ, Kim HW, Kim SS. Bifunctional Sensing Mechanism of SnO2-ZnO Composite Nanofibers for Drastically Enhancing the Sensing Behavior in H2 Gas. ACS Appl Mater Interfaces. 2015 Jun 3;7(21):11351-8.doi: 10.1021/acsami.5b01817. Epub 2015 May 20. PMID: 25950738.
- [29] Xiao z, Kong L, Ruan S, Jing Y, Yao Z, Wang C, 2018. Recent development in nanocarbon materials for gas sensor applications. Sensors and Actuators B: Chemical, Volume 274, 2018, Pages 235-267, ISSN 0925-4005, Doi:10.1016/j.snb.2018.07.040.
- [30] Kumar, R., Al-Dossary, O., Kumar, G. et al. Zinc Oxide Nanostructures for NO2 Gas-Sensor Applications: A Review. Nano-Micro Lett. 7, 97–120 (2015). Doi: 10.1007/s40820-014-0023-3.