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Structural and optical properties of Cu₂ZnSnS₄ thin film Prepared By Chemical Bath Deposition for Solar cell Application

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Abstract: Nanocrystalline Cu2ZnSnS4 (CZTS) thin film was successfully synthesized using a relatively simple and convenient Chemical Bath Deposition (CBD). The thin film was characterized by UV-Vis spectroscopy and X-ray Diffraction technique for confirmation of nanocrystalline material. The absorption spectra show in the visible range and the band gap of CZTS was about 1.68 eV, which was near the optimum value for photovoltaic solar conversion in a single-band-gap device. The polycrystalline CZTS thin films with Kesterite crystal structure have been obtained by XRD. The average crystalline size of CZTS thin film is 27 nm.

Keywords: CZTS, CBD, band gap, XRD etc

I. INTRODUCTION

Kesterite Copper zinc tin sulfide is a quaternary semiconducting compound which has received increasing interest since the late 2000s for applications in photovoltaic cells [1]. It is necessary to fabricate low cost, high conversion efficiency solar cells without material degradation for widely used electricity generation [2]. The band gap of CZTS is 1.5eV, and high optical absorption coefficient beyond 10⁴ cm, due to great deal of interest in research of CZTS thin film solar cells [3].

CZTS thin films have been prepared by various techniques such as physical techniques sequential evaporation technique [4], sputtering [5], Thermal Evaporation method [6] E-Beam Evaporation [7], pulsed laser deposition [8], spray pyrolysis [9], Chemical vapor deposition [10], and chemical techniques such as Dr. Blade method [11], Spin coating [12], Electro-deposition [13], Sol Gel technique [14], Chemical Bath deposition (CBD) [15]. The CBD technique does not require a high quality target and vacuum, which is great advantage of industrial purpose.

In the present study, we report that the CZTS thin film has prepared by CBD method. The structural properties studied by Xray Diffraction method (XRD) Bruker AXS, Germany (D8 Advanced). Optical absorption studies were carried out employing UV-VIS spectrophotometer (Hitachi model U-2900).

II. EXPERIMENTAL

The CZTS thin films have been deposited on the glass substrates using chemical method. The Cu₂SO₄, ZnSO₄, SnSO₄ and Na₂S₂O₃ are used as sources of Cu⁺, Zn^{2+,} Sn4⁺ and S²⁻ ions, respectively. The chemical bath containing 0.05 M Cu₂SO₄, 0.1 M $ZnSO_4$, 0.05 M SnSO₄ and 0.2 M Na₂S₂O₃ solutions in equal volume ratio was prepared by mixing them in a beaker. The solutions of aqueous ammonia added to it. The final pH of the resulting solution is 12. Previously cleaned glass substrate was immersed in the bath and then the bath was heated up to 40 °C for 75 min [15].

III. RESULTS AND DISCUSSION

Fig. 1 shows XRD patterns of as deposited CZTS thin films on glass substrate. The CZTS thin film shows the crystalline nature. The average crystalline size of CZTS thin film is around of 27 nm [15].

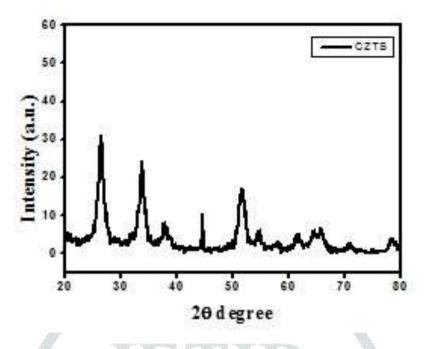


Fig. 1. XRD of CZTS thin film

The UV-ViS absorption and band gap spectra of each of the Copper Zink Tin Sulfide (CZTS) film. The absorption spectra range from 700 to 900 nm in visible range. The band gap of CZTS thin film is calculated show 1.68 eV, it is match or near to solar cell absorber layer [15]. Fig 2 and Fig 3 shows the absorption and band gap respectively.

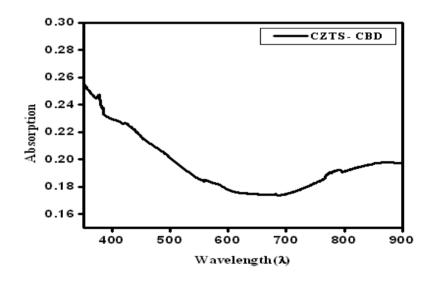
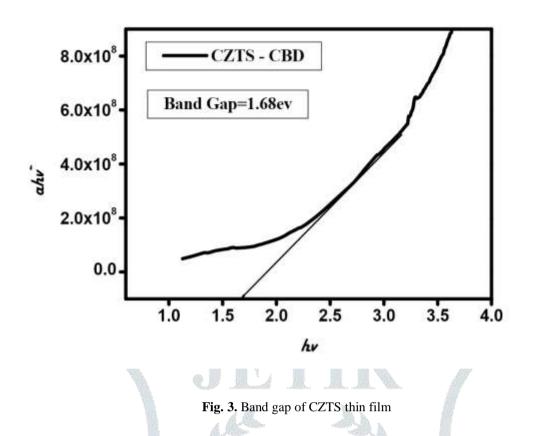


Fig. 2. Absorption of CZTS thin film



IV. CONCLUSION

The CZTS thin film has prepared by chemical bath deposition at 40 °C for 75 min on glass substrate. The XRD show the CZTS has polycrystalline nature and the crystalline size 27 nm. The band gap of CZTS thin film was 1.68eV found by optical properties. The chemically prepared nanocrystalline CZTS is polycrystalline nature and band gap is suitable for solar cell application.

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