

# Comparison between optical properties of doped and undoped GeO<sub>x</sub> Nanowires

<sup>1</sup>A. S. Katkar, <sup>2</sup>Y. B. Kholam, <sup>3</sup>P. N. Shelke

<sup>1</sup>Department of Physics, Dr. B.N. Purandare Arts and Smt. S.G. Gupta Commerce College, Valvan, Lonavla 410 403, India.

<sup>2</sup>Department of Physics, B.R.Gholap College, Sangvi, Pune 411027, India

<sup>3</sup> Department of Physics, Anantrao Pawar College, Pirangut, Pune 412 115, India.

**Abstract:** Cr-doped germanium oxide nanowires, germanium oxide nanowires have been synthesized using physical vapor deposition method. For growth of nanowires and Cr-doped nanowires the experimental conditions were similar except doping source. The doped germanium oxide nanostructures such as Cr-doped germanium oxide nanowires will be effective in influencing the optical properties of germanium oxide nanostructures, since they may tune their band gap energy.

**Keywords:** Germanium oxide; Doping; Nanowires; Nanotubes; Optical properties.

## I. Introduction

One-dimensional (1D) nanostructures are very promising candidates because of their potential for elucidating promising applications in magnetic, optical, and electronic devices [1]. Owing to contribution to understand fundamental concepts in mesophysics and potential applications in nanoelectronic devices, recently special attention has been paid to 1D nanostructures such as nanotubes and nanowires. Germanium oxide (GeO<sub>x</sub>) is of particular interest, since it is a transparent conducting oxide (TCO) with high potential in optoelectronics communication devices such as optical waveguides for integrated optical systems [2]. The linear coefficient of thermal expansion and refractive index of GeO<sub>2</sub> glass is higher as compared to SiO<sub>2</sub> glass [3]. The GeO<sub>x</sub> is a blue luminescent material and plays vital role in vacuum technology. Hence fabrications of GeO<sub>x</sub> nanostructures are always useful for future optical nanodevices. The low-dimensional nanostructures of GeO<sub>x</sub>, such as nanowires, nanoneedles, nanowhiskers or nanotubes [4,5] are potentially useful for various industrial applications. Using various physical and chemical methods such as physical vapor deposition, carbothermal reduction, laser ablation and thermal oxidation, different nanostructures: nanowires, nanoneedles, nanotubes and nanowhiskers of GeO<sub>x</sub> have been prepared by germanium (Ge) and GeO<sub>2</sub> precursor powders [6-8]. The doping in GeO<sub>x</sub> nanostructures with rare earths elements, transition metal elements (Er and Eu) and metal impurities (Sn and Mn) shows that doping impurities can tune the optical properties of GeO<sub>x</sub> nanostructures [9-13]. The Ge nanowires, GeO<sub>x</sub> nanowires and nanotubes have been synthesized and their optical properties were investigated [14, 15]. The GeO<sub>x</sub> nanostructures have wide band gap energy and their optical properties promised interesting applications as in fabrication of optoelectronic devices. The doped GeO<sub>x</sub> nanostructures (nanowires and nanotubes) will be effective in influencing the optical properties of GeO<sub>x</sub> nanostructures since they may tune their band gap energy. As suitable impurities can affect physical properties of GeO<sub>x</sub> nanostructures, the waveguiding behavior of undoped and doped germanium oxide nanowires and nanotubes would be interesting and will be useful to fabricate future optical nanodevices.

In view of this, in present work, GeO<sub>x</sub> nanowire and Cr doped GeO<sub>x</sub> nanowires are synthesized by using a simple vapor transport nanotubes in the visible region is explored.

## II. Experimental

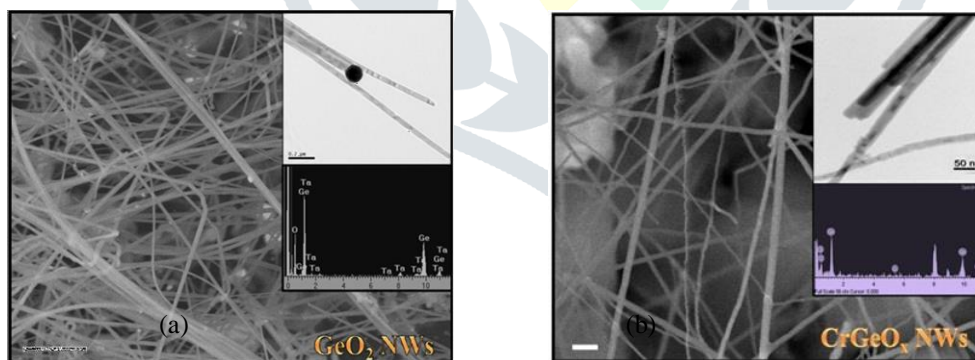
The commercial germanium (Ge), germanium oxide (GeO<sub>2</sub>), chromium oxide (Cr<sub>2</sub>O<sub>3</sub>), (all from Alfa Aesar and 99.99 % pure) powders were used as precursor materials. The p-type silicon (001) wafer was ultrasonically cleaned in ethanol for 15 min. The 3 nm thick gold film was deposited on wafer at room temperature (RT) by using an electron beam evaporation

system at a pressure  $\sim 6 \times 10^{-6}$  Torr. These gold coated silicon wafers were used as substrates for further experimentation. The  $\text{GeO}_x$  nanowires, Cr-doped  $\text{GeO}_x$  nanowires, were grown on silicon wafer substrates by a vapor transport method using a three zone furnace. For preparation of  $\text{GeO}_x$  nanostructures, mixture of Ge and  $\text{GeO}_2$  with a 1:3 ratio was placed in an alumina boat and heated at 1100 °C (zone-I) at a rate of 20 °C/min with the reaction time of 60 min. and with 100 sccm Ar flowing through the tube. During this, gold coated silicon wafer substrates were placed in zone-II (at 750 °C) and to zone-III (at 850 °C) to grow  $\text{GeO}_x$  nanowires. After completion of reaction, the furnace was allowed to cool naturally to room temperature (RT). Similarly, for the preparation of Cr-doped  $\text{GeO}_x$  nanowires, the mixture of Ge,  $\text{GeO}_2$ ,  $\text{CrO}_3$  in 3:9:3 ratio was used and gold coated silicon wafer substrates were placed in zone-II (at 750 °C) and gold coated silicon wafer substrates were placed in zone-III (at 850 °C).

The morphological study of the resultant films was done by using field emission scanning electron microscope (FESEM, JEOL JSM-6500F, accelerating voltage = 10 kV) and transmission electron microscope (TEM, JEOL JEM-2010, 200 kV) equipped with an energy-dispersive spectrometer (EDS). EDS was used for analysis of chemical composition of undoped and doped  $\text{GeO}_x$  nanowires. The influence of Cr, dopant on the photoluminescence properties of undoped and doped  $\text{GeO}_x$  nanowires in the visible region have also been explored.

### III. Result and Discussion

The substrate temperature plays vital role in the morphological change in  $\text{GeO}_x$  nanostructures. The white products were obtained on substrates during experimentation at temperatures of 750 and 850 °C. Figure 1 (a) shows FESEM image, corresponding TEM image and EDS spectrum of the  $\text{GeO}_x$  nanowires. It confirms the growth of nanowire structure with diameter and length in the range of 25 - 30 nm and 10 - 15  $\mu\text{m}$  respectively for product obtained at 750 °C. The insets of figure 1(a) show TEM image of the nanowires with the gold tip and EDS spectrum of corresponding nanowires. The EDS spectrum confirms the composition of Ge and O. Thus at substrate temperature of 750 °C,  $\text{GeO}_x$  nanowires are formed. Figure 1(b) depicts FESEM image of Cr-doped  $\text{GeO}_x$  nanowires, corresponding TEM image and EDS spectrum as insets. The FESEM image of the product obtained at 750 °C confirms the growth of high density nanowire structure with diameter and length in the range of 25 - 30 nm and 10 - 15  $\mu\text{m}$  respectively. The insets of figure 1(c) show TEM image and EDS spectrum of corresponding nanowires. The EDS spectrum confirms the doping of Cr in  $\text{GeO}_x$  nanowires.



**Figure 1** FESEM image (scale bar = 1  $\mu\text{m}$ ) of  $\text{GeO}_x$  nanowires (Insets: corresponding TEM image and EDS spectrum), (b) FESEM image (scale bar = 100 nm) of Cr-doped  $\text{GeO}_x$  nanowires (Insets: corresponding TEM image and EDS spectrum)

Figure 3 (a) shows that photoluminescence (PL) emission spectrum of the as-prepared  $\text{GeO}_x$  nanowires measured upon photoexcitation at 325 nm. In order to have closer insights for the origin of emission, spectral feature are fitted with Gaussian functions. In case of  $\text{GeO}_x$  nanowires, the best fit of the emission was obtained with two Gaussian functions, with the peaks centered at 420 nm (in blue region) and 472 nm (in bright blue region). Similarly, in case of Cr-doped  $\text{GeO}_x$  nanowires [Figure 2 (b)], the peaks are centered at 424 nm (in blue region) and 476 nm (in bright blue region).

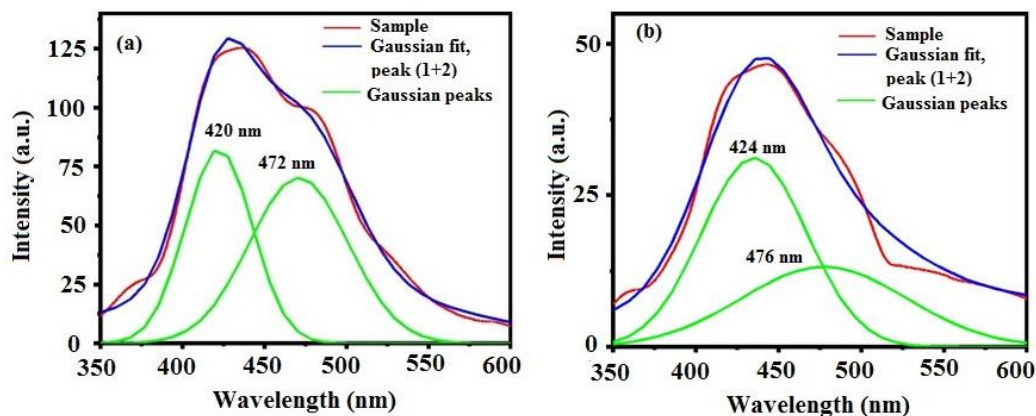


Figure 3 PL spectra of (a)  $\text{GeO}_x$  and (b) Cr-doped  $\text{GeO}_x$  nanowires at RT

Since the  $\text{GeO}_x$  nanowires and Cr-doped  $\text{GeO}_x$  nanowires are prepared at high temperature (1100 °C) and GeO is volatilized at this point, the oxygen vacancies and oxygen-germanium vacancy pairs easily exist in the products. It is thus reasonable that the blue light emission and bright blue emission can be attributed to oxygen vacancies and oxygen-germanium vacancies centers. The shift in peak position illustrates that the Cr-doping can slightly tune the structure of energy level and band gap of  $\text{GeO}_x$  nanomaterials. Further, Cr-doping can increase the defects and oxygen vacancies in  $\text{GeO}_x$  nanomaterials.

#### IV. Conclusions

The undoped  $\text{GeO}_x$  nanowires and Cr-doped  $\text{GeO}_x$  nanowires are synthesized by using a simple vapor transport method at high temperature. The red shift in photoluminescence spectrum of Cr-doped  $\text{GeO}_x$  nanowires as compared to undoped  $\text{GeO}_x$  nanowires shows that the optical properties of the  $\text{GeO}_x$  nanowires can be tuned by Cr doping. It suggests the possible control of defects in nanowires and oxygen vacancies in  $\text{GeO}_x$  nanowires. Thus the optical properties of the  $\text{GeO}_x$  nanowires can be controlled by using different doping elements, which will improve their application region in functional devices.

#### References

- [1] C. Ye: *Sci. Adv. Mater.* Vol. 2 (2010), p. 365.
- [2] C. Yan, M.Y. Chan, T. Zhang and P.S. Lee: *J. Phys. Chem. C* Vol. 113 (2009), p. 1705.
- [3] Z. Y. Yin and B.K. Garside: *Appl. Opt.* Vol. 21 (1982), p. 4324.
- [4] Z. Jiang, T. Xie, G.Z. Wang, X.Y. Yuan, C.H. Ye, W.P. Cai, G.W. Meng, G.H. Li and L.D. Zhang: *Mater. Lett.* Vol. 59 (4) (2005), p. 416.
- [5] H.W. Kim, J.W. Lee, M.A. Kebede, H.S. Kim and C. Lee: *Current Appl. Phys.* Vol. 6 (2009), p. 1300.
- [6] Y.H. Tang, Y.F. Zhang, N. Wang, I. Bello, C.S. Lee and S.T. Lee: *Appl. Phys. Lett.* Vol. 74 (1999), p. 3824.
- [7] P. Hidalgo, B. M'endez and J. Piqueras: *Nanotech.* Vol. 16 (2005), p. 2521.
- [8] H.Y. Dang, J. Wang and S.S. Fan: *Nanotech.* Vol. 14 (2003), p. 738.
- [9] P. Hidalgo, B. M'endez and J. Piqueras: *Nanotech.* Vol. 18 (2007), p. 155203.
- [10] P. Hidalgo, B. Liberti, Odr gue-a canonde and J. Piqueras: *J. Phys. Chem. C* Vol. 113 (39) (2009), p. 17200.
- [11] P. Hidalgo, B. M'endez and J. Piqueras: *Nanotech.* Vol. 19 (2008), p. 455705.
- [12] A.S. Katkar, Y.-C. Chu, L.-W. Chu and L.J. Chen: *Cryst. Growth & Des.* Vol. 11 (2011), p. 2957.
- [13] Y. Mao, J.Y. Huang, R. Ostroumov, K.L. Wang and J.P. Chang: *J. Phys. Chem. C* Vol. 112 (7) (2008), p. 2278.
- [14] A.S. Katkar, P.N. Shelke and Y.B. Kholam: *Inter. J. Chem. & Phys. Sci.* Vol. 4 (2015), p. 45.
- [15] A.S. Katkar, P.N. Shelke and Y.B. Kholam: *Inter. J. Chem. & Phys. Sci.* Vol. 5 (2015), p. 30.