



Synthesis and luminescence properties of $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphors for radiation dosimetry

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Abstract

The goal of this work was to investigate the relevant dosimetric and luminescent properties of $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphor for radiation dosimetry. The structural property was studied through X-ray diffraction technique. Additionally, the photoluminescence (PL), thermoluminescence (TL) and optically stimulated luminescence (OSL) behaviors of $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphor was studied. The XRD pattern of $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphor fully matched with the International center for diffraction data (ICDD) file. The PL spectra of as-prepared $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphor showed characteristic emission in near-UV region, The TL glow curve of $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphors were consist overlapping peaks in temperature 50–300°C. Also $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba) phosphors show excellent CW-OSL response under β irradiation.

Keywords: Radiation dosimetry; Luminescence properties; X-ray diffraction; Thermoluminescence; Optically Stimulated Luminescence; $\text{MSO}_4:\text{Eu}$ (M=Ca, Ba).

1. Introductions

Radiation dosimetric methods are used for the estimation of dose absorbed by radiation in a detector material. These methods are required for estimation of absorbed dose in various applications of radiation, such as personnel and environmental dosimetry, retrospective, accidental, dosimetry and medical applications of radiation. The use of Thermoluminescence (TL) as a method for radiation dosimetry of ionizing radiation has been established for many decades and has found many useful applications in various fields, such as personnel, environmental, medical, archaeological, geological dating and space dosimetry. Several high

sensitive TL phosphor materials and thermoluminescent dosimeters (TLDs) are now commercially available in different physical forms. There are many commercial TLD systems which are being used for various dosimetric applications and even presently, TL is a popular technique in the field of radiation dosimetry, particularly in personnel monitoring [1-4].

Thermoluminescent materials based on rare earth (RE)-doped alkaline earth sulphates have been studied since 1970 [5]. Sulphates are known to be good Thermoluminescent materials. Sulphate based RE doped phosphors, particularly alkaline earth sulphates, because of its high sensitivity, ease of preparation and stability of response in adverse climates, have already been very popular for use in radiation dosimetry, using thermoluminescence [6]. The main characteristics of some Eu activated sulphate TL materials are tabulated in **Table 1**.

Table 1. The main characteristics of some Eu activated sulphate based TL material

S.N	Material	Z_{eff}	Glow peak Temp (°C)	Synthesis Method	Application	Ref.
1.	BaSO ₄ :Eu	30.3	190	Precipitation method	Radiation dosimetry using OSL as well as TL	[7]
2.	BaSO ₄ : Eu	30.3	-	Recrystallization method	Dosimetric applications	[8]
3.	SrSO ₄ :Mn	46.5	346	Wet chemical precipitation method	High temperature radiation dosimetry applications	[9]
4.	SrSO ₄ :Eu	46.5	231	Acid evaporative method		[10]
5.	SrSO ₄ :Eu	46.5	312	Chemical precipitation method	Research in mixed fields detection dosimeters	[11]
6.	SrSO ₄ :DyTb	46.5	217	Co-precipitation method	Estimating high dose of gamma rays	[12]
7.	CaSO ₄ : Eu	15.3	180	-	TLD	[13]
8.	CaSO ₄ : Eu	15.3	173	Chemical co-precipitation method	High dose in space craft payload	[14]

In the last two decades an alternative technique namely OSL has been developed and widely used. The OSL technique is the best of all the known techniques for measurement of radiation exposure since the out process does not involve problems of blackbody radiation and thermal quenching as in TL [15-18].

Many studies have searched for new OSL materials with suitable properties for dosimetry, including phosphate, borate, sulfate and halides [19-34]. Rare earth activated MSO_4 is an interesting material for TL/OSL, it has a low effective atomic number with hydrated Orthorhombic phases [35]. Rare earths (RE) activated sulfate based phosphors are widely used for radiation dosimetry.

The purpose of this work was to investigate the dosimetric and luminescence properties of a $\text{MSO}_4:\text{Eu}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphor, this phosphor is developed via co-precipitation method for radiation dosimetry application. However, to the best of our knowledge, there are no reports on the Comparative studies of TL/OSL properties of $\text{MSO}_4:\text{Eu}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphor.

2. Experimental details

MSO_4 ($\text{M}=\text{Ca}, \text{Ba}$) phosphors activated with Eu were prepared by co-precipitation method described in our earlier works [36]. The stoichiometry of the reaction is maintained by formula $\text{M}_{1-x}\text{SO}_4:x\text{Eu}^{2+}$ ($\text{M}=\text{Ca}, \text{Ba}$). The nitrate precursor of Calcium/ Barium was dissolved in 100 ml of double-distilled water with addition of the stock solution prepared for Eu_2O_3 . Solution in glass beaker under stirring to form a homogeneous aqueous solution and it was confirmed that precursor was dissolved in water. 10 ml H_2SO_4 solution was added drop by drop into the mixed aqueous solution of $\text{M}_{1-x}(\text{NO}_3)_2: x\text{Eu}$ ($\text{M}=\text{Ca}, \text{Ba}$) under rigorous stirring at room temperature and white precipitation formed. After that, the MSO_4 ($\text{M}=\text{Ca}, \text{Ba}$) precipitate was centrifuged and washed several times by water to remove the excess residual salts. The precipitate was dried at 60°C for 2 hr by optical heating. The dried sample was annealed at 900°C for 1 hr to get white crystalline powder of $\text{MSO}_4:\text{Eu}^{2+}$ ($\text{M}=\text{Ca}, \text{Ba}$). The complete process involved in the reaction was represented as a flow chart in **Fig. 1**.

1.

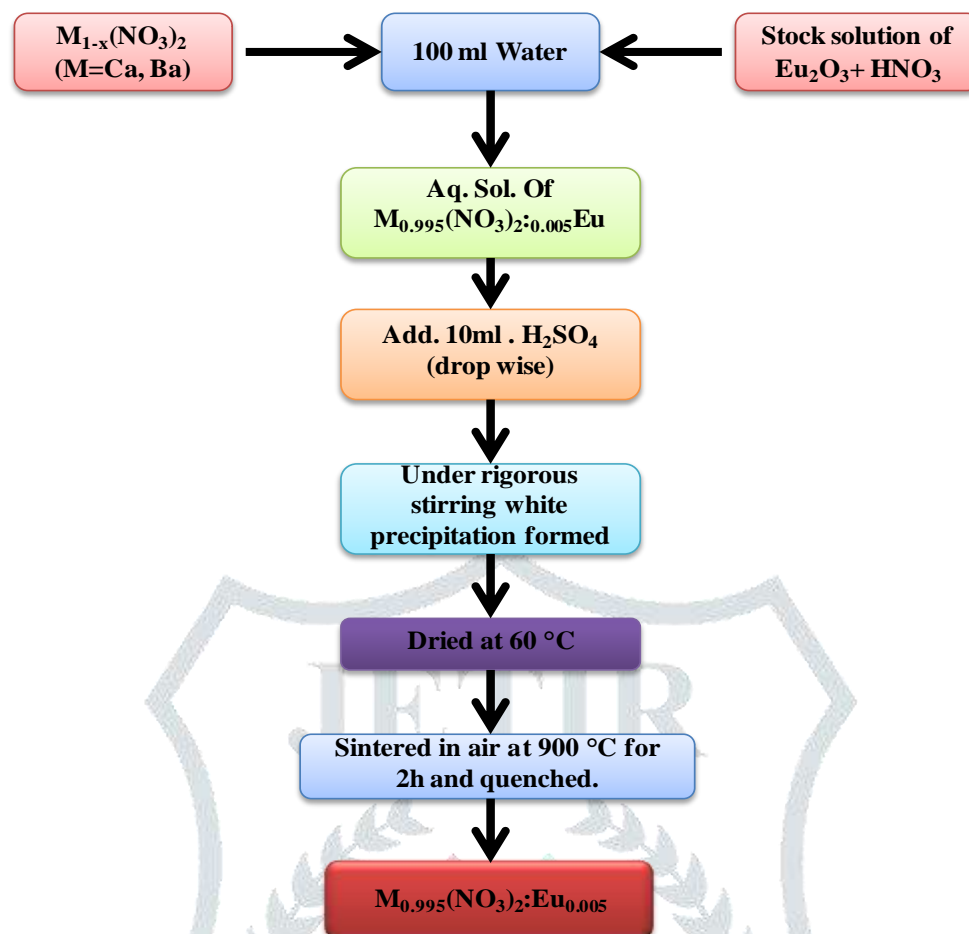


Fig. 1 Flow chart of $M_{(1-x)}SO_4:xEu^{2+}$ (M=Ca, Ba) synthesized by Co-Precipitation method.

3. Results and discussion

The structure of the as-prepared samples were analyzed by Rigaku Miniflex X-ray diffractometer, using monochromatic $CuK\alpha_1$ ($\lambda = 1.5405 \text{ \AA}$) radiation in the 2θ range of $10-60^\circ$. Photoluminescence was studied on a Hitachi F-7000 fluorescence spectrophotometer. Emission and Excitation spectra were recorded using a spectral slit of 2.5 nm for each window. For studying the TL and OSL response, all the samples were irradiated using $^{90}Sr/^{90}Y$ beta source with the dose rate of 20mGy per minute. All OSL measurements were carried out using an automatic Risø TL/OSL-DA-15 reader system which can accommodate up to 48 discs at IGCAR . Blue LEDs emitting at 470 nm (FWHM = 20 nm) are arranged in four clusters each containing seven individual LEDs. The total power from 28 LEDs at sample position is $w80 \text{ mW/ cm}^2$. A green long pass filter (GG-420) is incorporated in front of each blue LED cluster to minimize the amount of directly scattered blue light from reaching the detector system. The standard PMT used in the Risø TL/OSL luminescence reader is a bialkali EMI 9235QA, which has an extended UV response with maximum detection efficiency

between 300 and 400 nm. To prevent scattered stimulation light from reaching the PMT, the Risø reader is equipped with a 7.5 mm Hoya U-340 detection filter, which has a peak transmission around 340 nm (FWHM ~80 nm)

3.1. X-ray diffraction pattern

The structure of $\text{MSO}_4:\text{Eu}^{2+}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphor was orthorhombic.

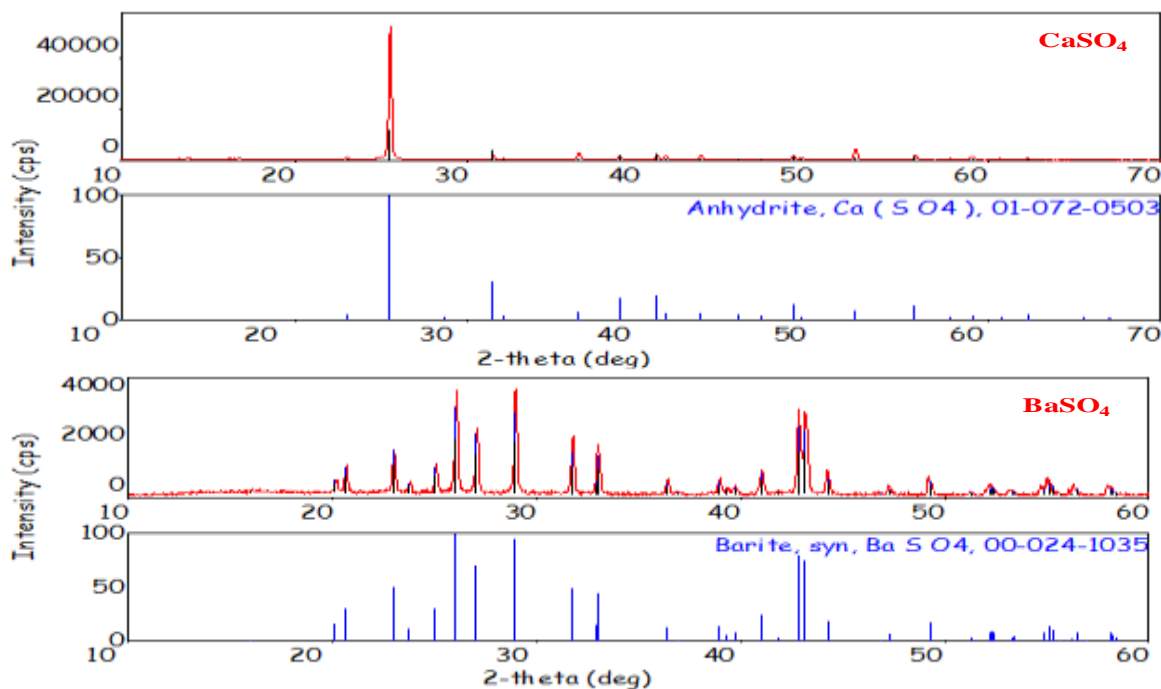


Fig. 2 X-Ray diffraction patterns of $\text{MSO}_4:\text{Eu}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphors with ICDD patterns

In order to determine the phase purity, chemical nature of the phosphor, X-ray diffraction (XRD) analysis was carried out. **Fig. 2** show the XRD pattern of $\text{MSO}_4:\text{Eu}^{2+}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphor along with the standard XRD pattern (ICDD Card No. 01-072-0503 and 00-024-1035). The XRD pattern shows the formation of pure CaSO_4 and BaSO_4 phase. The addition of the dopant (Eu) does not seem to have effect on the XRD pattern which suggests that the dopant was incorporated in the lattice.

3.2 Photoluminescence Properties (PL)

The combine excitation and emission spectra of $\text{MSO}_4:\text{Eu}$ ($\text{M}=\text{Ca}, \text{Ba}$) phosphors were give in **Fig. 3**. The excitation and emission spectra of $\text{BaSO}_4:\text{Eu}$ phosphor was observed at 377 nm and 326 nm respectively. The excitation and emission spectra of $\text{CaSO}_4:\text{Eu}$ phosphor was observed at 384 nm and 320 nm respectively.

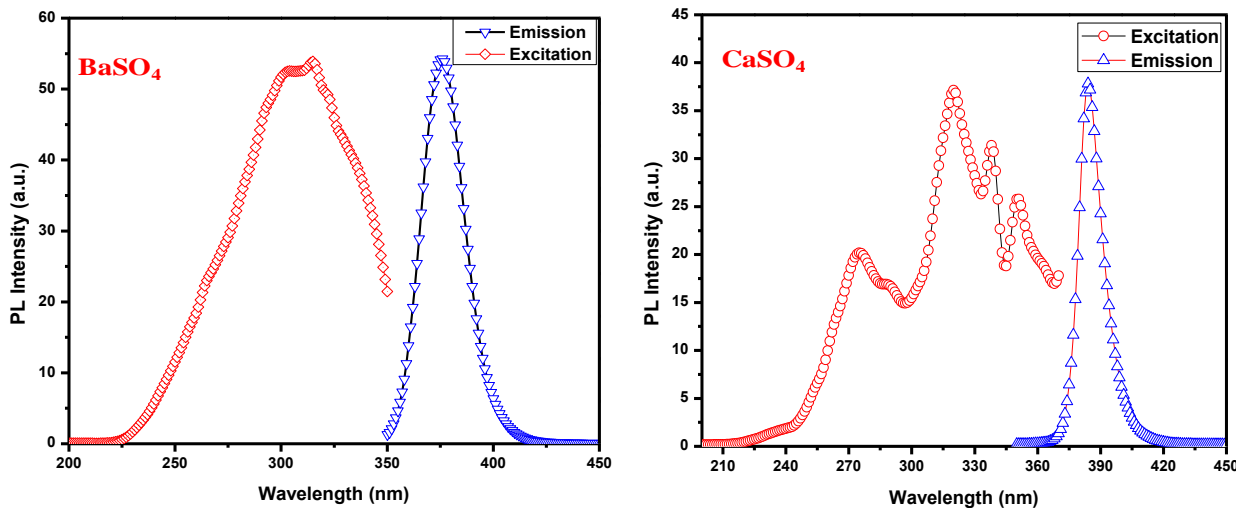


Fig. 3 Excitation and emission spectra of MSO₄: Eu (M=Ca,Ba) phosphors

3.3 Thermoluminescence (TL)

Thermoluminescence glow curve of MSO₄: Eu (M=Ca, Ba) phosphors under β irradiation as shown in Fig. 4. The TL glow curve of the MSO₄: Eu (M=Ca, Ba) phosphor consists of overlapping peaks in temperature rang 50-300°C. The TL sensitivity of BaSO₄:Eu phosphor was 5 time sensitive than TL sensitivity of CaSO₄:Eu phosphor.

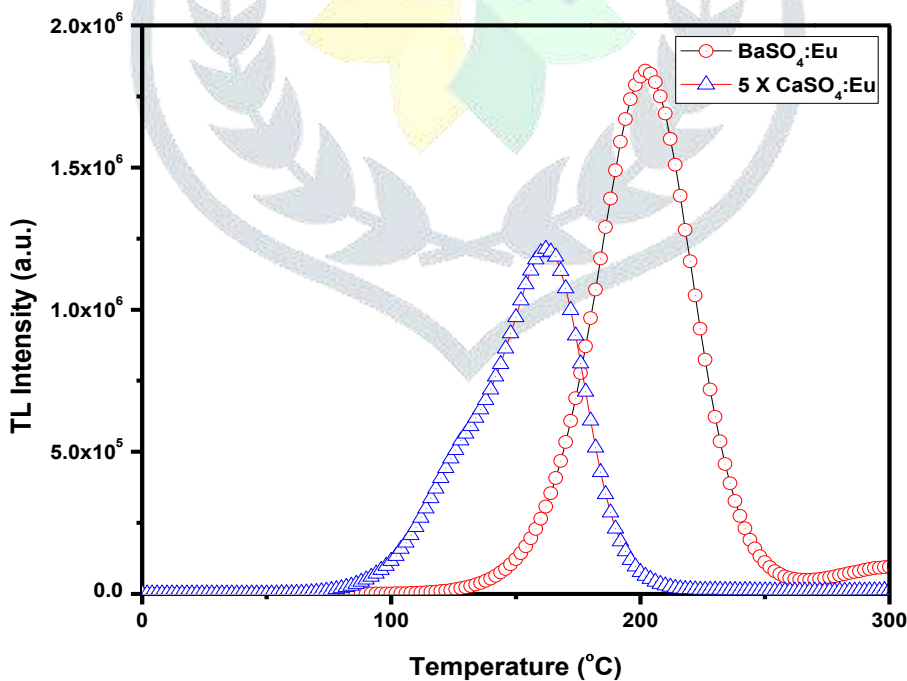


Fig. 4 Comparison of TL responses MSO₄: Eu (M=Ca,Ba) phosphors under beta irradiation

3.4 Optically stimulated luminescence

The CW-OSL response of prepared $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphor under 100mGy of β irradiation as shown in **Fig. 4**. The OSL sensitivity of compared with two different [37]. In the first method, The OSL sensitivity of $\text{BaSO}_4:\text{Eu}$ phosphor was found to be about 40 times to that of $\text{CaSO}_4:\text{Eu}$ phosphor. OSL sensitivity of $\text{CaSO}_4:\text{Eu}$ phosphor was more than $\text{BaSO}_4:\text{Eu}$ phosphor by using second method. The difference in the sensitivities in the two methods is because of the fact that the OSL decay in the $\text{BaSO}_4:\text{Eu}$ based phosphors is faster as compared to $\text{CaSO}_4:\text{Eu}$ phosphor.

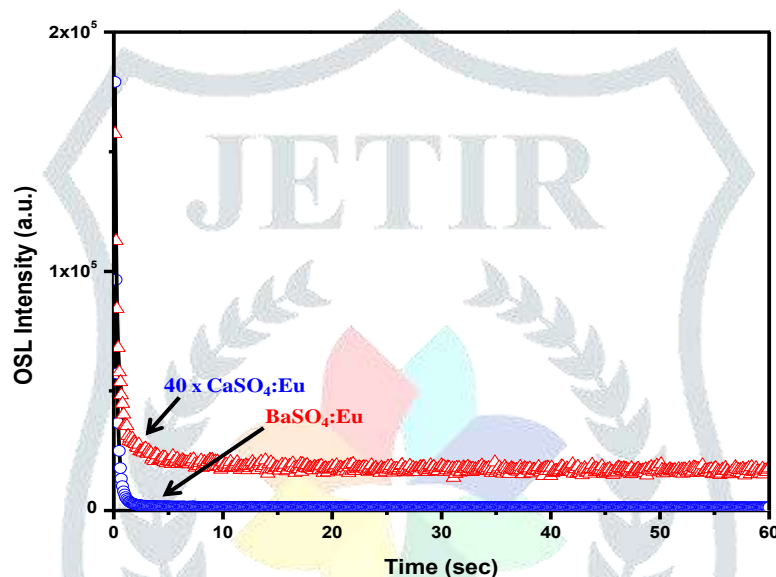


Fig. 4 Comparison of OSL responses $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphors under beta irradiation

4. Conclusions

In this reports $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphor was synthesized via Coprecipitaion method. The XRD pattern of prepared $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphor was perfectly matched with ICDD file. The TL glow curve consist overlapping peaks in temperature 50-300°C range. The emission spectra of prepared phosphor was obsreved in 350 – 450 nm range. The CW-OSL decay pattern of prepared $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphor was same than CW-OSL decay pattern of commercial available $\alpha\text{-Al}_2\text{O}_3:\text{C}$ phosphor. The OSL sensitivity of $\text{BaSO}_4:\text{Eu}$ phosphor was found to be about 40 times to that of $\text{CaSO}_4:\text{Eu}$ phosphor. The OSL decay in the $\text{BaSO}_4:\text{Eu}$ based phosphors is faster as compared to $\text{CaSO}_4:\text{Eu}$ phosphor. The effective atomic number of prepared $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) phosphor is nearly s 15. Although $\text{MSO}_4:\text{Eu}$ ($M=\text{Ca}, \text{Ba}$) is not a

material equivalent to tissue but, due to its high TL and OSL sensitivities, this phosphor can be proposed as a suitable candidate for radiation dosimetry, of course, after further progress in the studies.

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