

SOLVENT EXTRACTION OF OIL AND ITS ECONOMY

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Abstract: The processing of oil seeds and oil containing fruits becomes vital because the people need more and more oil for human nutrition and also for technical applications. There are three different methods for the extraction of oil are conceivable: (1) pressing (2) extraction by solvent and (3) pressing followed by solvent extraction. Pressing is a very old method which is replaced by the use of solvent which is more efficient, especially for oil seeds with lower oil content such as soybeans. Seeds with higher oil content are repressed before extraction by solvent. We discussed the different aspects of oil processing of the raw material via extraction either by pressing or solvent extraction. We discussed Factors Influencing the Extraction Efficiency, Choice of solvent, Separation of Solvent and Oil, Relative Yields and Advantages of solvent extraction. Oil extraction process which we are going to use it increases the yield of oil%. We also studied Edible Oil Production and Consumption, Indian vegetable oil economy.

Keywords: Mechanical Pressing, Solvent Extraction, Solvent Recovery System, Hexane

I. INTRODUCTION

Oil production from different sources is important to provide oil for the use as food or in several technical applications, such as fuel, lubricants, or raw material etc. The main point is the separation of the oil from the oil-bearing material with subsequent purification of the raw oil, but looking at the oil production as a whole not only the oil processing is important for the production of high quality oil, but also the pretreatment of the raw material, and the storage conditions unit processing have to be taken into consideration. The aim of the extraction method is to increase the oil yield with simultaneous maintenance of the oil quality. While mechanical extraction of seeds or fruits, a old method for the processing of oil. This method is nowadays used only for fruit oils or specialty oils, in the extensive extraction process in large facilities, mechanical and solvent extraction are applied.

II. PRODUCTION PROCESS

Edible oil technology can be grouped into two: mechanical pressing and solvent extraction. For oil seeds with high oil content such as groundnut, first mechanical pressing will be applied to extract the oil. The remaining oil in the expeller cake will then be extracted using solvent. Oil seed with low oil content such as soybean, sunflower, rapeseed etc. solvent extraction is generally considered as the best alternative.

A. Mechanical Pressing

The aim of the pressing process is to separate the oily phase from the solid phase of the seed material. In large scale facilities, two different pressing techniques are used for the separation of oil from the seed. The first technique is a more extensive pressing of the oil seed without further extraction by solvent. This mechanical pressing is useful for oil seeds with high oil content, where it is not a problem to get extra release of some percent of oil in the residue. This has the advantages of lower investment and maintenance costs and higher oil quality. Oil press consists of a rotating screw shaft (Fig. A.1) within a horizontal barrel. This screw shaft transports the seeds through the screw flight, in order to increase the pressure on material during the passage, the volume of the screw increases or the diameter of the screw flight decreases in the direction of the seed flow resulting in less space for the seed material and increasing pressure onto the material. Depending on the needs of the seed material, it is possible to vary the slope of the screw which results in a change of the pressure increase within the screw press. The barrel is surrounded by flat bars, which form a confine around the screw (Fig. A.2). Between the bars is some space, in which oil can flow out of the barrel while most of the solid material leaves the screw press at the end. By varying the space between the bars, it is possible to change and to adjust the pressure within the barrel. Another probability is to influence the pressure within the barrel is the size of the press cake exit at the end of the barrel and the slope of the screw.

Fig. 1: Screw press without barrel cage to show worm sections



Fig. 2: Barrel of a screw press surrounded by flat bars during pressing of oil seeds Solvent Extraction

Extraction with solvents is the most effective method for the recovery of oil. When performed at low temperature, the solvent extraction has another advantage over screw pressing: better quality of oil produced. This is because during expelling a sudden heating of the oil can occur, changing some parameters of its quality. The aim of extraction by solvent is to remove a large amount of oil from the oil seed. Hexane is extensively used as a solvent, because it is cheap, has good oil solubility at a relatively low temperature, has an appropriate boiling temperature, is non-corrosive to metal, inert to the oil. It is stable under the operating conditions, is immiscible with water, which causes easy separation of water from the oil, and it is easily and completely removed from the bottom product with low energy input and without harm to the raw oil. During solvent extraction of the oil seed, an intensive contact between solvent and press cake is necessary to achieve a complete removal of oil. The solvent is heated to 50-60°C, but it is important to avoid boiling when hot press cake comes into contact with the solvent. Press cake comes with high temperature from the mechanical pressing process, a further temperature increase is only necessary when the extractor starts working. Temperature is important for the ex-

tractions since viscosity of the solvent is reduced so it can easily flow into oil cake and the solubility of the extract increases with higher temperatures. The composition of the extract is influenced by the extraction temperature. While most oils mainly consist of triacylglycerides, minor components such as phospholipids, chlorophyll, free fatty acids, color pigments, and degradation products of oxidative reactions are coextracted by the solvent. This amount of minor components increases largely with the temperature. For example, an increase in temperature from 313 K to 331 K raises the content of phospholipids in rapeseed oil from 0.2 to 0.8%. Important factors influencing the result of solvent extraction are moisture, which can come from the surface of the press cake or from poor water/hexane separation after distillation. This moisture can avoid an optimal penetration of the press cake by the solvent, resulting in low extraction rates and high residual solvent in the meal. The extraction rate depends on temperature and moisture and also on the adjustment of equilibrium between diffusion of solvent into cake particles to dissolve oil and diffusion of oil from within the particles out into the solvent for which a certain time is necessary. The highest amount of oil is extracted at initial condition of process: 75% in the first 25 min, whereas the total oil yield depends on the further number of extraction steps and the total extraction time. The extraction of the remaining 25% oil from the press cake takes more time. The structure and the behavior of the cake particles are important for the result of the extraction process. Excessively fine press cakes resulting from a large number of handling of the cake before processing, from poor settings of the screw press during pressing or from improper moisture content of the raw seed material negatively influence the percolation rate of the solvent through the cake. While soybean flakes are stable enough to survive the mechanical and hydraulic strain during the extraction process. The result is small particles with a worse surface/volume ratio, which leads to a remarkable less percolation of the solvent through the cake bed. This again results in less extraction efficiency, higher residual oil content in the press cake and also a high level of solvent in the meal. Thus, a large number of extraction steps and a higher solvent requirement are needed to achieve a sufficient oil yield.

1) Process Solvent extraction of the press cake is carried out by using a continuous countercurrent extraction process moving the press cake and the solvent in different directions (Fig. B). This method is efficient in reducing the oil content of the press cake to a minimum, while the amount of solvent required remains comparably low. During the process, the cake comes into contact with solvent many times and after a certain time of adjustment of equilibrium the solvent/oil mixture, called a miscella, is allowed to drain away from the cake. With each step, the amount of oil in the cake is minimized, but since it is not possible to remove all of the solvent from the cake as some amount remains together with oil in the cake. The number of extraction steps depends on the performance of the extractor. More than ten successive wash stages have been used. However, an extensive extraction with fewer steps is possible if the cake is saturated by solvent allowing it to soak over a sufficient time followed by flushing the enriched miscella from the surface of the particles and replacing it with a leaner miscella. In such a case, only two or three extraction steps are required with a reduced number of pumps, thereby reducing costs of energy, capital, and maintenance. Fig. 3: Extractor For the extraction of the press cake or oil cake, a mixture of hexane and oil is used and the concentration of oil in hexane decreases with every extraction step. At the end of the process, before the meal leaves the extractor it will be washed with pure solvent.

III. FACTORS INFLUENCING THE EXTRACTION EFFICIENCY

Primary requirement of solvent extraction for separation or removal purposes is a high distribution ratio of the solute of interest between the two liquid phases. Continuous and countercurrent distribution techniques can be used for the cases where low distribution ratios are present. It is generally desirable to attain a high value as possible for the development of simple analytical procedures. Therefore numbers of different techniques are used for enhancing the distribution ratio. These depend on the nature of the species being extracted and extraction system. The attainment of selectivity in an extraction procedure is very important.

IV. CHOICE OF SOLVENT

Selection of suitable solvent for effective separation is very important. Metal chelates and many organic molecules, being essentially covalent compounds do not inflict many restrictions on the solvent and the general rules of solubility are used. In ion association systems and particularly in oxonium type ions, the role of solvent is very important. This is due to inclusion of solvent in the formation of extractable species. The choice of solvent depends on several important properties such as the good solubility, efficiency at low or ambient temperatures, low inflammability, boiling point and toxicity. The solvent must be easily recoverable and it is environmentally friendly. It is obviously impossible to come up with a solvent that fulfills these requirements fully. One solvent that has come very close to these criteria is hexane. The acceptance for this solvent has been such that it is somewhat of a monopoly in the solvent extraction of vegetable oils. Similarly, the degree of miscibility of the two phases, the relative specific gravities, viscosity and tendency to form emulsions should be considered. With regard to safety, the toxicity and inflammability of the organic solvents must be considered. The desired characteristics of a solvent can be achieved by using a mixed solvent system. An example of this is the use of mixtures of alcohols and ethers for the extraction of the thiocyanate complexes of metals. Another method of achieving desired characteristics of the extracting solvent is to use organic diluents.

V. SEPARATIONS OF SOLVENT AND OIL

The next step of solvent extraction process is the separation of solvent from oil. After the solvent-extraction process, the resulting miscella contains 20–30% oil in solvent. For an economical procedure, the recovery and re-utilization of the solvent is important. Separation of solvent needs energy and equipment, the aim must be to obtain as much oil as possible from the miscella with an economical need of energy. After leaving the solvent extractor, finest particles of the seed material resulting from the treatment during solvent extraction, also from the treatment during mechanical pressing, have to be separated from the miscella, because they can influence the heat transfer during distillation. The amount of species depends on the equipment used for the extraction. While direct solvent extraction without prepressing results in comparably higher amounts of particles, the amount is much lower if prepressing is used before solvent extraction. The separation of solvent and oil is achieved by distillation methods by using a series of stills, stripping columns, and condensers. In general, at two or three stage process is carried out. In a first stage, hexane and steam vapors from the desolventizer toaster are used for the removal of hexane from the meal are applied as heat source. In this stage, the mixture reaches temperatures of about 50°C. After that, the resulting oil enriched liquid is treated in a steam heated exchanger where the concentration of hexane decreases to about 5% of the miscella mass at a temperature of about 80°C. The hexane is again sent to the extractor for further use. Finally, the concentration of hexane has to be reduced to less than 800 ppm to reach a flash point of the oil higher than 250°C. Residues of hexane in the oil are finally removed during deodorization. To reach the required amount of hexane in the oil, the miscella is passed through an oil stripper tower under pressure of 0.13126 atm and temperature of about 373 K. A temperature higher than 378 K badly affects the quality of the crude oil, steam distillation is used. For this purpose, the oil is heated to 363–373 K in a preheater and then steam is injected from the bottom to increase the effectiveness of the process.

VI. RELATIVE YIELDS

The advantage of the solvent extraction process is the relatively high efficiency attained in the extraction of oil from oil-bearing materials. The usual oil content of the expeller oil cakes in India is between 68 percent by weight; while the deoiled meal has oil content less than one percent. Table (a) gives the relative yields of the products by different methods employed, for the recovery of oil from groundnut kernels.

Component	Yield percent of total oil		
	Ghani	Screw pressing	Solvent extraction
Oil	42.5	45.5	49.5
Meal	57.5	51.5	50.5
Percent efficiency	85.0	91.0	99.0

Table 1: Comparative Yield of Products by Different Methods: Processing Groundnut Kernels (50 percent oil by weight)

VII. ADVANTAGES

The advantages of the solvent extraction industry may be summarized as follows:

- 1) The recovery of oil from oil seeds is very high.
- 2) It can profitably save the oil from damage otherwise going as waste in the oil cakes.
- 3) No damage is done to the nutritional value of the entering materials in the process.
- 4) Nitrogen of the deoiled meal is supplied in such a form that it is easily made available to the soil, when used as manure.
- 5) It is quite possible to revive village ghani, driven by bullocks in a most efficient manner when worked in conjunction with a solvent extraction plant.
- 6) Deoiled meal produced under hygienic conditions of the plant from the edible oil seeds, may very well be utilized as human food.
- 7) The production of synthetic vegetable protein fibre is made possible.

VIII. EDIBLE OIL PRODUCTION AND CONSUMPTION

Increasing population, rising income levels and improved supply conditions are expected to continue to drive edible oil consumption to 17.4 million tons in 2012/13, up 1.08 million tons over the previous year. Palm oil is largest consumed edible oil. Because of its versatility in blending with other edible oils, competitive prices, and increased usage across, sweet food and butter industries, its food use consumption will rise to 8.4 million tons in 2013/14. Soft oils such as soybean and rapeseed are the largest edible oil which is 3.2 million tons and 2.6 million tons, respectively. Peanut, cottonseed and sunflower together constitute the 5.8 million tons. The per capita edible oil consumption in India is increasing (currently estimated at 14.04 Kg for 2012/13) remains far below the estimated world average per capita consumption of 22.4 Kg.

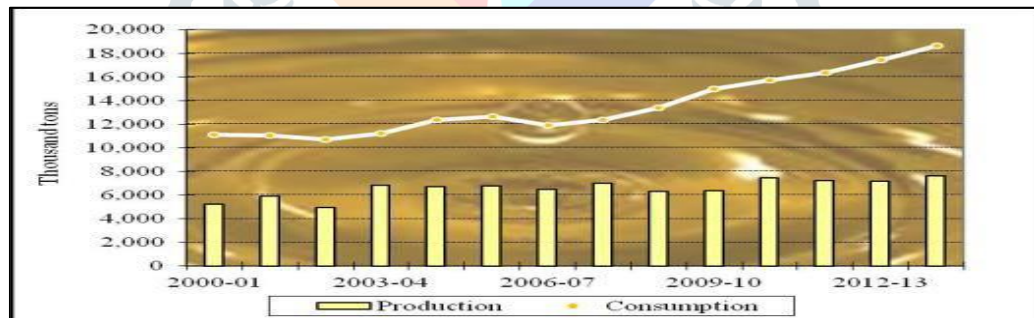


Fig. 1: India: Edible Oil Production and Consumption

IX. ECONOMY

Indian vegetable oil economy is the fourth largest in the world. The country accounts for 12-15% of global oil seeds area, 6-7% of vegetable oils production (next to U.S.A, China and Brazil) and 910% of the total edible oils consumption (FAO, 2011). Currently, India accounts for 6.8% of the oil meal production, 5.9% of the oil meal export, 6.1% of the vegetable oil export, 9.00% of the vegetable oil import and 9.3% of the edible oil consumption of the world (Sonnad et al., 2011). India ranks first in the production of most of the minor oil seeds (castor, sunflower and sesame). In the case of major oil seeds, India ranks first in the production of groundnut, third in rapeseed-mustard and fifth in soybean. All India Crop/Seasonwise Area, production and yield of oil seed crops during XIth plan period released by Directorate of Economics & Statistics, GOI, New Delhi is given in table 1. Table 2: All India Crop/Seasonwise Area, production and yield of oil seed crops during XIth plan period released by Directorate of Economics & Statistics, GOI, New Delhi

X. CONCLUSION

As we know, more than 98% of the oil production worldwide is carried out by solvent extraction, but for specialty oils, such as extra virgin olive oil or virgin rapeseed oil, oils produced in rural areas, or for pre-pressing before solvent extraction. The advantage of pre-pressing is that a press cake is formed from the small flakes which allow good solvent contact and percolation in the extractor. Extraction by combining screw pressing and solvent extraction has the economical advantages of low cost using a screw press and higher oil yield using solvent extraction. The advantage of solvent extraction over mechanical pressing can be clearly shown in the case of soybean by the fact that oil recovery is over 97% when solvent extraction is used as against about 80% when mechanical pressing is used. The yield percent of total oil by using ghani is 42.5 with screw pressing is 45.5 and finally with solvent extraction is 49.5 (on the basis of 50 percent oil by weight) which is evident that the oil yield is higher by using solvent extraction. The percent efficiency is 85, 91 and 99 percent for ghani, screw pressing and solvent extraction respectively. The production and consumption of oil in India is continuously increasing from year 2000-01 to 2012-

13. The consumption of oil in 2012-13 is more as compared to the production of oil. Indian vegetable oil economy is the fourth largest in the world. The country accounts for 12-15% of global oil seeds area, 6-7% of vegetable oils production (next to U.S.A, China and Brazil) and 9-10% of the total edible oils consumption. So we conclude that solvent extraction is most preferable and economical process for the production of edible oil.

REFERENCES

- [1] Economic Research Service of United States Department of Agriculture. (October, 2002). Oil Crop Situation and Outlook Yearbook. Available: <http://usda.mannlib.cornell.edu/reports/erssor/field/ocs-bby/ocs2002.pdf>.
- [2] 2004 Soya & Oilseed Bluebook, Soyatech, Inc., Bar Harbor, Maine, 2003.
- [3] T.J. Brummand C.R. Hurburgh, Jr., Quality of the 2002 Soybean Crop from the United States, Report to the American Soybean Association.
- [4] Aalrust E, Beyer W, Ottofrickenstein H (1992) Enzymatic method for reducing the amount of phosphorous-containing components in vegetable and animal oils. Europe Patent 0,513,709.
- [5] Ademosun OC (1982) Mechanised production system of oil palm produce in Nigeria: a preliminary study on the establishment of location-allocation models. *Agr Syst* 8: 193-207
- [6] Anonymous (2000) Monographs on the evaluation of carcinogenic risk to humans. Some industrial chemicals. IARC-World Health Organization International Agency for Research on Cancer, vol. 77, Lyon, France.
- [7] Anonymous (2001) Scientific Committee on food (2001). Opinion on 3-monochloro-propane-1,2-diol (3-MCPD). Updating the SCF opinion of 1994 adopted on 30 May 2001.
- [8] Anonymous (2003) Ergänzendetoxikologische Bewertung von 3-MCPD unter besonderer Berücksichtigung der Gefährdung von Kindern. Stellungnahme des BfR vom 9 Juli 2003
- [9] H.G. Kirschenbauer, Fats and Oils, Reinhold Publishing, New York, 1944, pp. 122-123.
- [10] Anonymous, in The Extraction of Oil from Seeds and Nuts, June 1997.
- [11] Anonymous, in Chronology, May (2000).
- [12] V.D. Anderson U.S. Patent 637,354, 1900.
- [13] Anonymous. (2002, March 18). Anderson International Corp. Available: <http://www.andersonintl.com/history.html>.
- [14] Abete, P., G. Testa, et al. (2009). "PUFA for human health: Diet or supplementation?" *Current Pharmaceutical Design* 15(36): 4186-4190.
- [15] AOCS (2005). Official methods and recommended practices of the AOCS. Gaithersburg, M.D., American Oil Chemists' Society.
- [16] Aryee, A.N.A. and B.K. Simpson (2009). "Comparative studies on the yield and quality of solvent-extracted oil from salmon skin." *Journal of Food Engineering* 92(3): 353-358.
- [17] Bermúdez Aguirre, D., T. Mobbs, et al. (2011). Ultrasound Applications in Food Processing. *Ultrasound Technologies for Food and Bioprocessing*. H. Feng, G. Barbosa-Canovas and J. Weiss. New York, Springer: 65-105.
- [18] Bligh, E.G. and W.J. Dyer (1959). "A rapid method of total lipid extraction and purification." *Canadian Journal of Physiology and Pharmacology* 37(8): 911-917.

