MILK ADULTERATION DETECTION TECHNIQUE AND METHODOLOGY

1Manoj Khotakar M, 2Dr Renuka Prasad B
1Student, 2Associate Professor & Associate Dean
3Department of MCA,
3RV College of Engineering, Bengaluru, India

Abstract: Milk in its common structure has a high food esteem, since it contains a wide assortment of supplements which are fundamental for appropriate development and upkeep of the human body. In ongoing many years, there has been an upsurge in milk utilization around the world, particularly in non-industrial nations, and it is currently framing a critical piece of the eating routine for a high extent of the worldwide populace. Milk is regularly exposed to misrepresentation (through corruption) for monetary benefit, yet it can likewise be defiled because of poorly educated endeavors to improve cleanliness conditions. Milk diluted, inferior cheaper materials may be added such as reconstituted milk powder, urea, and cane sugar, even more harmful chemicals including melamine, formalin, caustic soda, and detergents. These augmentations can possibly cause genuine wellbeing related issues. This survey is organized to be an 'adulterant based 'study instead of 'techniques based' one, where qualitative tracking for most of the common adulterants are listed and quantitative detection methods are limited to few major adulterants of milk. Apart from regular techniques, recent developments in these detection techniques have also been reported. Nowadays milk is being adulterated in more sophisticated ways that demands for cutting edge research for the detection of the adulterants. This review intends to contribute towards the common knowledge base regarding possible milk adulterants and their detection techniques.

I. INTRODUCTION

Milk is a supplement rich fluid food which incorporates protein and lactose. All through the world, more than six billion individuals consume milk and milk items. Entire milk, butter and cream have significant degrees of saturated fat. The sugar lactose is found in milk. Milk is an emulsion or colloid of butterfat globules inside a water-technique liquid that contains carbohydrates and protein aggregates with minerals. The pH of milk goes from 6.4 to 6.8 and it changes over the long haul. Typical bovine milk contains 30-35 grams of protein for each liter of which about 80% is arranged in casein micelles. Complete proteins in milk represent 3.2% of its composition. Milk contains a few distinctive carbohydrate including lactose, glucose, galactose, and different oligosaccharides. Metropolitan interest started to develop, as consumer buying power expanded and milk got viewed as a necessary everyday product. In the course of the most recent thirty years of the nineteenth century, interest for milk in many pieces of the nation multiplied or, at times, significantly increased.

As the interest developed, to arrive at the necessities of individuals the milk adulteration was begun. This adulteration of milk has been a danger to people in general. The milk contamination is the ill-conceived adding of unfamiliar synthetic mixtures to drain for different points, for example, expanding milk weight available to be purchased by adding water, expanding milk shelf-life usability especially in warm and hot seasons during transportation by adding detergents.

Techniques to identify adulterants in milk incorporate estimation of freezing point depression, electrical admittance spectroscopy, single-frequency conductance measurements, digital image chromatography, ultraviolet (UV) visible light spectroscopy, and enzyme linked immunosorbent assay. A range of focused logical techniques have been created to recognize specific foreign properties in milk; these incorporate high-performance liquid chromatography with single-channel UV-absorbance detection (HPLC–UV), and size exclusion chromatography and HPLC–mass spectrometry.

II. ADULTERANTS IN MILK

Adulterants in milk primarily incorporate expansion of vegetable protein, milk from various species, addition of whey and watering which are known as financially inspired adulteration. These adulterants don't represent any extreme wellbeing hazard. Nonetheless, a few adulterants are too unsafe to possibly be disregarded. A portion of the significant adulterants in milk having genuine unfavourable wellbeing impact are urea, formalin, detergents, ammonium sulphate, boric acid, caustic soda, benzoic acid, salicylic acid, hydrogen peroxide, sugars and melamine.

Regular boundaries that are checked to assess milk quality are- fat percentage, SNF (Solid-not-Fat) percentage, protein content and freezing point. Adulterants are included in milk to build these boundaries, in this way expanding the milk quality in exploitative manner. For instance, cane sugar, starch, sulphate salts, urea and common salts are added to expand solid-not-fat (SNF). Urea, being a characteristic constituent of crude milk, has a most extreme cut off forced by FSSAI (Food Safety and Standards Authority of India) Act 2006 and PFA Rules 1955 which is to be 70 mg/100 ml. commercial urea is added to milk to increment non-protein nitrogen content. Likewise, melamine is added to build protein content dishonestly. Ammonium sulphate is added to increase the lactometer reading by keeping up the density of diluted milk. Formalin, Salicylic acid, Benzoic acid and Hydrogen peroxide go about as additives and increment the timeframe of realistic usability of the milk. Since milk fat is extravagant, a few makers of milk and dairy items eliminate milk fat for extra monetary profit and remunerate it by adding non-milk fat like vegetable oil. Detergents are added to emulsify and break down the oil in water giving a foamy arrangement, which are the ideal qualities of milk.
Shockingly, a portion of the adulterants have serious health impact, now and again over the long haul. The ingestion of melamine at levels over safety limit can incite renal failure and death in infants [4]. Both the peroxides and detergents in milk can cause gastrointestinal difficulties, which can prompt gastritis and inflammation of the intestine. Extreme starch in the milk can cause diarrhea because of the impacts of undigested starch in colon, however, accumulated starch in the body may demonstrate exceptionally lethal for diabetic patients [3]. Urea in milk overburdens the kidneys as they need to filter through more urea content from the body [3]. In addition, carbonate and bicarbonates may cause interruption in hormone signalling that regulate development and reproduction.

### III. DETECTION TECHNIQUES

The discovery strategies can be separated into two different ways qualitative techniques and quantitative techniques. Qualitative detection of adulterants in milk is simple color based chemical reactions. Qualitative detection can be performed in any Bio safety Level 1 Laboratory with accessibility of chemical reagents and important safety measures. Significant disadvantages of these strategies are the realities that these are substantial for a restricted scope of fixations and are not adequately precise. Qualitative detection of a portion of those regular adulterants in milk has been examined in table 1. Quantitative strategies to distinguish adulterants in milk incorporate measurement of freezing point depression, electrical admittance spectroscopy, single-frequency conductance measurements, digital image chromatography, ultraviolet (UV) visible light spectroscopy, and enzyme linked immunosorbent assay.

Directed methodologies give explicit data on explicit adulterants and their abundance. Though untargeted approaches give less data on the idea of the adulteration. Untargeted examination of chromatographic and spectral data comprises of assessing the whole spectrum, and applying multivariate information investigation for the huge number of variables and samples created from these techniques. A scope of focused scientific techniques have been created to recognize explicit unfamiliar properties in milk; these incorporate high-performance liquid chromatography with single-channel UV-absorbance detection (HPLC–UV), and size exclusion chromatography and HPLC–mass spectrometry [1].

Improvements in infrared spectroscopic (IR) instrumentation and the mixture with chemo metric strategies have made this strategy priceless in determining food authenticity. IR is a quality affirmation device to determine functional and compositional examination of food ingredients, process intermediates, and finished products. Chemo measurements is an interdisciplinary examination that applies multivariate information investigation, which is often combined with information rich instrumental methods (like MS and infrared spectroscopy) utilized subjectively for categorizing obscure samples with similar qualities and quantitatively for deciding adulterants in samples. IR is profoundly attractive for investigation of milk components since it is direct to utilize, it can dissect tests with insignificant or no sample planning, it gives fast and online analysis, it has high affectability and explicitness, and it is feasible to run numerous tests on a solitary sample. Other spectroscopic techniques, to be specific, mid-infrared (MIR) and near-infrared, have been applied for the assurance of a few milk properties, including milk composition (protein, fat, and lactose), however these investigations are restricted with respect to milk authenticity. For instance, MIR spectroscopy combined with 2-dimensional correlation has been applied to distinguish adulteration by adding melamine, urea, and glucose to milk. The peak positions and shapes show contrasts between the control and adulterated milk, therefore approving the capability of this technique to recognize the milk adulteration.

**Table 1: Rapid qualitative detection of different mixed adulterants in milk**

<table>
<thead>
<tr>
<th>Adulterant</th>
<th>Procedure</th>
<th>Observation</th>
<th>Limit of detection (v/v) (Sharma et al. 2012)</th>
<th>References</th>
</tr>
</thead>
<tbody>
<tr>
<td>Detergent</td>
<td>A. Take 5 ml in a test tube and add 0.1 ml 0.5% Bromocresol Purple (BCP) solution.</td>
<td>Appearance of violet colour indicates the presence of detergent. Unadulterated milk shows faint violet color.</td>
<td></td>
<td>(Singhal, 1980); (Arvind Singh et al. 2012)</td>
</tr>
<tr>
<td></td>
<td>B. Take 5 mL of milk sample into a 15 mL test tube. Add 1 ml of Methylene blue dye solution and 2 ml chloroform. Vortex the contents for about 15 sec and centrifuge at about 1100 rpm for 3 min.</td>
<td>Relatively, more intense blue color in lower layer indicates presence of detergent in milk. Relatively more intense blue color in upper layer indicates absence of detergent in milk.</td>
<td>0.0125%</td>
<td>(Rajput, Sharma, &amp; Kaur)</td>
</tr>
<tr>
<td>Pulverized soap</td>
<td>Take 10 ml milk sample in a test tube. Add equal quantity of hot water to it, then add 1 – 2 drops of phenolphthalein indicator.</td>
<td>Appearance of pink color indicates presence of soap</td>
<td></td>
<td>(Arvind Singh et al. 2012); (Kamthania et al. 2014); (Ghodekar 1974)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>(Batis et al. 1981)</td>
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</tbody>
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### Coloring matter

<table>
<thead>
<tr>
<th>Step</th>
<th>Description</th>
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<tbody>
<tr>
<td>A.</td>
<td>Take 10 mL milk sample in attest tube. Add 10 ml diethyl ether. After shaking, allow it to stand.</td>
</tr>
<tr>
<td>B.</td>
<td>Make the milk sample alkaline with sodium bicarbonate. Dip a strip of filter paper for 2 hours.</td>
</tr>
<tr>
<td>C.</td>
<td>Add a few drops of hydrochloric acid to milk sample.</td>
</tr>
</tbody>
</table>

- Appearance of yellow color in ethereal layer indicates the presence of added color.
- Appearance of red color on filter paper indicates the presence of annatto. Treatment of this paper with stannous chloride gives pink color.
- Appearance of pink color indicates azo dyes.

Lechner and Klostermeyer 1981

(De Souza et al. 2000)

### IV. Technical Interventions

A specialized methodology towards this issue can be a little convenient and prudent milk test unit. The unit ought to have basic tests that the purchasers can do to check the nature of their milk. The pack ought to be reasonable and effectively accessible on the lookout and ought to be financed by the public authority. It ought to have a lactometer for testing the level of water in milk and few other compounds for other essential tests like powder milk, detergent, and urea. With such a strategy, purchasers can check the nature of milk prior to getting them from merchants. Merchants selling adulterated milk ought to be examined and his practices ought to be prohibited.

### V. Conclusion

The high dietary benefit of milk and its overall minimal expense contrasted and other protein sources has made it structure a critical piece of the diet of many populations worldwide. In any case, expanded interest around the world has made milk inclined to massive levels of fraudulent activity. Milk is a high-hazard product of worry for deceitful exercises for monetary benefit whereby culprits may expand sanitation risks and lessen wholesome quality through deliberate contaminated or potentially negligence under helpless cleanliness conditions, a lack of preservation, and no cooling facilities. This issue is more intense in the creating and immature nations because of absence of sufficient checking and law requirement. Existing regular recognition methods are not continuously advantageous and open in these nations making it hard to address the assorted methods of deceitful adulteration in milk. This calls for joined efforts from mainstream researchers and the administrative specialists through the turn of events, execution and spread of better methods for the recognition of milk adulteration. Also, awareness and admittance to data can assume imperative part in these areas to overcome this issue. Some of these simple detection strategies at the consumer level and best of the art techniques at the authority level can finish this issue for the people, including a millions of children in developing countries.

### REFERENCES

3. Singuluri & Sukumaran, Milk Adulteration in Hyderabad, India – A Comparative Study on the Levels of Different Adulterants Present in Milk DOI: 10.4172/2157-7064.1000212