



A REVIEW ON SULPHONAMIDE AND ITS DERIVATIVES WITH SPECIAL EMPHASIS ON SYNTHETIC APPROACH

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ABSTRACT

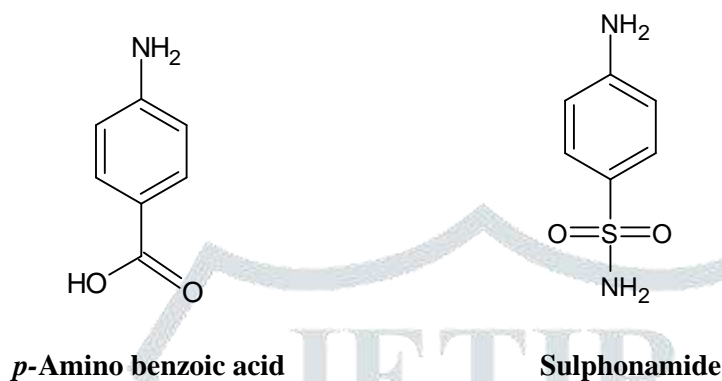
Sulphonamide is a lot of significant in restorative science and furthermore in engineered natural science. As of late there are a few techniques created for the union of Sulphonamide. Sulphonamide is a significant class of heterocyclic mixtures which has wide scope of natural properties. The current work is manages a few novel combination of Sulphonamide subsidiaries. Since, Sulphonamide is the most seasoned engineered moiety that is utilized as hostile to bacterial specialist which is utilized for controlling a few bacterial illnesses. Here, likewise different sulfa tranquilizes additionally shows against bacterial action like Sulphonamide, these medications incorporates Sulphathiazole, Sulphadiazine, and so forth Sulphonamide compound are particularly significant for different natural exercises because of their numerous organic exercises, low poisonousness and cost viability. In this paper we zeroed in on survey of different engineered approaches for subbed sulphonamides subsidiaries.

Key words: Sulphonamide, Sulphathiazole, Sulphadiazine, anti- bacterial, effectiveness, Heterocyclic compounds, synthetic schemes.

INTRODUCTION

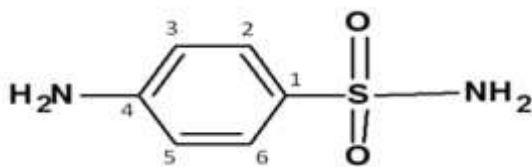
For treating various disease there is discovery and development of new agents are reported in medicinal chemistry practices. In which various heterocyclic compounds play an important role in development of various synthetic organic compound or various therapeutics agents. Sulphonamide serve as antibacterial agent that contain sulpha drugs that are derived from sulphonamides and it is used to prevent growth of bacteria. The compound of *p*-amino benzene now known as Sulphonamide was first synthesized by Gelmo in 1908 as an intermediate study of azo dye. In 1935, a German scientist prepared a red dye i.e. 4-sulphonamide-2, 4-aminobenzene and after three years Domagk introduce the various functions of the compound the named it

Protonsil. Then the scientist named was Trefouel et.al. (1935) at Pasteur Institute discovered that the breakdown of Protonsil in *p*-amino benzene Sulphonamide, now it is known as sulphonamides. In modern chemotherapy and the concept of prodrug was utilized with the introduction of Sulphonamides. The Sulphonamides are all white colored crystalline powders, it is mostly poorly soluble in water. The solubility parameters was influenced by the nature of substituent on -SO₂N₂ group. The presence of free amino acid -NH₂ is very essential for antibacterial activity. Pharmacologically all Sulphonamides shows similar activity, only they differ from each other in solubility, rate of absorption, distribution, metabolism and excretion.



Except anti-microbial and anti-bacterial activity sulphonamide moiety also shows various activities including thiazide derivatives (hydrochlorothiazide, metolazone), loop diuretics (furosemide, torsemide), sulfonyleureas (glipizide, glyburide) and some COX-2 inhibitors (celecoxib) and acetazolamide.

SAR of Sulphonamide



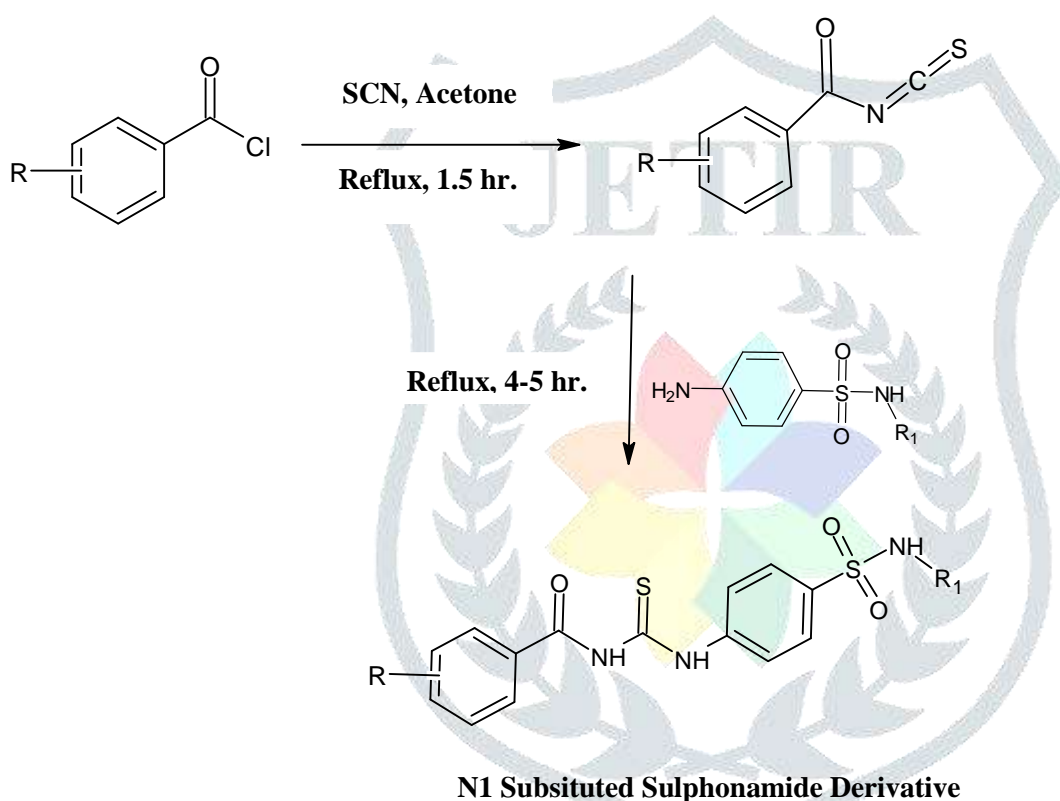
- The minimum structural requirement for the antibacterial activity is the basic Sulphonamide skeleton.
- For the activity the Amino and Sulphonyl groups are necessary and are placed in 1 and 4 position of the ring.
- The N₄ amino group can be modified into prodrug, which is then converted into free amino acids in vivo.
- Sulphur atom should be directly linked to the benzene ring.
- Replacement of benzene ring by another ring or by other substituents it decreases the activity.
- On N₁ substituted sulphonamides activity varies with the nature of substituent at the amino group with electron rich character to -SO₂ group. It increases the bacteriostatic activity.

Synthetic studies

By expending literature reviews various Sulphonamide derivatives was synthesized as follows:

Scheme- 1

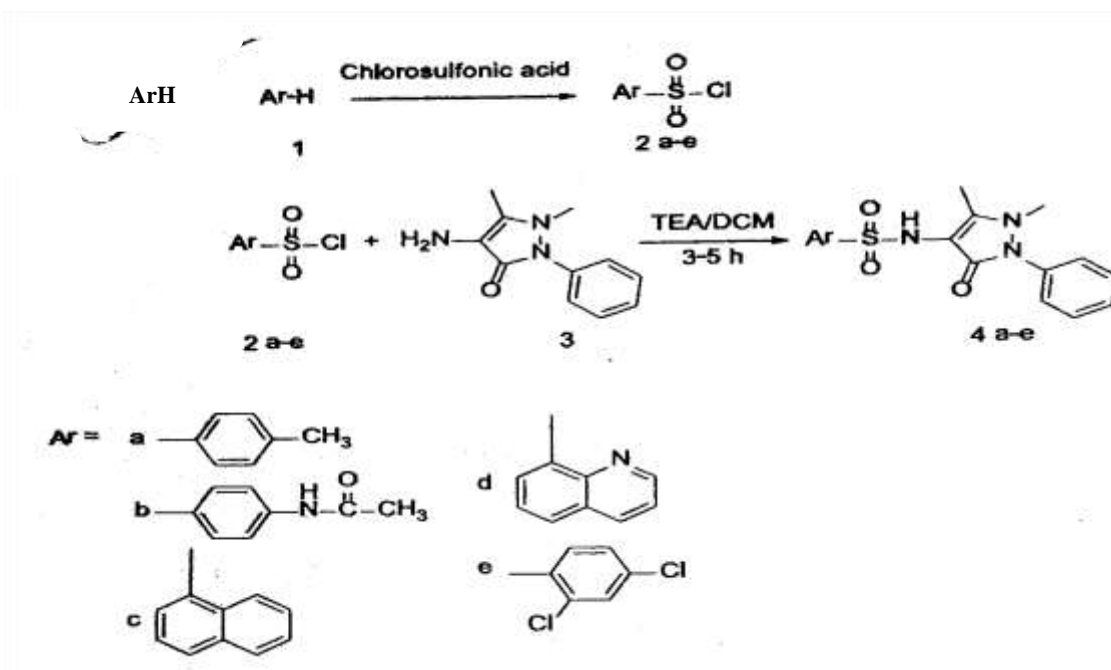
A solution of substituted benzoyl chloride (1mmol) in acetone (20ml) was added dropwise to a suspension of potassium thiocyanate (1mmol) in a acetone (30ml), and the reaction mixture was refluxed for 1.5hrs to afford substituted benzoyl isothiocyanate. The reaction was confirmed by TLC after completion of the reaction, the substituted sulphonamides (1mmol) was added and the mixture was stirred under reflux for 4-8 hrs. Upon completion of the reaction will be (checked with TLC), and resulting precipitate was collected by filtration and recrystallized from dimethyl-form amide or ethanol or water to obtained a pure product.



Scheme 1- Synthesis of Sulphonamide derivative from benzoyl chloride

Scheme- 2

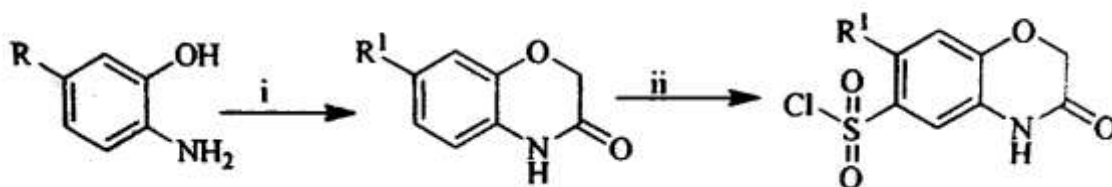
A mixture of substituted sulfonyl chlorides 2a-e (0.01), Ampyrone 3 (0.01mol) and TEA as a base in CH_2Cl_2 (20ml) was stirred at room temperature for 3-5 hrs. Further, ice-cold water was added into it and again stirred for another 20min. The product enriched with organic layer was separated and dried over Na_2SO_4 and concentrated in vacuum. The separated crude solid was dried and recrystallized in aq. Methanol to give the pure products in 70-80% yield.



Scheme 2- Synthesis of Sulphonamide derivative from Pyrazole

Scheme- 3

Treatment of substituted o-amino phenol 3a-b with chloroacetyl chloride in presence of benzyl triethyl ammonium chloride (TEBA), afforded 2H-benzo[b][1,4] oxazin-3(4H)-one 4a-b which was further subjected to chlorosulfonylation to give compounds 5a-b. The intermediate 3-oxo-3,4-dihydro-2H-benzo[b][1,4]oxazine-6-sulfonyl chloride 5a-b was attained by the two-step synthetic route.

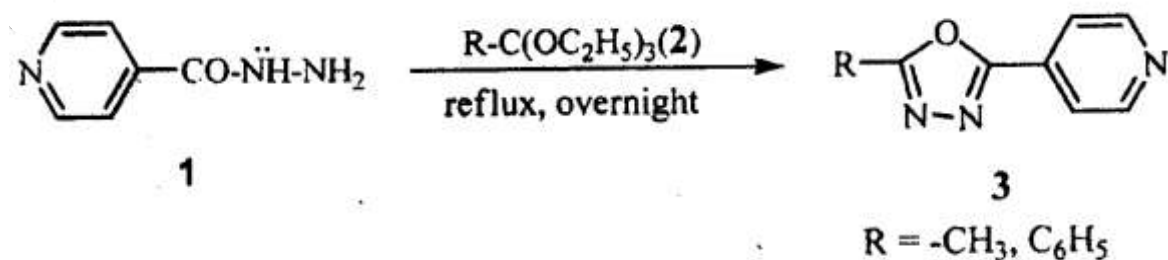


Where, R₁=H, R₁=CH₃

scheme - i) chloroacetyl chloride, TEBA, NaHCO₃, CHCl₃, 0-55°C, 67% ii) ClSO₃H, 0°C, 1h, 66%

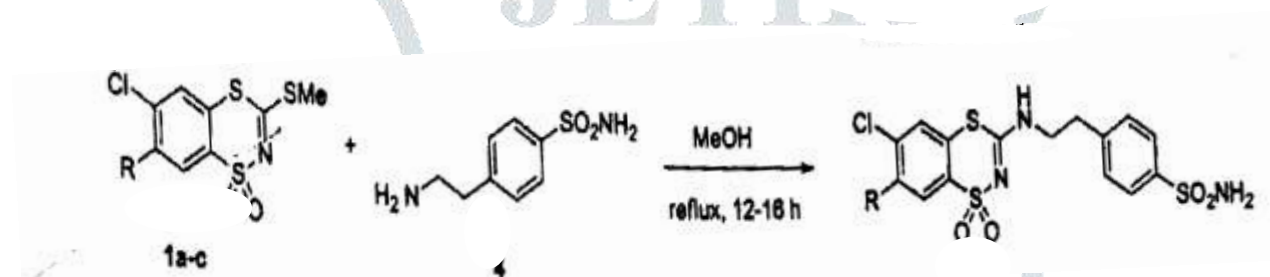
Scheme-4

Synthesis of 4-(5-methyl-1,3,4-oxadiazole-2-yl) pyridine. To a solution of isonicotinic acid hydrazide 25g (182.3mmol) in triethyl orthoacetate (135 ml) was added and reflux for 24h. The excess triethyl orthoacetate was distilled under reduced pressure, and residue was washed with cold ethanol. The residue was recrystallized from ethanol and obtained as brown crystals.



Scheme-5

A mixture of the appropriate 3- methylthio-1, 4, 2-benzodithiazine 1, 1-dioxide and 4-(2-aminoethyl) benzenesulphonamide in anhydrous methanol 40ml, was refluxed with MeSH had ceased (12-16hrs). After cooling to room temperature, the reaction mixture was left overnight. The precipitate was collected by filtration, washed with methanol (4*2ml) and dried. In this manner the benzene Sulphonamide derivatives are obtained.



CONCLUSION

In the above survey we go through the different standard exploration papers from presumed diaries. From the above survey we observed that there are different manufactured ways to deal with to incorporate the sulphonamides and subbed sulphonamides subsidiaries. The Structure movement relationship studies are audited to see what the progressions in design will mean for the pharmacological action of atoms. A portion of the critical methodologies of amalgamation are evaluated which promised us a decent high return item which shows the assorted pharmacological exercises including hostile to microbial, against disease, hostile to crazy, mitigating, stimulant exercises. Thus, it is presumed that sulphonamides are finished arrangement of medication with numerous pharmacological exercises which are shown to be helpful for improvement of humanity.

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