



STUDY OF MONTHLY VARIATIONS IN SELECTED PHYSICO-CHEMICAL PARAMETERS OF MANIKA LAKE WATER AT MUZAFFARPUR, BIHAR (INDIA)

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ABSTRACT:

Selected physico-chemical parameters of Manika Lake water were studied from January 2016-December 2016, to evaluate the purity of water of the Lake water. Here Temperature ranged between 10°C to 36°C, pH 7.5 to 8.3, Turbidity 38.0 NTU to 54.0 NTU, Alkalinity 124 mg/l to 158 mg/l, Hardness 118.0 mg/l to 158 mg/l, Dissolved Oxygen 4.3 mg/l to 8.7 mg/l, BOD 2.0 to 4.38, COD 22.6 to 48.4, Concentration of Potassium 3.2 mg/l to 4.6 mg/l, Nitrogen 0.21 mg/l to 0.41 mg/l, Phosphorus 0.14 to 0.42 mg/l and Electrical Conductivity from 248.62µS/cm to 560.36µS/cm respectively. From the data obtained here, it was noted that except few parameters such as Dissolved Oxygen was below the standard as given by WHO for fresh water during summer. Similarly, the value of Electrical Conductivity was above the given standard value from the month of June to December. Other parameters were within the limits.

KEY WORDS:

Manika Lake Water, Monthly Variations, Physico-Chemical Parameters, Purity of water, Standard Value

INTRODUCTION:

Manika Lake water is being used by the locals, for bathing, washing, fish farming and irrigation. The purity of water is day to day deteriorating because of dumping of domestic wastes both solid and liquid, direct run off of rain water from the agricultural lands, dumping of remains of worship materials and other domestic products by the villagers. The population growth, changes in life styles and negligence of this precious gift of nature, all have make the water unfit for common people for use. (Kupoluyi *et al*; 2018; Mushtaq *et al*; 2020; and Saha *et al*; 2021). The chemical, physical and biological aspects of water are known as water quality (WQ) (Khan *et al*; 2017). The biological activities results in the decomposition of litter in the aquatic systems (Kumar and Saikia, 2021). This in turn influences the water quality (WQ) mainly BOD and Dissolved Oxygen. According to Salomova (2021), the water quality is

responsible for human health and water borne diseases. Consumption of such polluted water causes several diseases to mankind (Carbal pinto *et al*; 2020), Eugwa, Ferdinand and Unachukwu (2019). Trombadore, reported that about 80% rural people in India do consume such polluted water and suffer from different water borne disease. They further reported that only diarrhoea causes six lakhs annual death in India itself. Manika Lake water is also degrading day to day due to human activities and rain water runoff during rainy season from the coastal agricultural fields.

Physico-chemical analysis of different water bodies have been done by several workers. Some of them are being mentioned here. Such as Naik and Purohit (1996); Boyd (1981); Dwivedi and Pandey (2002); Sukesh Kumar (2003); Sachidananda murthy and Yajurvedi (2004); Shah *et al* (2006); Chaurasia and Pandey (2007); Kiran (2010); Kataria *et al*; (2011); Kavita and Yadav

(2012); Nirmala *et al*; (2012); Mahajan and Tank (2013); Harney *et al*; (2013); Begum *et al*; (2014); Hawes *et al* (2014); Dixit *et al*; (2015); Karaemer *et al*; (2015); Team and Zabib (2016); Sharma *et al*; (2017); Ranjan *et al*; (2017); Ramnathan and Amsath (2018). The study of physico-chemical parameters of water is important to observe the functional relationship and productivity of aquatic ecosystem as they are regulated by the dynamics of their chemical, physical and biotic environment (Boyd, 1981). In order to evaluate the quality of water bodies and for their management and effective utilization, there are need to study the physico-chemical parameters either monthly or season wise. This will help to main them within optimum range (USEPA, 1989).

In the present study Manika Lake water samples were collected monthly. The sample have been analyzed for different physico-chemical characteristics such as temperature, pH, turbidity, alkalinity, Hardness, Dissolved Oxygen, Chemical Oxygen Demand (COD), Biological Oxygen Demand (BOD), Chloride, Potassium, Nitrate, Phosphate and Electrical Conductivity etc. in the year 2016.

MATERIALS & METHODS:

Temperature was measured with laboratory thermometer 100°C. pH was measured with the help of digital pH meter. Turbidity was measured in the laboratory with the help of turbidity meter. Dissolved oxygen was measured with Winkler's method. Here, 300 cc BOD bottle was used. Before use the bottle was washed properly and before use was rinsed with sample water at the study site. The bottle was completely submerged inside the Lake water. Inside the cork was removed and water was allowed to completely fill the bottle so that there was no air space. The cork was placed again and bottle was taken out of the Lake water. At the bank the dissolved oxygen was fixed by adding 2 ml $MnSO_4$ + 2 ml alkali iodide azide, followed by addition of 2 ml sulfuric acid. The cork was placed again and in the laboratory, the sample water was titrated with 0.025N sodium thiosulphate, using starch as an indicator. Alkalinity, hardness, BOD, COD, all was determined by titrimetric method following methodology as described in APHA (2005). Similarly, chloride was calculated with titration. Potassium was also calculated the titration. Nitrogen was calculated with the Kjeldahl method. Electrical Conductivity was measured with conductivity meter. The unit is $\mu S/cm$.

RESULTS AND DISCUSSION:

Temperature of pond water depends on the air temperature, depth of the water body and

duration of sun light falling on the surface. Due to this in the present study, a great variation in temperature ranged was noted from January to December. The highest temperature of Lake Water 36°C was noted in the month of June because this month is the hottest month of the year. Similarly, lowest temperature was 10°C in January because this is the coldest month of the year. pH of the Lake water was also the maximum 8.3 in this month, while it was lower in the month of September, that pH ranged between 7.4 to 8.3. Turbidity of the Lake water ranged between 38.0 NTU to 56.0 NTU. Here the lowest value was found in the month of October, while the highest value was noted in the month of May respectively. Alkalinity off the Lake water ranged between 124 mg/l to 158 mg/l from January to December of the study period. Here lowest value was found in January while the highest in the months of May and August respectively. Hardness of water ranged between 122.0 mg/l to 158 mg/l. Here the lowest value was noted in the month of March while the highest in the month of April respectively. The dissolved oxygen of the Lake water was also calculated. Here, the maximum value 8.7 mg/l was observed in the month of January, while the minimum value 4.3 mg/l was observed in the month of June itself. So value of Dissolved Oxygen ranged between 4.3 mg/l to 8.7 mg/l respectively Biological Oxygen Demand or BOD ranged between 2.0 mg/l to 4.38 mg/l. Here the lowest value was observed in the month of November while the highest value in February respectively.

Chemical Oxygen Demand of the Lake water was also calculated. It ranged between 22.6 mg/l to 48.4 mg/l from January to December in the study year. Concentration of potassium in the Lake water was also calculated which ranged between 3.2 mg/l to 4.6 mg/l respectively. The value of nitrate in the pond ware ranged between 0.21 mg/l to 0.41 mg/l receptively from January to December. Similarly, concentration of phosphorous ranged between 0.14 mg/l to 0.42 mg/l from January to December of the study year. The Electrical Conductivity of Lake water ranged between 248.62 $\mu S/cm$ to 560.36 $\mu S/cm$ respectively from January to December. So from the table-1, it may be noted that all the parameters studied here varied considerably from January to December of the study year.

DISCUSSION:

The value of surface water temperature is directly related with the value of air temperature. Water temperature influences the physiology of the organism living in the water body. It influences the

biochemical reactions by influencing the enzyme activities. High temperature may increase the toxicity of many substances such as heavy metals in water for domestic use. The value obtained in this study was higher than that of the recommended WHO (2006) standard of 25°C from March to October of the study year. High water temperature enhances the growth of microorganisms and cause changes in taste, odor, colour and concentration of dissolved oxygen. This may also reduce the efficiency of settling of solids that the waste may contain, because of the resistance that high viscosity offers to the downward motion of the particles as they settle. (Jayalakshami *et al*; 2011; Team and Zebib, 2016). pH of the Lake water was also observed. It was mostly alkaline and highest pH was in the month of January that was 8.3. Present findings are in agreement with the findings of Wani *et al*; (1990); Nandan and Patel (1992); Singh *et al*; (2023). Turbidity in water refers to the cloudiness caused by the suspended particles. It is optical characteristic off water. It is a measure of the concentrations of the smallest particles that do not settle. It may also be caused by phytoplanktons consumption of turbid water may cause different diseases. Even fishes are affected with high turbidity. In the present findings high turbidity was noted in the month of June that was 56 NTU.

Alkalinity is a measure of the capacity of water to neutralize acids. Bicarbonates, (Baking Soda) Carbonates and hydroxides remove H⁺ ions and lower the acidity of water, and thus pH is increased. In the present study the alkalinity ranged between 124 mg/l to 158 mg/l. Present findings are in agreement with the findings of Engwa *et al*; (2019); Bhat and Pandit (2020); Dey *et al*; (2021).

Hardness refers to the concentration of dissolved minerals, primarily calcium and magnesium in water. Hard water is high in dissolved minerals. Here bicarbonates, chlorides and sulfates of calcium and magnesium are important. In the present study, monthly variations in hardness of Manika Lake water was determined which ranged between 136 mg/l to 158 mg/l respectively. Maximum value was observed during summer. Present findings therefore, corroborate with the findings of Kumar (1995); Nair (2002); Hulyal and Kaliwal (2011) and Singh *et al*; (2023). Dissolved Oxygen is the free oxygen present in water. It may be absorbed from the atmosphere or be released by the photosynthetic organisms in the aquatic ecosystem. Because temperature influences the capability of oxygen to be dissolved in water therefore, during summer it was only 4.3 mg/l while in winter it was 8.7 mg/l. 4.3 mg/l is below the standard fixed by WHO. Present findings are

supported by the findings of Halim *et al*; (2018); Rani (2019); Mushtaq *et al*; (2020); Salomova (2021) and Prasanjit *et al*; (2022). The Bio Chemical Oxygen Demand of the Lake water might alter the oxygen cycle and the oxygen balance. This BOD increase may cause depletion of oxygen in the water bodies. BOD observed was lower during low oxygen contents in water. Similar findings have been reported by Harney *et al*; (2013). Hawes *et al*; (2014); Chemical Oxygen Demand (COD) indicates the presence pollutants which are being oxidized chemically. It also requires oxygen for the reactions. Here it varies between 22.6 mg/l to 48.4 mg/l. Liu *et al*; (2020) reported that COD is a good indicator of water bodies. Value of COD is directly related with the concentration of soluble organic compounds. In general COD is generally higher than BOD as more organic compounds are oxidized chemically than biologically.

Concentration of potassium was also studied in the study period. Its concentration ranged between 3.2 mg/l to 4.6 mg/l from January to December. The amount of pond water was under the standard as is recommended for the fish pond (Kavita and Sheela, 2012). Nitrate concentration in the present work also varied from January to December. The concentration of nitrate ranged between 0.21 to 0.41 mg/l Here lower concentration was noted in the month of December and January. All organisms require nitrogen for the basic process of life. Presence of nitrates in the water sample is suggestive of some bacterial action and bacterial growth. Above findings are supported by the findings of Majumdar *et al*; (2006); Kavita and Sheela (2012). The concentration of phosphorus was also studied in different months of the year. Here the concentration varied from 0.14 mg/l to 0.26 mg/l respectively. Chemical fertilizers are the main source of phosphorous in aquatic ecosystem as the residues are brought here along with the rain water runoff during the rainy season. It is reported that phosphate level over 0.4 to 1.5 mg/l are not suitable for a fish pond. They can promote growth of different weeds. (Hepher and Pruginin, 1981, Mukherjee *et al*; 2022). Electrical Conductivity of Manika Lake water ranged between 248.62µS/cm to 560.36µS/cm respectively. Electrical conductivity is the ability of an aqueous solution to conduct electric current. It is a useful tool to evaluate the purity of water (Acharya *et al*; 2008). It is due to abundant dissolved salts, which comes in the water bodies through different source. Variations in the value of Electrical Conductivity have been reported by Kataria *et al*; (2011) and Srivastava and Kanungo

(2013), Dixit *et al*; (2015). Present observation is thus in agreement with the above workers.

CONCLUSION:

Monitoring of health of the Manika Lake water is essential as the water is being used by the people near it. Physico-chemical analysis of different parameters will suggest about the health of water. Some of the parameters can be managed by the administration body asking the locals not to dump their domestic wastes may be solid or liquid. The value of dissolved oxygen will suggest the suitability of Lake water for fisheries. Similarly,

BOD and COD are of great importance. Therefore, regular physico-chemical analysis of Lake water is essential.

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TABLE:

Showing monthly variations in selected Physico-chemical parameters of Manika Lake water at Muzaffarpur.

Months	Jan	Feb	Mar	Apr	May	Jun	Jul	Aug	Sep	Oct	Nov	Dec
Temp. (°C)	10.0	18.0	26.0	29.0	33.0	36.0	34.0	30.0	28.0	27.0	23.0	14.0
pH	8.3	8.1	7.6	7.8	8.0	7.6	8.0	7.8	7.4	7.6	7.54	7.7
Turbidity (NTU)	52.0	53.0	46.0	49.0	56.0	54.0	47.0	46.0	42.0	38.0	40.0	47.0
Alkalinity (mg/l)	124.0	143.0	146.0	149.0	158.0	148.0	132.0	158.0	152.0	134.0	138.0	132.0
Hardness (mg/l)	136.0	118.0	122.0	152.0	154.0	146.0	158.0	144.0	131.0	138.0	148.0	139.0
Dissolved Oxygen	8.7	8.0	7.8	6.5	5.2	4.3	5.9	6.4	6.87	7.3	7.7	7.9
Biological oxygen Demand	3.45	4.38	3.50	3.72	3.18	3.14	3.12	3.26	2.65	2.70	2.0	3.40
Chemical Oxygen Demand	35.2	48.4	32.5	22.6	27.5	29.3	22.8	23.2	24.6	24.5	25.7	29.8
Potassium (mg/l)	3.6	4.3	4.6	4.4	4.3	3.2	3.4	3.7	3.8	3.7	3.2	3.4
Nitrate (mg/l)	0.22	0.35	0.32	0.41	0.32	0.30	0.26	0.22	0.21	0.23	0.22	0.21
Phosphate (mg/l)	0.14	0.26	0.32	0.21	0.15	0.14	0.19	0.26	0.22	0.42	0.20	0.24
EC (µS/cm)	270.6	278.4	250.15	248.6	280.1	315.4	455.2	510.1	560.3	470.1	405.2	310.6
	2	5	15	2	8	4	0	8	6	5	4	2

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