JETIR.ORG ISSN: 2349-5162 | ESTD Year : 2014 | Monthly Issue JURNAL OF EMERGING TECHNOLOGIES AND INNOVATIVE RESEARCH (JETIR)

An International Scholarly Open Access, Peer-reviewed, Refereed Journal

Development and Validation of a Stability Indicating HPTLC Method for the Determination of Prucalopride succinate in Bulk and Tablet dosage form

Ram S. Sakhare^{1*}, Preeti G. Wangarwar¹, Mukesh H. Muratkar¹, Sunil S. Hindole¹ ^{*1}Channabasweshwar Pharmacy College (Degree), Latur-413512

ABSTRACT:

A simple, precise, and accurate stability-indicating high-performance thin layer chromatographic method for analysis of Prucalopride succinate It has been demonstrated and proven that in the presence of its degradation products. Optimum separation among Prucalopride succinate and the results of its degradation was achieved by use of silica gel 60F254 as stationary phase with Toluene: Methanol: Glacial acetic acid (7:2.5:0.5 v/v/v) as a mobile phase. Densitometric quantification was performed at 240 nm. The retention factor was revealed to be 0.25 ± 0.10 for parent drug. The drug was subjected to different stress conditions like acid, base hydrolysis, oxidation, thermal degradation and photolysis. In accordance with ICH Q2 (R1) guidelines, the approach was successfully validated. The method exhibited an adequate linearity (r =0.995), selectivity, precision and accuracy. Prucalopride succinate in bulk and pharmaceutical dose form can be routinely analysed using the developed method.

Keywords: Prucalopride succinate, Stability indicating method, ICH, HPTLC, Validation.

INTRODUCTION

Prucalopride (PRU) (Fig. 1) chemically is 4-amino-5-chloro-2, 3-dihydro-N-[1-(3-methoxy propyl)-4piperidinyl]-7-benzofurancarboxamide butanedioate). It is a dihydro benzofuran carboxamide derivative from the benzofuran family, and it possesses enterokinetic activity by specifically stimulating 5-HT4 receptors^{[1].} It performs a specific action on the stomach muscle⁻ Wall thus, helping to reinstate the regular working of the human bowel^{[2].} There was an enhancement in the bowel motion frequency but no considerable consequence on the transit time of the colon^{[3].} The literature review reveals that few analytical methods have been reported for its quantitative estimation in pharmaceutical formulations, which include UV-spectrophotometry, HPLC, Stability indicating HPLC and UHPLC-MS-MS^[4-15]. Stability indicating HPTLC method is not reported. Consequently, the objective of the current work was to develop and validate a novel HPTLC method for the measurement of prucalopride succinate in pharmaceutical formulation that was simple, quick, selective, economical, and stability suggesting.

Figure 1 Structure of Prucalopride succinate.

MATERIALS AND METHODS

Chemicals and Reagent

Pure Prucalopride succinate was generously gifted by Alkem laboratories (Mumbai, India). The pharmaceutical dosage form used in this study was Prudac 1 (Zydus Cadila, India) Procured from the local market and labeled to contain1mg of prucalopride succinate per tablet. The solvents and chemicals used in the study were of AR grade. (Merck specialties Pvt. Mumbai, India-based Ltd.).

Chromatographic and Instrumentation Conditions

Chromatographic separation of the drug was performed on Merck TLC plates precoated with silica gel $60F_{254}$ (10×10) from E. Merck, applying with a CAMAG Linomat 5 sample applicator (Switzerland).Weighing of the chemicals was done using Shimadzu Corporation Japan, BL-220H balance. Transonic Digital S sonicator (Athena Technology, ATS-6-LCD) was utilized for the sonication. At 240 nm, the CAMAG thin layer chromatography scanner was used for densitometric scanning. operated by win CATS software version 1.4.2. The photostability chamber (Thermolab Scientific Equipment, TS0000200G) was used for the forced degradation study.

Selection of solvent

After trial for checking solubility in different solvents, the drug solubility found in methanol so, analytical grade methanol was used.

Selection of detection wavelength

Further dilutions from the standard stock solution (100 ng/l) were produced using methanol and scanned between 200 and 400 nm to get spectra. It was found that the drug displayed a sizable amount of absorbance at 240 nm. This is a representation of the Prucalopride UV spectrum.

Figure: 2 UV Spectrum of prucalopride succinate

Preparation of Standard Solution

Standard stock solution was prepared by dissolving 1 mg of prucalopride succinate drug in 10 mL of methanol to get working standard solution of concentration 100 ng/ μ l from which 5 ml was further diluted to 10 ml to get solution of 50ng/ μ l.



Figure 2 Densitogram of standard solution of Prucalopride succinate (150ngband-1,Rf= 0.25±0.10)

Preparation of Sample Solution

Tablet powder equivalent to 1 mg was transferred to 100 mL volumetric flask containing 5 mL of methanol and the contents were sonicated for 15 min and the resulting sample stock solution was filtered the volume was made up to the mark with methanol to obtain the final concentration of 50 ng/band.

Chromatographic conditions: The mobile phase optimization trials were performed using Toluene: Methanol, Benzene: Methanol, Ethyl acetate: Methanol in the various ratios. Hence the optimized mobile phase was Toluene: Methanol: Glacial acetic acid (7: 2.5: 0.5 v/v/v). The stationary phase used as TLC plates precoated with silica gel 60 F254 having dimension of 10 cm ×10 cm with 250 µm layer thickness. The standard solution of Prucalopride succinate was spotted on the dried, pre-coated TLC plate having band with 6 mm width. Densitometric scanning was performed on Camag thin layer chromatography scanner at 240 nm. The retention factor of prucalopride succinate was 0.25 ± 0.10 .

Stress degradation study:

To evaluate the stability indicating studies of the developed HPTLC method, forced degradation studies were carried out in accordance to the ICH recommendations. The standard drugs were exposed to an acid, a base, oxidation, wet heat, dry heat and photo-degradation studies. It was done to conduct the study at concentration of $100\mu g/\mu l^{-1}$.

© 2023 JETIR April 2023, Volume 10, Issue 4

Acid hydrolysis:

5 ml of the above standard drug solution was mixed 1 ml of 0.1 N hydrochloric acid and the volume was made with solvent. Solution was refluxed at 60°C for 30 min. The densitogram of the acid degraded sample showed additional peak at Rf value 0.62 with 19.04 % degradation.



Figure 3 Densitogram after acid hydrolysis with degradation product (D1,Rf=0.62)

Alkaline hydrolysis:

5 ml of the above standard drug solution was mixed 1 ml of 0.1 N sodium hydroxide and the volume was made with solvent. Solution was refluxed at 60°C for 30 min. The drug was revealed to be liable to base hydrolysis with 17.18% degradation. The densitogram had degradation product peak at Rf value of 0.40.





Neutral degradation:

5 ml of the above methanolic drug solution was mixed 1 ml of water was added and refluxed at 60°C for 30 min. About 12.10% degradation was observed for the drug under neutral hydrolytic degradation. There was decrease in peak region of drug as compared to initial area and no additional peak for degradation was seen.





Oxidative degradation:

Oxidative degradation was performed by mixing 5 ml of the above methanolic drug solution with 1 ml of 6 % hydrogen peroxide and refluxed at 60°C for 30 min. The peak area of drug was reduced as compared to the initial area indicating that the drug undergoes degradation in oxidative stress condition. 20.65 % degradation was observed with single degradation product at Rf value 0.48.





Thermal degradation

The reduction in the peak area of drug as compared to the initial area indicated 7.90% degradation but no additional degradation peak was observed. For a sample heated by dry heat, a representative densitogram was developed.



Figure 7 Densitogram obtained after dry heat degradation at70°C for 6 h

Degradation under dry heat:

The study was undertaken by placing drug powder in oven at 70°C for 6 h. A sample was removed from oven at suitable times and dissolved in methanol to obtain solution of 50 ng μ l⁻¹.

Photo-degradation:

The Prucalopride standard, 10 mg was exposed to UV light in a UV chamber and sunlight for 24 hours and appropriate dilutions were made in methanol to obtain final concentration of 50ng μ l⁻¹followed by 4 μ L

Figure 8 Densitogram obtained after exposure of drug to UV light

application, development and scanning under ideal chromatographic circumstances



Marked degradation in the densitogram was observed but the deteriorated products were well resolved from the drug indicating specificity of the method.

Results of stress degradation study of Prucalopride succinate

Stress conditions /duration	%Recovered	%Degradation
Acidic/0.1NHCl/Refluxedat60°Cfor 30min	80.96	19.04
Alkaline/0.1NNaOH/Refluxedat60°C for 30min	82.81	17.18
Oxidative/6%H ₂ O ₂ /Refluxedat60°Cfor 30min	79.35	20.65
Neutral/H ₂ O/Refluxat60°for30min	87.89	12.10
Photolysis: UV light 200 watthsquaremeter ⁻¹ 7days	99.09	
Dry heat /70°C/6h	92.09	07.90

Table:1 Summary of forced degradation studies

Validation of Method

The method was validated in compliance with ICH guidelines.

Linearity and Range:

Volumes 1, 2, 3, 4 and 5 μ l from standard drug solution (50 ng/ μ L⁻¹) were applied on the TLC plates with sample applicator in nitrogen stream. After spotting, plates were dried. Regression data obtained from calibration curve demonstrated excellent linear relationship over 50-250 ng band-1 concentration range.



Fig–3 Linearity curve of prucalopride succinate

Straight-line calibration curves were achieved in the 50-250 ng band⁻¹ concentration range through high correlation coefficient. Excellent correlation exists between peak area and concentration of drug within the JETIRFW06041 Journal of Emerging Technologies and Innovative Research (JETIR) www.jetir.org 321

concentration range indicated above. Correlation coefficient closer to 1 for the drug also proved the linearity.

Concentration	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Avg.
(ng/band ⁻¹)							Area*
50	1968	1924	2088	1994	2015	1932	1987
100	3441	3417	3554	3459	3513	3369	3459
150	5193	5104	5339	5113	5219	5027	5165
200	7336	7254	7459	7245	7369	7178	7306
250	8748	8662	8759	8703	8727	8573	8695

Table: 2 Linearity of Prucalopride Succinate.

Precision

Intra and inter day deviations in results were examined by recording the values of peak area after application, three distinct concentrations in linearity range. The lesser %RSD values (<2) obtained indicate the precision of the developed method.

1. Intra-day precision

It was determined by analyzing prucalopride standard solutions at three different concentrations in linearity range for thrice on the corresponding day. Each concentration was spotted in triplicate and % R.S.D. was determined. The %RSD value for Intra-day precision were between 0.82 to 1.03

Concentration applied (ng/band ⁻¹)	Average area	%RSD
100	3580	0.82
150	5295	1.03
200	7080	0.84

Table: 3 Intra-day precision

2. Inter-day precision

Standard drug solutions at three different concentrations on different three days during the course of a week were analyzed and % R.S.D. was calculated. The %RSD value for Inter-day precision were in the range of 0.79 to1.09

Concentration applied (ng band ⁻¹)	Average Area	%R.S.D.*
100	3586	1.09
150	5305	0.79
200	7061	0.84

Table: 4 Inter-day precision

Detection and Quantitation limit

From the linearity data the Limit of quantification and detection was calculated, using the formula LOD= $3.3 \sigma/S$ and LOQ= $10 \sigma/S$ where σ represents standard deviation of response i.e., y-intercept while S denotes the calibration's slope graph. The LOD & LOQ revealed to be 6.85 ng/band⁻¹ and 20.77ng/band⁻¹ respectively.

Specificity

The excipients from formulation did not show any interference with analyte peak and obtained peak purity value was in the limit representing method specificity. The non-interference of degradation product peaks with the drug peak also proved that method is specific.



Figure 9 Densitogram of marketed formulation with concentration 100ng/band-1 (Rf= 0.25 ± 0.08)

Assay

Under the ideal chromatographic conditions, 2 ml of the drug from the sample stock solution was applied to HPTLC plates and examined. 99.99% of the medication was found to be present.

Spotted concentration (ng/band ⁻¹)	%Recovery	Std. Dev.	%R.S.D.*
100	99.99	0.91	0.91

Table: 5 assay of prucalopride succinate

Accuracy

Recovery studies were undertaken to check method accuracy through addition of standard drug to pre-analyzed sample at three different levels (80, 100 and 120 %). The sample concentration selected was 100 ng band⁻¹ from tablet solution. The % average recovery was 100.28 ± 0.73 which indicated accurateness of developed method for determining drug dosage in tablet form.

Drug	Amount taken (ng/band ⁻¹)	Amount added (ng/band ⁻¹)	Amount found (ng/band ⁻¹)	%Recovery ± RSD
	100	80	180.29	100.15±0.82
Prucalopride succinate	100	100	199.82	99.90±0.67
	100	120	221.79	100.81±0.70

Table: 6 Results of recovery studies

Robustness

By performing the analysis, the robustness of the method was determined under conditions during which mobile phase composition (± 2 % methanol), wavelength (± 1 nm) was altered and the effect on the area of drug was noted. The study was performed at concentration level of 250 ng/band ¹.

Parameters	%R.S.D.*
Composition of mobile phase (±2%methanol)	1.01
Wavelength (±1 nm)	1.30

Table: 7 Robustness date

The overall review of validation parameters is given in below table:

Table: 8 Summary of validation parameters

Sr. No.	Parameters	Results
1.	Linearity(ngband ⁻¹)	50-250
2.	Correlation coefficient(r)	0.995
3.	Detection limit(ngband ⁻¹)	6.85
4.	Quantitation limit (ngband ⁻¹)	20.77
5.	Accuracy	100.28±0.73
	Precision(%R.S.D.)	
6	Intraday precision	0.82-1.03
	Interday precision	0.79-1.09
7.	Robustness	Robust

DISCUSSION

The development and validation of the prucalopride succinate medication in bulk and medicinal dose form is the purpose of the HPTLC approach. The absorption spectra for prucalopride succinate were recorded in the wavelength region of 200-400 nm and it determined to be 240nm. The accuracy of the method was confirmed by adding known amount of the pure drug to the formulation previously analyzed by this method and the analytical data was reported. The development and validation method and the stability studies of drug by HPTLC is important for selecting formulation and storage conditions for drug and drug products. Retention time prucalopride succinate 0.25 ± 0.10 was found shown in figure 6. The proposed method's intraday and interday precision was validated. The %RSD found less than 2.

CONCLUSION

For the estimation of prucalopride succinate in tablet dosage form, a stable HPTLC technique has been developed and validated that does not depend on excipients or degradant products under many circumstances, including photolysis, thermal oxidation, hydrolysis, and oxidation. The outcomes showed that the technique is very specific and that the drug and its degradation compounds were successfully separated.

REFERENCES

1. Keating GM. Prucalopride: A review of its use in the management of chronic constipation. Drugs. 2013;

73(17):1935-1950.

- Dongen V. effect of Prucalopride, a new enterokinetic agent, on gastrointestinal transit and anorectal function in healthy volunters. Alim Pharmacol Therapeut. 1999; 13(11):1493-1497.
- 3. Tack J, Stanghellini V, Dubois D, Joseph A, Vandeplassche L, Kerstens R. effect of Prucalopride on symptoms of Chronic Constipation. NeurogastroenterolMotil. 2014;26(1):21-27.
- Saranjit singh. Stability testing during product development in Jain NK pharmaceutical product development, CBS publisher and distributors, India, 2006, 272-293.
- Dr. G. Abrami, Dr. T. Vetrichelvan, Method development and validation of Prucalopride Succinate in bulk and tablet dosage form by RP-HPLC Method. Journal of innovations in pharmaceutical and biological sciences (JIPBS) ISSN: 2349-2759, vol-7, page no. 07-12, 2020.
- A.N. Campbell and J. Sherma, Development and Validation of High Performance Thin-Layer Chromatographic Method With Densitometric Detection For Determination Of Biscodyl In Pharmaceutical Tablets, Acta Chromatographica No.13, vol-3, page no.109-116, 2003.
- MJ. Ansari, S. Ahamad, K. Kohli, J. Ali, R.K. Khar, stability indicating HPTLC determination of curcumin in bulk drug and pharmaceutical formulations, journal of pharmaceutical and biomedical analysis, 132-138, 2005.
- Patel SK, development and validation of analytical method for estimation of Leflunomide in bulk and their pharmaceutical dosage Form, Austin Journal of Analytical and pharmaceutical chemistry, ISSN: 2381-8913, Volume 2 Issue 4 - 2015, Page No 01-10.
- 9. Emmanuel AV, Roy AJ, Nicholls TJ, Kamm MA. Prucalopride, a Systemic Enterokinetic, for the Treatment of Constipation. Alim Pharmacol Therapeut. 2002;16(7):1347-1356.
- 10. Zhi Sun1, Lihua Zuo1, Jian Kang, Lin Zhou, Mengmeng Jia, Zeyun Li, Zhiheng Yang, Xiaojian Zhang, Zhenfeng Zhu, development and validation of a sensitive UHPLC– MS/MS method for Quantitation of Prucalopride in Rat Plasma and its application to pharmacokinetics study, Page No. 328-333, 2016.
- 11. Baira Shandilya, Mahamuni Anupama, Jajula Atul, Awasthi Pradipbhai D. Kalariya M.V.N. Kumar Talluri, selective separation and characterisation of stress degradation products and process impurities of prucalopride succinate by LCQTOF-MS/MS 22, 2016.
- 12. ViragGophane, Ravi A Thakur, Development, Validation And Stability Indicating RP-HPLC Method for estimation of prucalopride Formulation, 2016.
- Dr. G. Abrami, Dr. T. Vetrichelvan, Method development and validation of Prucalopride Succinate in bulk and tablet dosage form by RP-HPLC Method. Journal of innovations in Pharmaceutical and biological sciences (JIPBS) ISSN: 2349-2759, vol-7, page no. 07-12, 2020.
- 14. Sangameshwar B. Kanthale, Sanjay S. Thonte, Sanjay S. Pekamwar, Debarshi K. Mahapatra development and validation of a Stability Indicating RP-HPLC Method for the Determination of Prucalopride succinate in Bulk and Tablet, International Journal of Pharmaceutical Sciences and Drug Research vol-12(2), 2020, Page No.166-174
- 15. Goutham Dev, Ashish Bojja and MukthinuthalapatiMathrusri Annapurna, development and validation of new analytical methods for the quantification of Prucalopride Succinate, Acta Scientific Pharmaceutical Sciences,

Volume 4, Issue 5, May 2020.74-77

- 16. Dr. G. Abirami, Dr. T. Vetrichelvan, M. Raman, Method development and validation of Prucalopride succinate in bulk and tablet dosage form by RP-HPLC method, Journal of Innovations in Pharmaceutical and Biological Sciences Vol. 7 (3): 07-12, Jul-Sep, 2020.
- 17. Vaibhavi N. Akhani, Mrs. Khushbu K. Patel, Ms. K. S. Patel, Dr. L.M. Prajapati and Dr. C. N. Patel, Development and validation of RP-HPLC method for estimation of prucalopride succinate in pharmaceutical dosage formworld Journal of pharmacy and pharmaceutical sciences, Volume 9, Issue 6, 1112-1122.
- Ashwini S. Chawathe , Purnima D. Hamrapurkar, Implementation of Quality by Design Approach for Analytical method development and validation for estimation of prucalopride succinate in the Bulk and Solid Dosage Form, IJPQA, Volume 11 Issue 4 October 2020 – December 2020, 510-517.
- R. S. Sakhare, S. S. Pekamwar, T. V. Gitte, Stability indicating high performance thin-layer chromatography method for simultaneous estimation of ambroxol hydrochloride and loratadine in pharmaceutical dosage form, Indian Drug, August 2018, 55(8); 44-51.

