



Synthesis & screening for antimicrobial activity of N-Glucosylated aryl substituted s-benzyl isothiocarbamide

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Abstract

Carbohydrates, “The building blocks of life” comprises the most abundant group of natural product. They are prime biological substances, which are metabolized as monosaccharides, oligo-saccharides and polysaccharides. They are indispensable for living organism, serving as skeletal structure in plants and also in insects. They also occur as food reserves in the storage organs of plants and in the liver and muscles of animals. In addition, they are an important source of energy required for the various metabolic source of energy required for the various metabolic activities of the living organism. The energy being derived as a result of their oxidation. They also serve to lubricate skeletal joint, to provide adhesion between cells and they confer biological specificity on the surface of animal cell.

Series of new 1-Tetra-O-Benzoyl-β-D-Glucosyl-3-aryl (nitro substituted aniline)-2-phenyl thiocarbamide 2-S-benzyl-isothiocarbamide was prepared by the interaction of the of 1-Tetra-O-Benzoyl- β -D Glucosyl-3-aryl (nitro substituted aniline)-2-S-benzyl-isothiocarbamide and Phenyl thiocarbamide in benzene medium. The Reaction was refluxed for 3hr in benzene medium. After completion of the reaction, the reaction mixture was brought to room temperature and the solvent removed under reduced pressure to obtain residue. This residue was triturated several times with petroleum ether (60-80°C) to afford a pale yellow solid. Product was purified from chloroform-petroleum ether. The newly synthesized compounds have been characterized by analytical and IR, ¹H NMR and Mass spectral studies. These compounds were screened for their antibacterial activities against *Escherichia coli*, *Staphylococcus aureus*. These compounds show appreciable activity towards these microorganisms.

Keyword: Phenyl Thiocarbamide, substituted s-Benzyl isothiocarbamide, Phenyl isothiocyanate, and Biological studies.

1. Introduction:

Carbohydrate derivatives have been extensively investigated including synthesis, characterization and biological activity. Partly due to the facts that many natural occurring saccharides and synthesized analogues exhibit various and potent biological activities and they have been widely employed as agrochemicals and pharmaceuticals¹⁻⁵.

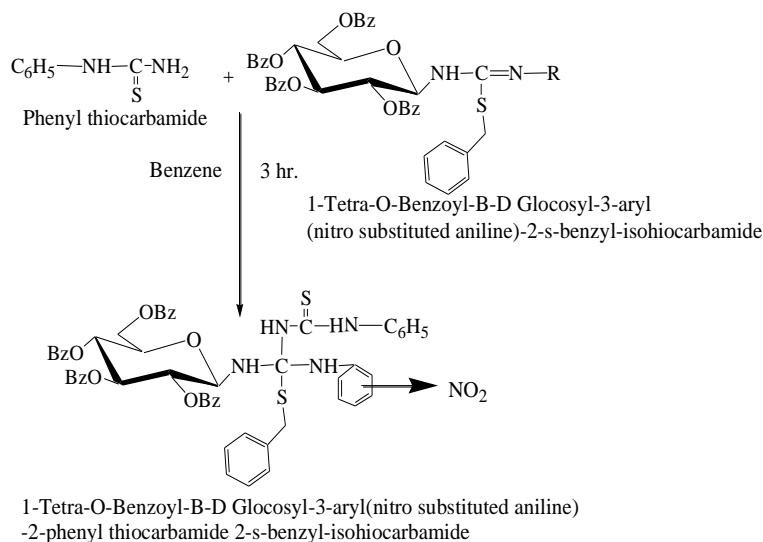
Sugar isothiocyanate and their thiourea and thiocarbamide derivatives exhibits wide range of pharmacological activities⁶⁻⁹ like antimicrobial, antiviral and antitumor. Isothiocyanates are important intermediates belonging to the family of compounds known as heterocumulenes. Isothiocyanates are versatile synthetic intermediates in organic chemistry due to their availability and their tendency to undergo nucleophilic addition and cycloadditions. Thiourea and its derivatives are a group of compounds possessing a wide spectrum of biological activities such as anticonvulsant, herbicidal and it is versatile reagent in organic synthesis. Also thiomaltosides are an important constitute of carbohydrate chemistry.

2. Experimental

All chemicals were research grade. Melting points determined are uncorrected. IR spectra were recorded in KBr on a FT-IR Perkin-Elmer RXI (4000-450cm⁻¹) spectrophotometer. ¹H NMR measurements were performed on a Bruker DRX-300 (300 MHz FT NMR) NMR spectrometer in CDCl₃ solution with TMS as internal reference. The Mass spectra were recorded on a THERMO Finnigan LCQ Advantage max ion trap Mass spectrometer. Thin layer chromatography (TLC) was performed on silica Gel G and spots were visualized by iodine vapour. The compounds describe in this paper were first time synthesized by the multistep reaction protocol.

Preparation 1-Tetra-O-Benzoyl- β -D Glucosyl-3-aryl (nitro substituted aniline)-2-phenyl thiocarbamides, 2-S-benzyl-isothiocarbamide

1-Tetra-O-Benzoyl-β-D Glucosyl-3-aryl (nitro substituted aniline)-2-phenyl thiocarbamides, 2-S-benzyl-isothiocarbamide was prepared by the interaction of the 1- tetra-O-benzoyl-β-D-glucosyl -3 aryl (nitro substituted aniline)-2-S-benzyl-isothiocarbamide and Phenyl thiocarbamide in benzene medium. The Reaction was refluxed for 3hr in benzene and then the benzene was evaporated. The formed product is washed and recrystallised by the petroleum ether (60-80°C).



Where, R= (a) Phenyl, (b) o-Nitro-aniline, (c) m- Nitro -aniline, (d) p- Nitro -aniline,

Results and discussion

Herein, we report the synthesis of various 1-Tetra-O-Benzoyl- β -D Glucosyl-3-aryl (nitro substituted aniline)-2-phenyl thiocarbamides, 2-S-benzyl-isohiocarbamide (**1-4**) by interaction of Phenyl Thiocarbamide (**1**) and 1-tetra-O-benzoyl- β -D-glucosyl-3-aryl (nitro substituted aniline)-2-S-benzyl-isohiocarbamide (**1-4**) in benzene medium. All products were crystallized from ethanol before recording the physical data (Table-1). The purity of compounds was checked by TLC. The spectral analysis¹⁰⁻¹² IR, ¹H NMR and Mass spectra of the product were observed.

1: IR (KBr): ν 3050 (Ar-H), 2779 (Alk C-H), 1728 (C=O), 1449 (C=N), 1250(C=S), 1100 (Characteristic of glucose), 652 (C-S), ¹H NMR (δ in ppm, CDCl₃): δ 7.4-6.4 (m, aromatic protons), δ 6.2 (s, N-H), δ 5.2-3.6 (m, glucosyl protons), δ 2.12-1.38 (Methyl protons) Mass (m/z): 994 (M⁺), 579, 420, 105; Anal. Calcd for C₅₆H₅₁N₄S₂O₉: C, 68.08; H, 5.16; N, 5.67; S, 6.48; Found: C, 68.12; H, 5.28; N, 5.72; S, 6.50.

On the basis of all above facts the product with m. p. 140°C was assigned the structure 1-tetra-O-Benzoyl- β -D Glucosyl-3-phenyl-2-phenyl-thiocarbamide, 2-S-benzyl-isohiocarbamide . When the reaction of phenyl thiocarbamide was extended to several other 1-tetra-O-benzoyl- β -D-glucosyl-3-aryl-2-S-benzyl-isohiocarbamide corresponding 1-tetra-O-Benzoyl- β -D Glucosyl-3-aryl-2-phenyl-thiocarbamide, 2-S-benzyl-isohiocarbamide has been synthesized.

2: IR (KBr): ν 30789 (Ar-H), 2875 (Alk C-H), 1728 (C=O), 1449 (C=N), 1229 (C-O), 1250 (C=S), 1100 (Characteristic of glucose), 652 (C-S), ¹H NMR (δ in ppm, CDCl₃): δ 7.4-6.4 (m, aromatic protons), δ 6.2 (s, N-H), δ 5.2-3.6 (m, glucosyl protons), δ 2.12-1.38 (Methyl protons) Mass (m/z): 1028 (M⁺), 579, 420, 150; Anal. Calcd for C₅₆H₅₀N₅S₂O₁₁: C, 65.11; H, 4.84; N, 6.78; S, 6.20; Found: C, 65.22; H, 4.92; N, 6.70; S, 6.18.

On the basis of all above facts the product with m. p. 145°C was assigned the structure 1-tetra-O-Benzoyl- β -D Glucosyl-3-o-nitro-phenyl-2-phenyl-thiocarbamide, 2-S-benzyl-isohiocarbamide .

Table -1: Physical data for characterization of compounds (1-4)

Compd	Yield %	R _f	M.P. °C	Analysis (%): Found (calcd)	
				N	S
a	70%	0.69	140 ⁰ C	5.72(5.67)	6.50(6.48)
b	85%	0.65	145 ⁰ C	6.70 (6.78)	6.18(6.20)
c	68%	0.59	152 ⁰ C	6.75 (6.78)	6.25(6.20)
d	83.6%	0.75	160 ⁰ C	6.82 (6.78)	6.15 (6.20)

C and H analysis was found satisfactory in all cases.

Antimicrobial activity¹³:

All the compounds have been screened for both; antimicrobial and antifungal activity by using disc diffusion assay. For this, sterile filter paper disc (6 mm) impregnated with fixed doses of compounds was placed on pre-inoculated surface. The disc bearing plates were incubated at 37°C for 24 hr. After incubation, zone diameters were measured. The compounds were taken at a concentration of 1 mg/mL using dimethyl sulphoxide as a solvent. Amikacin (100 μ g/mL) was used as standard for antibacterial

activity (100µg/mL). The compound was screened for antibacterial activity against *Escherichia coli*, *Staphylococcus aureus*, in nutrient agar medium. It has been observed that all the compounds showed good activity against bacteria.

Compound	<i>E. coli</i>	<i>S. aureus</i>
1(3a)	17	16
2(3b)	14	17
3(3c)	18	18
4(3d)	19	15
Amikacin	18	21

Zone of inhibition in mm. (15 or less) resistance, (16-20 mm) moderate and more than

Conclusion

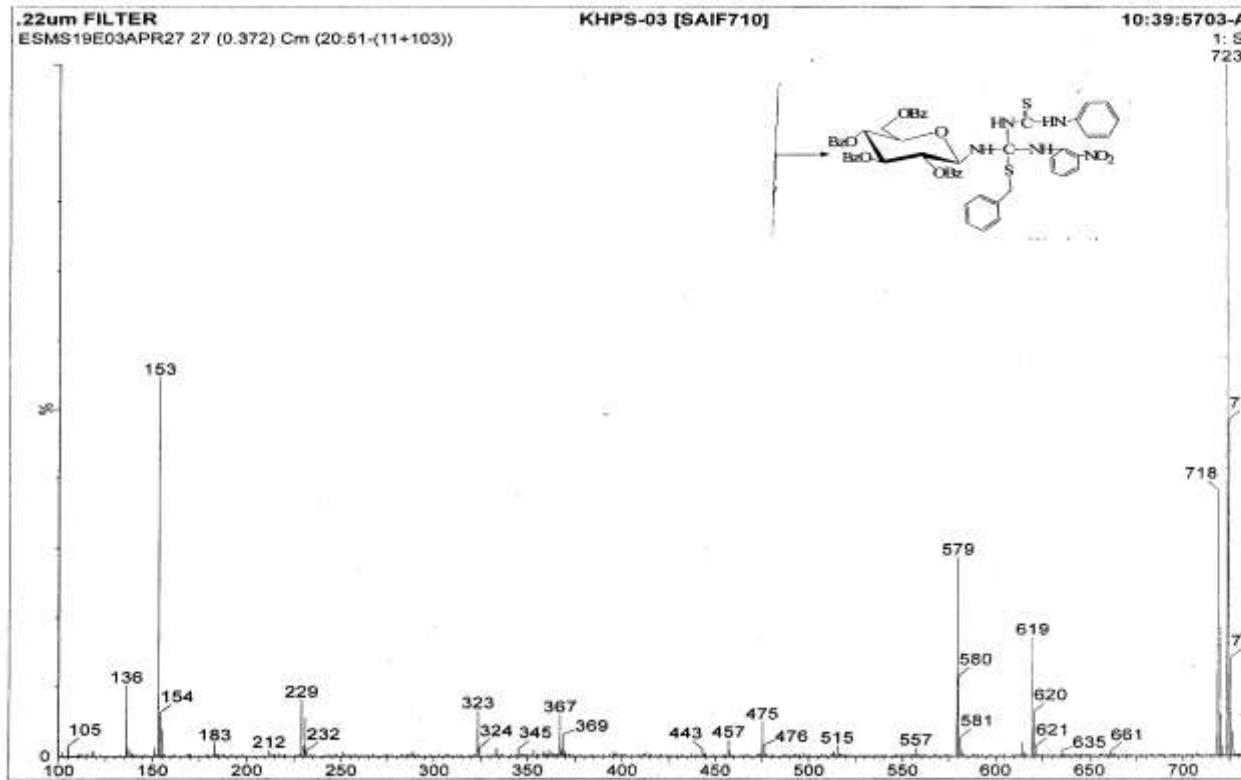
In this research work, the characterizations of newly synthesized products were established on the basis of IR, ¹H NMR, & Mass spectral studies. Various various 1-Tetra-*O*-Benzoyl- β -D Glucosyl-3-aryl (nitro substituted aniline)-2-phenyl thiocarbamides, 2-*S*-benzyl-isohiocarbamide were synthesized and yield of product ranged from 68-83%.

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References:

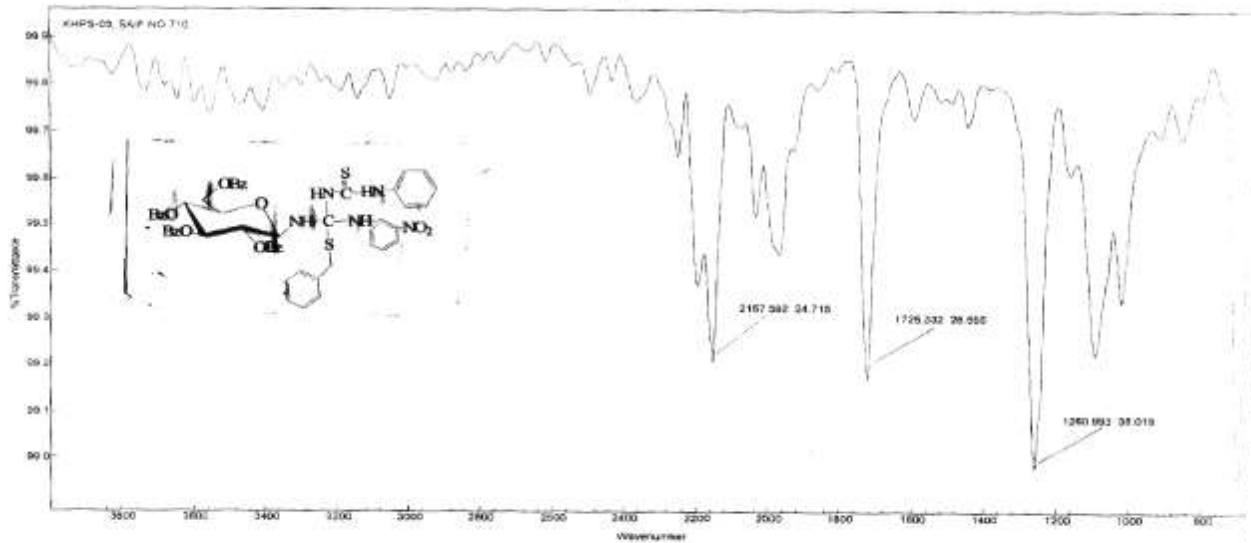
1. M. F. Abdel-Megged, M. A. Saleh, Y. L. Aly and I. M. Abdo *Nucleosides and Nucleotides*, **14** (9-10), 1985-1996 (1995).
2. Yoshiyuki Kobayashi, Masao Shiozaki and Osamu Ando, *J. Org. Chem.*, **60** (8) 2570-2580 (1995).
3. A. A. El-Barbary, M. A. Sakran, A. M. El-Madani and C. Nielsen, *Journal of Heterocyclic Chemistry*, **42** (5), 935-941 (2005).
4. V. Kikelj, K. Julienne, P. Janvier, J. C. Meslin and D. Deniaud, *Synthesis*, (21), 3453-3460 (2008).
5. M. M. Eid, S. A. Abdel Hady and H. A. W. Ali, *Archiv. Der. Pharmazie*, **323** (4), 243-245 (1990).
6. Saleh, M. A. Abdel-Megged, M. F. Abdo, M. A. And Shokr, A. B. M. 2002. Synthesis and Antiviral Evaluation of Some New Glycosylthioureas Containing A Quinazolinone Nucleus. *Nucleosides, Nucleotides, Nucleic Acids*, **21**(1), 93-106.
7. D. Jastalska and I. Dzierzynska, *Acta Pol. Pharm.*, **26**, 195-198 (1969).
8. Shusheng, Z. Tianrong, Z. Kun, C. Youfeng X. and Bo, Y. 2008. Simple and efficient synthesis of novel glycosyl thiourea derivatives as potential antitumor agents. *Eur. J. Med. Chem.*, **43**(12), 2778-2783.
9. Maya, I. Lopez, O. Fernandez-Bolanos, J. G. Robina I. and Fuentes, J. 2001. A Practical One Pot Synthesis of *O*-Unprotected Glycosyl Thioureas. *Tetrahedron Letters*, **42**, 5413-5416.
10. Silverstein R. M. and Webster F. X, 2011. *Spectrometric Identification of Organic Compounds* 6th ed., John Wiley and Sons, Inc, New York;
11. Williams D. H. and Fleming I, "Spectroscopic Methods in Organic Chemistry", 5th ed., Tata McGraw-Hill; (2004).
12. Dyer, J. R, 2010. *Applications of Absorption Spectroscopy of Organic Compounds*. PHI Learning Private Limited, New Delhi.
13. J. G. Collee and W. Marr Laboratory control of anti-microbial therapy. In: college J. G, Duguid J. P, Fraser A. G, Marmion B. P (eds) Mackie and Mc Cartney *Practical Medical Microbiology*. 14th edition. Edinburgh Churchill Livingstone. 1996, pp. 114-115.



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System - Agilent Cary 630 FTIR (ATR)

Sample Code: KHPS-03, SAIF NO.710



Analyzed By - Dr A.K. Mandal

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